

INVESTIGATION OF THE POROUS ENVELOPE EFFECT AT THE FAULT LAKE TAILINGS SITE

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INVESTIGATION OF THE POROUS ENVELOPE

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FINAL REPORT

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Executive Summary

In 1992, Noranda Technology Centre undertook a hydrogeological investigation of the Fault Lake tailings site. The tailings have been deposited in a kettle lake formed within glacial outwash sand and gravel. The site is unique in that, theoretically, a "porous envelope effect" may occur. If this is the case, flow through the tailings mass is low enough, relative to the surrounding, more permeable till, that impact to the ground water by tailings oxidation is insignificant at the regional scale. The specific objectives of the investigation were to analyze the chemical and physical hydrogeology of the site, to delineate areas affected by acid mine drainage generated from the tailings, and to verify the presence of the porous envelope effect.

The hydraulic conductivity (**K**) of the Fault Lake tailings measured 1.2×10^{-5} cm/s at a mid-level depth in the tailings and 3.6×10^{-6} cm/s in the deepest part of the tailings. Comparatively, the measurements of **K** for the glacial sediments averaged 1.6×10^{-3} cm/s. This is a two order of magnitude contrast in hydraulic conductivity. Flow modelling indicated that this is sufficient to route most regionally flowing groundwater around the tailings.

During the spring and fall, ponding occurs at the north dam, south dam and various berms. The water slowly infiltrates into the tailings and evaporates from the ponds. During the summer months, extensive ponding has not been observed. The water level in the tailings is perched higher and fluctuates greater than the regional water level. Regional groundwater flows to the northeast from the tailings dam at a velocity of about 2 m/yr. Groundwater flowing from the southerly dam goes south. Because the groundwater velocity is controlled by the hydraulic gradient, the velocities could have been higher during tailings disposal.

The tailings are characterized by two layers due to the disposal of different types of tailings: Layer 1 is pyrrhotite rich and Layer 2 is pyrrhotite poor. Layer 1 is centrally located on the tailings and in close proximity of the northerly spigot position. In the centre of Layer 1 pyrrhotite was identified to a maximum depth of 9 m, but was at highest composition in the upper 3 m where it is near 50%. Layer 2 is below Layer 1 in the centre of the tailings. In the southerly portion of the tailings Layer 1 pinches out.

Mineralogical analysis and acid-base accounting of the tailings indicated that carbonate mineral reserves are available for short-term neutralization of acid during the first stage of oxidation when rates are high, and silicate mineral reserves are abundant for long-term buffering. The neutralization potential of the tailings plays an important role for the attenuation of acidity and metals from sulphide oxidation, which were detected but have been attenuating in the tailings deposit.

Sulphide oxidation has been at its highest rate since deposition discontinued in 1978, yet little impact of sulphide oxidation was observed in the groundwater of the surrounding till. Sulphide oxidation products leaching from the tailings appear to be alleviated by the porous envelope effect. Several favourable factors contribute to create the porous envelope effect and to limit the observed metal concentrations downgradient of the tailings:

- (1) the hydraulic conductivity contrast between the tailings and the surrounding sediments;
- (2) the limited infiltration through the surface of the tailings;
- (3) the dilution of metals flushed from the tailings by water flowing around and below the tailings; and
- (4) the chemical attenuation of metals, which likely plays a large role both inside the tailings mass and in the surrounding sediments.

The porous envelope effect could probably be present at other locations near mine sites. Tailings deposition could possibly be done at these sites with little effect on groundwater quality, pending that thorough site evaluations are performed and that appropriate control is done at the time of deposition.

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1. INTRODUCTION

In 1992, Noranda Technology Centre (NTC) undertook a hydrogeological investigation of the Fault Lake tailings site. The site is unique in that, theoretically, a "porous envelope effect" may occur. If this is the case, flow through the tailings mass is slower than the surrounding, more permeable till, so that impact to the regional groundwater by tailings oxidation is insignificant.

The specific objectives of the investigation were to analyze the physical and chemical hydrogeology of the site, to delineate areas affected by acid mine drainage generated from the tailings, and to verify the presence of the porous envelope effect. A report documenting part 1 of the investigation was issued in September 1993 (NTC 1993). The current report documents part 2 of the investigation. Portions of the part 1 report are summarized in this report.

1.1 Background

The Fault Lake tailings site is located northwest of the Falconbridge Sudbury operations, approximately 3 km north of Falconbridge and 0.5 km east of the Sudbury Airport (**Fig. 1**). The tailings were deposited between 1965 and 1978 and were produced from the milling of nickel ore in the Sudbury area. **Figure 2** is an aerial photograph of the site in 1946, showing the site before tailings disposal activities. The region was characterized by several small depressions (kettles) in which many contained lakes (kettle lakes). Approximately 6.45 million tonnes of tailings containing as much as 50% pyrrhotite were deposited in a kettle lake depression. Tailings were discharged at two spigot locations on the west and were contained by dams to the north and south of the site (**Fig. 3**). The deposit has an approximate volume of $3.36 \times 10^6 \text{ m}^3$ and a surface area of 22.2 ha (55 acre). It sits in a 55 ha (136 acre) closed watershed.

In 1971, while deposition was active, the analysis of groundwater in one well located 2 km downgradient (northeast) of the Fault Lake tailings site indicated above-background sulphate levels of 382 mg/L, suggesting influence from tailings oxidation (International Water Supply, 1971). Groundwater and surface water monitoring and analysis done at later dates showed improvements in water quality.

1.2 Surficial Geology

Kettles, fluvial terraces, discontinuous crevasse fillings, and eskers within the Fault Lake tailings area are evidence of a glacial meltwater channel, partly choked with stranded ice blocks. The small round kettle lake depressions were formed after melting of stranded ice blocks which were caught among

the mass of glacial sediments. The sediments are assembled in longitudinal formations which follow a southeasterly meltwater flow direction, leading from Bowlands Bay, part of Lake Wanipitie. **Figure 4** shows the main overburden materials and their orientation in the area of the site. Overburden beneath the tailings and surrounding the site mainly consists of coarse to fine glacial outwash sands and gravels with some large boulders and silt lenses. Overburden thickness varies within the studied area, from 36 m to more than 60 m.

1.3 Summary of part 1 of the investigation

The draft report (NTC 1993, Appendix E) documented the field activities and computer flow modelling conducted in 1992 and 1993. The activities included the following:

- 1 Installation of 14 groundwater monitoring stations, consisting of 1 to 3 piezometers each,
- 2 Grain-size distribution analysis on recovered core samples,
- 3. Measurement of overburden hydraulic conductivities in the piezometers using a falling head test.
- 4. Measurement of water levels in the piezometers in December 1992 and March 1993,
- 5. Sampling and water quality analyses of groundwater taken from the piezometers in December 1992 and March 1993,
- 6. Two-dimensional saturated-unsaturated steady-state flow modelling of the site using SEEP/W computer software.

The report concluded that conditions for porous envelope containment may be occurring at the Falconbridge Fault Lake tailings site. Water quality sampling did not show any evidence of above background metal concentrations, which suggests minimal leaching of metals from the tailings. Flow modelling supported the porous envelope hypothesis, which corroborates the inference that impact to the regional groundwater by tailings oxidation is insignificant. Factors which contribute to low metal concentrations downgradient of the Fault Lake tailings are listed, as follows:

1. High hydraulic conductivity contrast between the tailings and the surrounding overburden sediments,

- 2. Low position of the water table relative to the tailings bottom,
- 3. Limited infiltration through the surface of the tailings,
- 4. Dilution of metals flushed from the tailings by groundwater flowing through the overburden, around and below the tailings,
- 5. Chemical attenuation of metals in the tailings and overburden.

1.4 Objectives of part 2 of the investigation

The specific objectives of the investigation were to complete the analysis of the physical and undertake the chemical hydrogeology of the site, to delineate areas affected by acid mine drainage generated from the tailings, and to verify the presence of the porous envelope effect. In part 1 of the investigation, groundwater monitoring stations were located outside of the original lake-shore boundary. A principal objective of part 2 of the investigation was to core into the deepest part of the tailings deposit, recover tailings samples for geochemical evaluation, install piezometers in the tailings and sample tailings porewater. With piezometers located in the tailings and below the watertable it was also possible to conduct field measurements of hydraulic conductivity of the tailings. The results were compared with estimates of hydraulic conductivity calculated in part 1 of the investigation using the grain-size distribution D_{50} and the modified Kozeny-Carman equation.

In part 1 of the investigation, using water level measurements in the piezometers and lake level elevations from the topography map, the regional flow direction was shown to be to the northeast, through the esker sediments and at an average flow velocity of 30 cm/yr. These findings were used to define the boundary conditions of the flow modelling. To support these findings, in part 2 of the investigation, surface water monitoring stations were established to monitor the water levels of the kettle lakes to the northeast and southwest.

Groundwater sampling from the piezometers and water quality analysis was continued in part 2 of the investigation, as well as sampling and analyses of surface waters at the monitoring stations.

2. METHODOLOGY

2.1 Field work

2.1.1 Coring and installation of groundwater monitoring stations FS15

Groundwater monitoring station FS15 was approximately located in the deepest part of the tailings, between FS5 and FS12 (**Fig. 5**). Prior to the drilling campaign, piezocone testing was conducted at the coring site by the University of British Colombia, Department of Civil Engineering, in-situ testing group (Davies 1994). Results not only showed the presence of layered tailings and a 9 m unsaturated zone but also indicated that the maximum depth to refusal was 41.1 m. This evidence confirmed the location of FS15. The plotted piezocone data can be found in Appendix D.

Drilling of the groundwater monitoring station FS15 was conducted on Dec 11-12, 1993 using a 15 cm ID hollow stem auger mounted on a Acker 82 drill rig. Two holes were drilled and one piezometer was installed in each. Piezometer FS15-A was installed at 35 m and FS15-B, at 24 m. The piezometers were constructed of 2.4 cm (15/16 inch) ID, schedule 80 PVC pipe with a 0.3 m (1 ft) PVC screened tip. They were installed by placing the pipe inside the hollow stem auger at the required depth. The auger was raised approximately 1.5 m and clean silica sand was packed around the PVC screen. A bentonite seal was placed above the sand to insure hydraulic isolation of the well. The auger was then pulled up which allowed saturated tailings to close in around the piezometer pipe. A second bentonite seal was placed above the tailings and the open hole was backfilled with sand. Bentonite seals and steel casings equipped with locking covers were also installed at the surface to protect the piezometers and prevent infiltration of water from surface. The elevation of the uncapped piezometer and ground surface were surveyed by Falconbridge Ltd. Exploration. Water level measurements were conducted on May 27, July 16 and August 27 of 1994. The borehole and piezometer installation logs and water level data are located in Appendix A.

While coring FS15-A, continuous vertical sampling of tailings was conducted in 5-ft intervals using a 5-ft (split spoon) sampling barrel. Upon retrieval, the samples were split into approximately 45 cm lengths and stored with minimal atmospheric contact to preserve in-situ conditions. Laboratory analyses performed on the recovered samples are described in Section 2.2.2.

2.1.2 Hydraulic conductivity measurement at FS15

Measurements of *in-situ* hydraulic conductivity were conducted at FS15-A and FS15-B using the "rising head test". The test is performed below the water table. The piezometer is pumped dry, simulating an instantaneous water level decline in the piezometer, and water level recovery is recorded with time. Water level recovery was monitored manually using a watch and water level indicator.

Interpretation of the water level versus time data was conducted using the Hvorslev (1951) method for point piezometers. As described in Freeze and Cherry (1979), the hydraulic conductivity (K) is determined using the following equation:

$$K' \frac{r^2 \ln \frac{L}{R}}{2LT_o}$$
(1)

where, T_o is the time lag or time that would be required for the complete equalization of the head differences if the original rate of inflow were maintained, L is the length and R is the radius of the piezometer intake or screen, and r is the inside radius of the piezometer pipe.

2.1.3 Trenching

The purpose of the trenching activity was to aerially characterize the nearsurface tailings. Three trenches were dug with a backhoe to a maximum depth of 2.5 m. Locations of the trenches are shown in **Figure 6**. The trench walls were logged in detail, noting visible grain-size and color changes as indicators of composition and sulphide oxidation. Tailings samples were recovered for moisture content determination and geochemical analysis as described in Section 2.2.3.

2.1.4 Installation of surface-water monitoring stations

At each station a 1-inch diameter steel bar was driven in the lake to which a "2x4" wood was attached. The 1 m staff plate was attached to the "2x4" with screws so that approximately half of the staff plate was submerged. Zero level on each staff plate was surveyed by Falconbridge Ltd. Exploration. The staff plates were installed in the kettle lakes to the northeast and southwest of the site in order to determine the regional water level elevations and longitudinal gradients. The location of the monitoring stations at Lakes 1-5 are shown in **Figure 5**. Monitoring stations were also installed in two lakes further to the southwest, shown in **Figure 1**. These two lakes were not referred to in part 1 of

the investigation (Appendix E) and were therefore identified as Lakes A and B. Water level readings were conducted on May 27, July 16 and August 27 of 1994. Survey coordinates and water level data are documented in Appendix B.

2.1.5 Water quality sampling

Water from the surface-water and groundwater monitoring stations was sampled in December 1993, May 1994 and August 1994. Before sampling, the depth to water level was measured and at least three well volumes were purged to remove standing water. After the water had been sufficiently recovered, depth to water level was re-measured and water samples collected.

The samples were collected using a peristaltic pump, a nitrogen-driven positive displacement pump, or the Waterra system. The groundwater samples were filtered using a 0.45 μ m (ACRO 50A) disposable in-line filter. Field measurements of pH, temperature, oxidation reduction-potential (ORP), and electrical specific conductance (EC) were recorded. Half of each sample was acidified in the field using reagent grade (2% v/v) hydrochloric acid (HCI) for metal preservation prior to analysis. All electrodes were calibrated before use and between samples. All sampling equipment was rinsed with distilled water before each sample was collected.

Water samples were transported to a field laboratory within six hours from collection. In the laboratory, measurements of pH were repeated on the non-acidified portion of the samples along with titration for acidity and alkalinity.

2.1.6 Measurement of near-surface pore-gas oxygen concentrations

The dominant method of oxygen transport in tailings has been shown to result from diffusion through partially gas-filled pores and is described by Fick's first law:

$$J' & D_e \frac{MC}{MZ}$$
(2)

where

 $\begin{array}{rcl} J & = & \text{diffusive flux of oxygen (moles } m^{-2} \text{ s}^{-1}) \\ D_e & = & \text{effective diffusion coefficient } (m^2 \text{ s}^{-1}) \\ C & = & \text{concentration (moles } m^{-3}) \\ Z & = & \text{depth in the tailings (m)} \end{array}$

The concentration gradient required for diffusion in tailings results from differences between gas concentrations in the pore spaces and the atmosphere. In pore spaces of reactive tailings, gaseous oxygen concentrations are lower than atmospheric values largely due to consumption by geochemical reactions related to the oxidation of sulphide minerals. The resulting gradient drives gaseous oxygen into the tailings at a rate which is governed by the diffusion coefficient. An oxidized front generally develops in tailings where un-oxidized tailings are actively oxidized and oxygen is readily consumed. As reactive tailings age, the oxidized front generally migrates deeper into the tailings which decrease the gradient. Many factors control the oxidation process and the character of the oxidation front but in general as oxidation progresses, the concentration gradient decreases.

On the surface of the Fault Lake tailings, a hard pan has developed which may retard the flux of oxygen to the sulphide tailings and limit oxidation of the tailings. An estimate of the flux may be calculated from Fick's first law by knowing the diffusion coefficient of the surface hard pan and the concentration gradient across the surface hard pan. As a reconnaissance to oxygen flux determination, pore-gas sampling probes were installed at several depths below the surface hard pan. Oxygen concentrations were measured and concentration gradients calculated. **Figure 7** shows the probe installation and sampling procedure. Each probe was installed in a 2-inch diameter hole, which was drilled to depth with a hand auger. Coarse sand was place around the probe tip and the hole was backfilled with bentonite. The probes were measured for gaseous oxygen concentration on December 13, 1993 and May 27, 1994. The gas was sampled by extraction of 5 cm³ volume with a hypodermic needle, and measured with a Teledyne portable oxygen analyzer.

2.2 Laboratory work

2.2.1 Analysis of water

Chemical analysis was conducted on groundwater samples taken from piezometers. Samples were split in the field and one portion promptly preserved by the addition of HCl to a concentration of 2% by volume. In the NTC analytical laboratory, the acidified portion of each water sample taken from the piezometers was analyzed for dissolved metal and major ions and the non-acidified portion was analyzed for chloride. Potassium (K) was analyzed using flame atomic emission, ferric iron (Fe³⁺) by colorimetry/volumetry, chloride (Cl) by turbidimetry, and all other elements by inductively coupled plasma spectrophotometry (ICP). All certificates of analysis appear in Appendix C.

The kettle lakes were sampled at the staff plate gauges. Grab samples were taken, filtered and analyzed using the same procedure as for the groundwater samples.

Quality assurance and quality control (QA/QC) testing for all sample batches was performed using replicate and standard samples. The samples were collected to evaluate reproducibility and accuracy of the analytical procedure and to assess the cleanliness of the equipment during sampling. The results of the quality assurance testing are included in Appendix C.

2.2.2 Solids and porewater analysis of FS15 core samples and trenching samples

Selected tailings samples recovered from the coring of borehole FS15-A and trenching activities were analyzed for mineralogy and porewater quality. Porewater was obtained by squeezing tailings samples at 0.8 MPa in a stainless steel loading cell using a pneumatic squeeze apparatus. The extracted porewater was promptly preserved by the addition of HCl to a concentration of 2% by volume. The samples were analyzed using ICP methods to determine major metal and ion concentrations. A second and non-acidified aliquot of porewater was used for pH, redox potential, and electrical specific conductivity measurements, and for Cl determination.

After squeezing, the remaining solids were oven-dried and analyzed at NTC for selected elemental composition using acid digestion procedures and ICP methods. An non-squeezed tailings sample of approximately 100 g was also oven dried and sent to Lakefield Research Laboratories for x-ray diffraction (XRD) testing using a Co target and Fe filter. Results of the XRD analysis provide qualitative mineralogical composition of the samples. Six core samples were selected for XRD-pattern interpretation and free silica quantification by Lakefield Research Laboratories.

2.2.3 Acid-base accounting of FS15 core samples

Acid-base accounting (ABA) was conducted on selected tailings samples recovered from cores of the FS15-A borehole. ABA is a static test which examines the acid-generating and acid-neutralizing capacity of a sample. ABA cannot reveal whether a sample will become acidic, it only measures the theoretical acid-base balance of a sample. Kinetic testing (e.g., humidity cells) is required to determine if/when and to what extent the acidity may be generated.

The maximum potential for acid production (AP) was calculated from the total S analysis using stoichiometric equation 3 of pyrrhotite oxidation. A factor of 31.25 was applied to the percent S, which assumes 1 mole FeS is neutralized

by 1 mole CaCO₃. Units are given in kg of CaCO₃ equivalent per tonne of tailings.

FeS % CaCO₃ %
$$\frac{9}{4}O_2$$
 % $\frac{1}{2}H_2O$ ' FeOOH % SO₄^{2&} % CO₂ % Ca^{2%} (3)

The acid-neutralizing potential (NP) was evaluated using the B.C. Research Initial Test method (Duncan and Bruynesteyn 1979). A 10 g sample was suspended in 100 mL of distilled water and stirred for approximately 15 minutes. The natural pH was recorded, and the sample was titrated to pH 3.5 with 1.0 N sulphuric acid using an automatic titrator. The test was continued until less than 0.1 ml of acid was added over a 4 hour period. The total volume of acid added was recorded and converted to kg of H₂SO₄ per tonne of sample. The choice of end point of pH 3.5 was based on an assumption that this represents the limit above which iron and sulphide oxidizing bacteria such as Thiobacillus ferrooxidans are not active. Therefore, if the theoretical acid production is less than that determined by the B.C. Research Test (i.e., the amount of acid required to titrate the solid to an end pH of 3.5), then biochemical oxidation can not be maintained. The test gave results in kg H_2SO_4 per tonne of material. Units were converted to kg of CaCO₃ per tonne tailings using stoichiometric equation 4, where 1 mole H_2SO_4 is neutralized by 1 moles CaCO₃.

$$CaCO_{3} \% H_{2}SO_{4} ' Ca^{2\%} \% CO_{2} \% SO_{4}^{2\&} \% H_{2}O$$
 (4)

The AP and NP are evaluated in the ABA method by two approaches: (1) the net neutralization potential (NNP) which is calculated by substraction of AP from NP, and (2) the ratio of NP to AP. In general, if the NNP is negative then the sample is declared "potentially acid generating", and if NP/AP > 2.5 then the sample is declared "net acid neutralizing". Paste pH was also conducted on the tailings samples which can sometimes indicate whether net acid generation has already developed.

2.2.4 Application of MINTEQA2 to trench and borehole sample analyses

Geochemical data for tailings porewater was interpreted with an equilibrium speciation computer model called MINTEQA2 (USEPA, Aug 1990). The program has a extensive thermodynamic database and utilizes thermodynamic principals to solve multiple-component chemical equilibrium reactions for gaseous, aqueous and solid phase interactions, including adsorption.

For each porewater sample analysis, temperature, pH, redox potential and aqueous components were inputed into the model. The model determines the mass distribution of possible aqueous ion species and the saturation indices (SI) of possible mineral solid phases. Saturation indices were calculated using the following equation:

$$SI + \log\left(\frac{IAP}{K}\right)$$
 (5)

where, SI = saturation index (unitless), IAP = ion activity product (mol/L), K = solubility product (mol/L).

SI values equal to zero indicate equilibrium with respect to a mineral phase. SI values less than zero indicate undersatuation and SI values greater than zero indicate supersaturation. Such information provide a basis for determining the probability of solids that may form in the tailings or that may be present and reacting with resident porewater.

3. RESULTS

3.1 Physical hydrogeology

3.1.1 Groundwater

Measurements of water level elevation in the piezometers were conducted in November of 1992, March and December of 1993, and May, July and Aug of 1994. All data were tabulated and plotted against time, and appear in Appendix A. Water level elevations in the kettle lakes were measured in May, July and August of 1994. These data appear in Appendix B.

A generalized watertable contour map (**Fig. 8**) was developed for the Fault Lake tailings area by using the water level data and the topographic contours taken from the survey map (Falconbridge Exploration survey, May 1984). Because shallow groundwater flow in hummocky unconsolidated terrain tends to be controlled by topography, the watertable contours were generally drawn to reflect the topographic contours. The galciofluvial sand and gravel deposits form the upland area, on which the Sudbury Airport and Fault Lake tailings were placed, and the glaciolacustrine deposits to the northwest and southeast form the lowlands (see **Fig. 4**). The shallow groundwater flow, indicated by the arrows on Figure 8, travels from the uplands to the lowlands. The flow direction is perpendicular to the watertable (head) contours, and where the contours are closely grouped, the flow gradient is large. South of the tailings the flow is to the west with a gradient of 0.02. North of the tailings, the flow is towards the northeast with a much smaller gradient. The flow gradient is commonly used to calculate the velocity of flow which is explained later in this section.

A cross-sectional representation of the data illustrates the longitudinal position of the watertable. The water level elevation in the shallowest piezometer was used as a close approximation of the watertable elevation which is acceptable where vertical head gradients are small, as they were in the regional overburden aquifer (Appendix A). Two cross sections were plotted (**Fig. 9**). Section A-A' extends southwest-northeast and illustrates the regional watertable position. Section B-B' extends southeast-northwest and illustrates the watertable position across the tailings.

Section A-A' is represented in **Figure 10**. The lakes located south of the tailings had a watertable elevation in the vicinity of 308 m and showed no south-southwesterly gradient. This indicates that the lakes are alined along a watertable contour. Golder Associates Ltd. in a hydrogeological investigation of the Falconbridge smelter area (1993), show the lakes to be located at the groundwater divide. The watertable below the northern portion of the tailings

and extending to the northeast was at 299 m and showed a gradient of 0.0002.

Section B-B' is represented in **Figure 11**. The figure shows the deepest part of the tailings below the regional watertable level of 299 m, and the watertable in the tailings perched above this level. The perched watertable also shows large changes in elevation. These large fluctuations typically occur in tailings (Woyshner and St-Arnaud 1994), a process which has been described by Abdul and Gillham (1984). It is related to the fact that the watertable can rapidly rise through the capillary fringe after a minor recharge event of very little volume. Vertical head gradients are also larger in the tailings (Appendix A), which are necessary to push water through finer grained material. At FS15 (located in the centre of the tailings), the gradient is downward when the water level is high and upward when the water level is low. Measured gradients were 0.55, -0.05 and -0.35, respectively in May, July and Aug of 1994.

Table 1 shows the results of the hydraulic conductivity tests for each piezometer. Measured hydraulic conductivities in the natural overburden units were highly variable, ranging between 8×10^{-1} cm/s (at FS10) and 2.5×10^{-5} cm/s (at FS4). The large variations in hydraulic conductivity are explained by the variability in soil types typical of ice-contact deposits which include silts, sands, gravels, and boulders. The higher values of hydraulic conductivity (such as at FS10 and FS14) would occur where fast meltwater flows would have formed accumulations of well-sorted sands and gravels. The lower hydraulic conductivities occur where glacial abrasion and slow meltwater flows would have left silts. The hydraulic conductivity values also suggest that silts may be present within void spaces between boulders (at FS4, for example).

The geometric average of all hydraulic conductivity measurements is 1.6 x 10⁻³ cm/s. This value would be representative of a clean to silty medium sand, and is considered to be representative of the overall effective hydraulic conductivity of the ice-contact deposits in which the tailings lie.

The hydraulic conductivities were measured in the tailings at FS15-A and at FS15-B. In FS15-A, at a 35 m depth, the hydraulic conductivity was 3.6×10^{-6} cm/s, and in FS15-B, at 24 m, it was 1.2×10^{-5} cm/s. The results are similar to those calculated in part 1 of the investigation (NTC 1993) using the grain-size distribution D₅₀ and the modified Kozeny-Carman equation (Bear 1972). The resulting estimates averaged 1.2×10^{-5} cm/s which is identical to the measurement at FS15-B. The lower conductivity at FS15-A supports the inference that the tailings are more consolidated at a deeper depth.

The average linear groundwater flow velocity (v) in the overburden and tailings can be estimated by the Darcy equation.

$$\mathbf{v} = \mathbf{K}\mathbf{i}/\mathbf{n} \tag{6}$$

Using the average hydraulic conductivity (K), hydraulic gradient (i), and an estimated porosity (n), the calculated velocity for various portions of the site were estimated.

Calculated groundwater velocity.

Flow Location	Hydrauli c Gradien t	Hydraulic Conductivit y (cm/s)	Porosit y	Velocit y (cm/da y)
Maximum vertical flow through the saturated tailings	0.55	1 x 10 ⁻⁵	0.45	1.1
Westerly lateral flow through the overburden south of the tailings	0.02	1 x 10 ⁻³	0.30	5.8
Northeasterly lateral flow through the overburden north of the tailings	0.0002	1 x 10 ⁻²	0.30	0.6

3.1.2 Unsaturated zone

Table 2 shows the measured water content of the tailings samples recovered from borehole FS15-A in December 1993. The watertable level in the tailings was approximately 15 m below the surface, and above the watertable, alternating wet and dry layers were present. The alternating wet and dry layers are also present in the trench samples taken in May 1994 (**Table 3**), particularly in Trenches 2 and 3. In Trench 1 the water content ranged between 28 and 39% which is similar to the water content in the wetter layers of Trenches 2 and 3. Results show that the tailings are unsaturated, and therefore not inhibiting oxidation.

Dry densities were measure on samples taken from the trenches. **Table 4** shows the results and the calculated volumetric water contents. The dry density of the samples taken from Trench 1 were lower than those from Trenches 2 and 3. Volumetric water content was also higher in samples taken from Trench 1 than those from Trenches 2 and 3.

3.2 Chemical hydrogeology

This section presents the results of the chemical analyses performed on the water samples collected from the piezometers and kettle lakes. Sampling was conducted in December 1992, March and December 1993, and May and August 1994. The results of the December 1992 and March 1993 analyses were presented in the report documenting part 1 of the investigation (NTC 1993) but are also summarized in this section. Results from the December

1993, May and August 1994 analyses are shown in **Tables 5 through 11** and discussed in this section. Concentrations of nickel, iron and sulphate characterized the general water quality found at the site and are discussed in most detail. Other metal and ion concentrations and physico-chemical parameters were determined and are listed in the tables.

3.2.1 Composition of regional waters

The background groundwater monitoring station FS2 showed a near neutral pH, owing to the alkalinity in the water. The alkalinity in the groundwater taken from the deeper piezometer was around 80 mg/L as CaCO₃. The shallower piezometer showed 20 mg/L of alkalinity as CaCO₃. This suggests that alkalinity is being consumed by water percolating through the vadose zone. Acidity was less than 10 mg/L as CaCO₃. Nickel concentrations were generally less than 0.025 mg/L. Analysis for iron showed a maximum concentration of 0.5 mg/L as Fe³⁺ but were commonly less than the 0.025 mg/L detection limit. Sulphate concentrations ranged between 30 and 43 mg/L.

Date	Piezomete	Ni	Fe⊤	Fe ³⁺	S	SO₄²⁻
	r	mg/L	mg/L	mg/L	mg/L	mg/L
Dec 92	FS2-A FS2-B	0.012 0.010	0.027 0.018		12.2 12.2	34 32
Mar 93	FS2-A FS2-B	< 0.005 < 0.005	< 0.005 0.033		11.0 11.1	
Dec 93	FS2-A FS2-B	< 0.025 < 0.025	0.43 < 0.025	0.50 < 0.05	14.6 11.9	
May 94	FS2-A	< 0.025	< 0.025	< 0.05	12.5	41
	FS2-B	< 0.025	< 0.025	< 0.05	8.7	30
Aug 94	FS2-A	< 0.025	< 0.025	< 0.15	15.4	43
	FS2-B	< 0.025	< 0.025	< 0.15	10.9	30

Background groundwater monitoring station FS2 analytical data.

Grab samples taken at the surface water monitoring stations near the tailings site and downgradient further to the northeast (Lakes 1-5) generally resembled the groundwater taken from the background monitoring station (FS2). The pH was above 6 and the alkalinity ranged between 25 and 70

mg/L as CaCO₃. Nickel and iron concentrations were below the 0.025 mg/L detection limit and sulphur ranged between 2 and 8 mg/L (6-24 mg/L SO_4^{2-}).

Further to the southwest, at stations Lake A and Lake B, grab samples showed a different composition. The pH was lower, ranging between 4 and 5, and the alkalinity was depleted. Nickel concentrations were 0.1-0.3 mg/L at Lake A and 0.03-0.08 mg/L at Lake B. Iron was detectable at 0.06 mg/L in a sample from Lake A and 0.2 mg/L in a sample from Lake B. Sulphur concentrations resembled those measured in the other lakes. These values show how regional surface waters can vary owing to such factors as atmospheric deposition or production of organic acids.

3.2.2 pH, alkalinity and acidity

Groundwater sampled from the overburden aquifer generally had pH values above 6, owing to the alkalinity in the water. Alkalinity measurements averaged 75 mg/L as CaCO₃ and had a sample standard deviation of 42 mg/L as CaCO₃. Highest alkalinities in the overburden aquifer, ranging from 105 to 170 mg/L as CaCO₃, were observed beneath the tailings at FS3, FS5 and FS13. In the saturated zone of the tailings, the alkalinity was higher that any observations in the overburden aquifer; values ranged from 246 to 270 mg/L as CaCO₃ for samples taken from station FS15. This may suggest that portions of the tailings are currently a source for alkalinity, increasing the alkalinity in the overburden aquifer above background levels.

In the piezometers northeast of the site, pH values fell below 5 for samples collected in May and August 1994. Alkalinities were also depleted. Because these data were lower than those of background waters in close proximity to the tailings, this suggests that acidity may be migrating from the tailings at the northerly dam. The values, however, were not lower than those of Lakes A and B, further to the southwest. Measurements of acidity at the stations northeast of the dam were low, generally less than 10 mg/L as CaCO₃ but showed maximum levels of 26, 16 and 38 mg/L as CaCO₃, respectively at FS8, FS9 and FS10. Downgradient of the southerly dam, at FS1, the pH, acidity and alkalinity resembled background levels.

	Dec 93	М	ay 94	A	ug 94
Station	рН	рН	Alkalinity mg/L CaCO₃	рН	Alkalinity mg/L CaCO₃
FS8-A	6.2	4.1	8		
FS9-A FS9-B FS9-C	5.5 6.7 6.4	6.3 6.0 4.6	62 66 6	5.9 6.1 	70 100
FS10-A FS10-B	6.6 6.0	6.2 3.1	22 6	5.3 4.3	30 10

Field pH and alkalinity in groundwater samples taken from stations northeast of the site.

3.2.3 Iron concentrations

Above-background levels of iron were consistently observed in samples taken from the stations to the northeast of the site (FS8, FS9 and FS10). The highest observed concentration was 23.5 mg/L at FS8 in December 1993 but in May 1994 it had reduced to 8 mg/L. This trend was also observed in samples taken from FS9. The source of the iron may likely be the tailings at the dam where water commonly ponds and flushes that portion of the tailings. However, because sulphate levels are not elevated (section 3.2.5), it may also be caused by the dissolution of siderite (FeCO₃) in the aquifer, a precipitate remnant of tailings discharge.

Station	Dec 92	Mar 93	Dec 93	May 94	Aug 94
FS8-A	1.31	0.019	23.5	7.95	
FS9-A	0.023	0.091	0.204	< 0.025	< 0.025
FS9-B	0.042	0.139	0.119	< 0.025	< 0.025
FS9-C	0.055	0.746	1.87	0.794	
FS10-A	0.034	0.002	0.807	0.783	0.343
FS10-B	0.710	0.003	0.856	0.686	0.168

Observed above-background iron concentrations (mg/L) in overburden aquifer.

3.2.4 Nickel concentrations

Above-background concentrations of nickel were measured at the stations to northeast of the site (FS8, FS9, and FS10) and below the southerly portion of the tailings (FS3 and FS13). Samples taken from the shallow piezometer at FS10 consistently showed values above 0.5 mg/L, ranging between 0.5 and 1.2 mg/L. At FS9, 0.4 mg/L was measured twice. The highest concentration of nickel in the saturated zone of the tailings (at FS15) was 1.2 mg/L. Therefore, the source of nickel in the overburden aquifer may likely be the ponded area at the northerly dam, and possibly other ponded areas in the southerly portion of the tailings. Alternatively, because nickel coprecipitates with iron, it may have been released with the dissolution of siderite.

	-				
Station	Dec 92	Mar 93	Dec 93	May 94	Aug 94
FS3-A FS3-B	0.014	0.138 0.039	 < 0.025	 < 0.025	< 0.025 1.428
FS3-C					
FS8-A	0.007	< 0.005	< 0.025	0.544	

Observed above-background nickel concentrations (mg/L) in overburden aquifer.

FS9-A	0.010	< 0.005	0.079	0.066	0.077
FS9-B	< 0.005	< 0.005	< 0.025	< 0.025	0.084
FS9-C		0.046	0.418	0.407	
FS10-A	0.006	0.019	0.055	< 0.025	0.041
FS10-B	0.009	0.532	0.853	1.200	0.739
FS13-A				0.059	0.116

3.2.5 Sulphur concentrations

Above-background sulphur concentrations (as SO_4^{2-}) were encountered in samples taken from stations below the tailings (FS3, FS4, FS5 and FS13). Highest concentrations were measured below the southerly portion of the tailings, in FS3 and FS13. Nickel concentrations were also above background levels at FS3 and FS13. The presence of the tracer-labelled drilling water was encountered in all samples taken from station FS3. Samples should therefore be obtained from this station at a later date to confirm the elevated concentrations, after all the drill water evacuates.

Above background levels of sulphur were observed in samples taken from FS1, the monitoring station below the southerly dam (but nickel was not detected). This suggest that sulphate is likely migrating from the tailings and to the south, towards the New tailings area. The New tailings area is an active site of high pyrrhotite composition.

Below the northerly portions of the tailings, at FS4 and FS5, sulphur concentrations were above background levels but not greatly elevated (32-66 mg/L). At the stations located downgradient, northeast of the site where nickel and iron were detected, pH was depressed and alkalinity was depleted (at FS8, FS9 and FS10), sulphate concentrations were at background levels. This suggests little impact from tailings oxidation.

Station	Dec 92	Mar 93	Dec 93	Ма	y 94	Au	g 94
	S	S	S	S	SO4 ²⁻	S	SO4 ²⁻
FS1-A FS1-B FS1-C	73.7 85.7 11.8	72.5 113. 11.2	75.2 135. 22 4	71.3 122. 9.5	210. 318. 32.4	78.9 118.	216. 319.
FS3-A FS3-B FS3-C	4.63 	49.5 82.2	 187.	 188. 	 412.	192. 111.	534. 303.
FS4-A FS4-B FS4-C	 	35.4 34.8 8.49	 	58.3 16.6 10.3	166. 50.6 49.8	62.8 18.3 11.5	190. 57.2 35.6
FS5-A FS5-B FS5-C	 	33.8 33.6 32.1	 	33.8 35.7 64.2	97.7 104. 187.	37.0 39.8 65.9	112. 120. 191.

Observed above-background sulphur concentrations (mg/L) in overburden aquifer.

FS13-A 276 .

3.2.6 Site manganese concentrations

Manganese was below the 0.005 mg/L detection limit in samples taken from the background monitoring station (FS2) and the surface water monitoring stations (Lakes 1-5). Above background concentrations of manganese were observed in samples taken from all of the other stations except FS1, the station below the southerly dam. Highest concentrations were 94.5 mg/L at FS3, 36.2 mg/L at FS4 and 69.1 mg/L at FS5.

3.3 Tailings mineralogy

3.3.1 Borehole FS15-A samples

Results of the x-ray diffraction (XRD) analysis on the tailings samples recovered from borehole FS15-A are shown in **Table 12**. Diffraction peaks were qualitatively identified and categorized as major, minor or trace. The upper 4 m were dominated by pyrrhotite and quartz with minor occurrences of chlorite and plagioclase. Below this, chlorite and quartz were dominant with small amounts of pyrrhotite. The amount of pyrrhotite decreased with depth until it was not detectable at a depth of 10 m. At these deeper depths, the mineralogy can be characterized as a complex assemblage with major chlorite and quartz, and minor mica, plagioclase, amphibole and calcite.

The percentage of quartz was determined on six samples **(Table 13)**. Results show that quartz is more plentiful at depth, in the tailings void of pyrrhotite. About twice the amount of quartz appeared in the four samples selected from 17, 21, 32, and 35 m (14-19%), as compared to the two samples selected from 0.5 and 1.5 m (7-8%).

Table 14 shows the results of the chemical analysis performed on the tailings solid samples recovered from borehole FS15-A. Most notable is the difference between the 2 samples taken from the upper 2 m and the other samples at depth. Near the surface, the composition of Fe and S were higher, while concentrations of AI, Ca, K, Mg and Mn were lower. These results corroborate the results from the XRD analysis, specifically, the decrease in the amount of pyrrhotite with depth. This finding indicates two layers which is illustrated in **Figure 12a** where the measured S concentrations were converted to FeS equivalent.

3.3.2 Trench samples

Detailed sampling of the upper 2.5 m of the tailings was conducted at three locations, Trenches 1-3, shown in **Figure 6**. Analytical data and field observations of color show that the tailings are layered. The layering also appears to extend the entire depth of the tailings, based on the piezocone test (Appendix D).

A 5-10 cm thick hard pan has formed on the surface of the tailings owing to sulphide oxidation. Unlike hard pans that precipitate below the depth of oxidation, the hard pan at Fault Lake has formed at the surface. Hard pans that form at or near the depth of oxidation are typically characterized by cementation of tailings by Fe(III) mineral. MINTEQA2 modelling, described later in this section, indicates that the porewaters are supersaturated with respect to these minerals, suggesting precipitation. Because the hard pan has formed at the surface, evaporative fluxes of water and solutes may have contributed to its formation. Hard pans precipitating in tailings and limit sulphide oxidation (Blowes et al. 1991). Where desiccation and weathering cracks have developed at the surface, the hard pan has apparently been lifted by ice heaving and sulphide oxidation appears to have been advanced at these spots. In areas where the hard pan has not been lifted, the depth to unoxidized tailings was approximate 20 cm.

Results from the ICP analyses on the tailings solid samples are shown in **Table 15**. Samples collected from Trench 1, located at the southerly portion of the tailings near FS13, were different in composition than those collected from Trench 2, Trench 3 and FS15-A. Most notably, iron and sulphur concentrations in Trenches 2 and 3 showed layering with low values in the range of those observed in Trench 1 (12-14% Fe and 0.8- 2.0% S) (**Fig. 13**). This indicates that the pyrrhotitic tailings are distributed at the surface in the central and northerly portions of the site. By subtracting the percent sulphate from the total sulphur composition (**Table 16**), results indicate that the sulphur at Trench 1 is primarily sulphate, while in Trenches 2 and 3 it is layered unoxidized sulphide (pyrrhotite).

Composition of other elements show that AI, Ca, K, Mg and Mn were lower in Trenches 2 and 3 than in Trench 1. In general the tailings at Trench 1 appear to resemble the composition of the pyrrhotite poor tailings found at depth at FS15.

3.4 Tailings porewater analyses

3.4.1 Borehole FS15-A samples

Table 17 shows the results of the chemical analysis conducted on porewater pneumatically extracted from the tailings samples. Concentrations of Fe, Ni, Zn and S were elevated in the upper two samples (1 & 2), and decreased with depth. **Figure 12b** shows the depth profile of Fe and S concentrations. The values measured in samples 1 and 2 are typical of sulphide oxidation. Lower values at depth may be controlled by secondary mineral precipitation/dissolution.

Porewater samples were also measured for pH, redox potential and electrical conductivity (**Table 18**). Results from samples 1 and 2 indicate sulphide oxidation. At depth, porewaters appear to be buffered by calcite (detected with XRD), which corroborates secondary mineral precipitation. For example, in the presences of calcite, Al³⁺ and Fe³⁺ precipitate as hydroxide minerals and Fe²⁺ and Mn²⁺ precipitate as a carbonate minerals. Ni²⁺ is generally soluble in the presence of calcite but may co-precipitate with iron minerals or with gypsum.

MINTEQA2 modelling (**Table 22**) of the porewater extracted from samples recovered from borehole FS15-A indicate supersaturation with respect to iron hydroxide minerals and jarosite in the 2 m of the tailings. This suggest that these minerals are precipitating. The porewaters are slightly supersaturated with respect to gypsum, which is interpreted as gypsum being at equilibrium with calcium and sulphate. The porewaters are also near saturation (slightly undersaturated) with respect to melanterite.

3.4.2 Trench samples

ICP analysis of porewaters pneumatically extracted from samples taken from the trenches show variable results (**Table 19**). Trenches 2 and 3 showed the highest concentrations of $Fe_{(T)}$, Ni, and S and showed distinctive layering (**Fig. 14**). As with the borehole FS15-A samples, the values measured are typical of sulphide oxidation. The concentrations declined with depth which suggest secondary mineral precipitation. Trench 1 generally showed lower values indicating little sulphide oxidation.

As with the borehole samples, the physico-chemical parameters measured on the extracted porewater from Trenches 2 and 3 indicated tailings oxidation (**Table 20**). The lowest measured pH was 2.1 from Trench 2, at a depth of 15 cm. The pH increased with depth. The highest measured value was 4.8. Trench 3 showed similar values but having a pH of 6.1 at the bottom of the trench. This indicates neutralization of porewaters. Trench 1 showed pH values greater than 6 below a depth of 1 m and not less than 4.4 above 1 m. Values of electrical conductivity were also less at Trench 1.

MINTEQA2 modelling of the porewater extracted from samples taken from the trenches indicates supersaturation with respect to iron hydroxide minerals and jarosite (**Table 22**). This suggest that these minerals are likely precipitating. Melanterite is near saturation and gypsum is at equilibrium with calcium and sulphate.

3.5 Acid-base accounting

A comparison of the acid potential (AP) and neutralization potential (NP) versus depth is illustrated in **Figure 12c** and data is found in **Table 21**. The AP of the upper samples clearly dominates the acid-base balance. The negative net neutralization potential (NNP) of the upper 7 m (5 samples) shows a potential of acid generation. The AP was calculated for the trench samples (**Table 16**) and show a high potential for acid generation at Trenches 2 and 3. At depth, below 14 m, the tailings are net acid neutralizing, owing to the NP/AP being greater than 2.5.

Overall, AP exceeded NP. An average of all 23 samples gave an AP of 95 kg CaCO₃/t and an NP 52 kg CaCO₃/t. Because the average NNP is -43 kg CaCO₃/t, the tailings show a potential for acid generation. The paste pH measurements of the samples (**Figure 12d**) indicate that acid generation products are present in the upper 4 meters of the tailings. Silicate minerals were also depleted in the upper 1.5-3 m of the tailings, where elsewhere they are abundant. Silicate minerals have a NP but not measured in the B.C. research test because the kinetics are slower than the length of the test. In slow moving tailings porewaters silicate mineral NP may be significant.

3.6 Pore-gas oxygen measurements

Pore-gas oxygen measurements were conducted at two depth profiles, near FS5 and FS15. The probes were installed in December 1993. At each site the probes were positioned at four depths below the hard pan. Measurements were conducted in December 1993 and May 1994 and are shown in **Table 23**. The data indicates that in December, when the surface was frozen, gaseous oxygen concentrations were near zero at the oxidized front (20 cm depth), owing to the oxidation of pyrrhotite. In May, after spring thaw, gaseous oxygen concentrations were 4-7% at the oxidation front and near zero 40-100 cm below the oxidized front.

4. DISCUSSION

Based on the x-ray diffraction analysis and elemental analysis, the Fault Lake tailings can be characterized by two distinct layers. Layer 1 is rich in pyrrhotite, and is centrally located on the tailings and in close proximity of the northerly spigot positions. Layer 1 pinches out at distal locations from the spigot positions. In the centre of the tailings Layer 1 extends from the surface to about a 9 m depth. The upper 3 m of Layer 1 is substantially sulphide rich. The sulphide composition of Layer 1 progressively decreasing with depth. Layer 2 is sulphide poor and is located below Layer 1. In the southerly portion of the tailings, where Layer 1 pinches out, Layer 2 appears to extend to the surface.

Acid-base accounting indicates substantial reserves of acid potential (AP) in Layer 1 and little in Layer 2. Neutralization potential (NP) is generally constant throughout the tailings but is depleted in the upper part of Layer 1. The net neutralization potential (NNP=NP-AP) of Layer 1 shows a potential for acid generation. Considering the extreme of the negative NNP and field observations, the upper part of Layer 1 is clearly acid generating. In fact, measurements of paste pH and extracted porewater pH indicate that net acid generation has already developed in the upper part of Layer 1. Layer 1 is unsaturated and water is not appreciably limiting oxygen availability. The observed gaseous oxygen concentrations in the surface tailings indicate active oxidation.

In Layer 2, the tailings are generally net acid neutralizing. Below a depth of 14 m in the centre of the tailings (at FS15) the NP/AP ratio ranged between 2.1 and 5.7 kg CaCO₃/tonne of tailings. The NP/AP value that indicates whether a sample can be considered acid generating is not well defined. A value less than 2.5 kg CaCO₃/tonne of tailings is often used, but kinetic test data by Morin et al. (1995) suggests the threshold could be lower.

The potential capacity of the tailings to neutralize sulphide oxidation from Layer 1 can be estimated by averaging the NNP measurements, giving a value of -43 kg CaCO₃/t. This suggests that the tailings, overall, have a theoretical potential for acid generation. However, since the duration of the B.C. Initial Test was generally 24-48 hrs, the test evaluated short-term NP (e.g., carbonate NP). Long-term NP from silicate minerals was not evaluated but is significant because silicate minerals are abundant and porewater flow velocities are low. Chlorite, mica and plagioclase are abundant in the Fault Lake tailings. At other sites (e.g., Waite Amulet), silicate minerals have been shown to be effective buffers. In addition, as the oxidation front extends into the tailings (with time), the concentration gradient of O_2 will decrease which will also decrease the flux and availability of oxygen for sulphide oxidation (St-

Arnaud 1994). A surface hard pan has also formed owing to pyrrhotite oxidation which may additionally limit oxidation, and as the hard pan develops further, the rate of oxidation may continue to decrease. In short, tailings oxidation is greatest immediately following exposure to the atmosphere and decreases with time. During the decades following deposition, carbonate minerals are highly utilized for neutralization, when ample carbonate reserves are present. As the oxidation rate decreases, long-term neutralization is available by silicate minerals.

Potential concerns include: (1) preferential flow paths through the tailings which may deplete the NP along the flow path and provide a conduit for acidity and metals mobility, though no clear evidence of this is seen; and (2) sulphide oxidation along the perimeter of the tailings where the thickness of Layer 2 may be minimal.

Sulphide oxidation in the upper part of Layer 1 has released Fe, Ni, Si, Zn and S (as SO₄), as seen in the elevated porewater concentrations. Below this, porewater concentrations of Fe, Ni, Si, Zn and S are low and pH values are near neutral. Neutral pH values suggest buffering of acidic porewaters by carbonate minerals (Blowes 1984). Trace amounts of calcite and the elevated Ca concentrations in the porewater of sample 3 support this explanation. The low concentrations of Fe, Ni, Si, Zn and S suggest precipitation, as trace amounts of gypsum have precipitated in the upper part of Layer 1.

Currently, groundwater sampled from piezometers located directly northeast of the tailings show slightly depressed pH values and depleted alkalinities, but not less than those observed in lakes to the southwest (Lakes A and B). Concentrations of nickel and iron were also marginally above-background levels, but show acidities and sulphate concentrations similar to background levels. Groundwater samples taken from other piezometers located beneath the tailings (upgradient) and from the kettle lakes located further downgradient, however, show background levels. This may suggest that the source of these oxidation products may be behind the tailings dam where ponded water has been observed. The fact that levels of pH, alkalinity and nickel improve downgradient of the tailings and are at background levels in the lakes further downgradient suggests thatmetals are effectively attenuated.

Levels of sulphate marginally above-background levels were detected below the northwesterly portion of the tailings (at FS4 and FS5). Elevated levels of sulphate were not detected in other stations to the northeast. This may be owing to the porous envelope effect and/or the flow path may be further towards the west. In the southerly portion of the tailings, limited sulphide minerals were observed. As a result, tailings porewaters generally indicated limited sulphide oxidation products. In the overburden groundwaters below the southerly portion of the tailings, above background levels of sulphate were detected. Directly downgradient of the southerly dam, above background levels of sulphate were also detected. This indicates that sulphate is leaching from the tailings and migrating south towards the New tailings area. This is not seen as being problematic because the maximum concentration detected at the downgradient station (FS1) was low (319 mg/L), and the New tailings area has its own containment.
5. CONCLUSIONS

- 1) The piezometric elevations throughout the Fault Lake site, combined with lake elevations and topographic contours, indicates that the regional groundwater flow direction south of the tailings is toward the east. North of the tailings, it is towards the northeast.
- 2) The water level in the tailings is perched higher than and fluctuates larger than the regional watertable.
- 3) The average bulk hydraulic conductivity of the glacial outwash soil material surrounding the tailings is estimated at 1.6 x 10⁻³ cm/s. The hydraulic conductivity of the Fault Lake tailings measured 1.2 x 10⁻⁵ cm/s at a mid-level depth in the tailings, and 3.6 x 10⁻⁶ cm/s in the deepest part of the tailings.
- 4) Two-dimensional groundwater flow models conducted in part 1 of the investigation (NTC 1993) showed that groundwater flow is diverted around the tailings mass due to the hydraulic conductivity contrast between the tailings and the surrounding sediments. The models also showed that flushing of the tailings mass by groundwater should not contribute significantly to the regional groundwater flow system under present water table conditions, as well as under conditions of moderate rise in water table level.
- 5) Sulphide minerals were identified in the upper 9 m of the tailings deposit and concentrated in the upper 3 m. Aerially they were concentrated in the central and northerly portions of the tailings, in proximity of the northerly spigot location.
- 6) Sulphide oxidation was detected but geochemical processes have been attenuating acidity and metals in the tailings deposit. Analysis of tailings porewater showed elevated levels of nickel, iron, and sulphate indicating the presence of sulphide oxidation products within portions of the tailings deposit near the surface. Metal concentrations are attenuated in the deeper parts of the tailings. Apparent high variability in measured metal concentrations could be caused by variations in the intensity of oxidation across the surface of the tailings due to surface effects such as drying and cracking.
- 7) Acid-base accounting indicated a net deficit in short-term neutralization potential (NP) and a potential for acid generation. Long-term NP was not evaluated but may be a significant part of total NP since porewater velocities are low and silicate minerals are abundant. In addition, as

tailings oxidation progresses (and as NP is depleted), the rate of oxidation and the release of oxidized products should decrease, owing to increased depth to unoxidized tailings. Therefore, reserves of shortterm NP are available during the early period of high oxidation rates, and reserves of silicate minerals are abundant for long-term buffering.

- 8) Sulphide oxidation has been at its highest rate since deposition discontinued in 1978, yet little impact of sulphide oxidation was observed in the groundwater of the surrounding till. The largest offsite impact is directly northeast of the tailings, where pH values are slightly depressed and alkalinities are depleted, but not less that Lakes A and B further to the southwest. Concentrations of nickel and iron are marginally above background levels. The source of the sulphide oxidation products may likely be at the tailings dam where ponding of water has been observed. Alternatively, siderite dissolution in the aquifer, precipitated during tailings deposition, may be a source for nickel and iron (Walter et al. 1994). Nevertheless, the values improve with distance, and further downgradient, the kettle lakes show background levels. This suggests qualities owing to the porous envelope effect.
- Factors which contribute to limit metal concentrations downgradient of the Fault Lake tailings are:
 - the large hydraulic conductivity contrast between the tailings and the surrounding sediments,
 - the limited infiltration through the surface of the tailings,
 - the dilution of metals flushed from the tailings by water flowing around and below the tailings, and
 - chemical attenuation of metals in the tailings and overburden.

These factors could probably be present at other locations near mine sites. Tailings deposition could be done at these sites with little effect on groundwater quality pending that thorough site evaluations are performed and that appropriate control is done at the time of deposition.

6. **RECOMMENDATIONS**

- 1) Two years of water quality monitoring have been conducted. Additional sampling of the groundwater and surface-water monitoring stations should be continued to confirm the observed trends.
- 2) The acid-base accounting on tailings samples recovered from borehole FS15-A suggest that the Layer 2 tailings will not be acid generating. Kinetic testing should be conducted to determine if/when and to what extent acidity may be generated from Layer 2 tailings. Samples are presently sealed, frozen and stored at NTC.
- 3) Avoid disturbing the tailings and exposing fresh unoxidized pyrrhotite to the atmosphere. Avoid large ponds or water covers that will dissolve secondary minerals and increase flushing to the overburden groundwater.

7. CLOSURE

Field work and preliminary data analyses for the work were performed by P. Tibble and S. Aiken. N. Michelutti assisted in field work during the summer of 1994. Numerical modelling in part 1 of the investigation was done by B. Aubé. M. Li reviewed the manuscript. L. St-Arnaud coordinated the project and reviewed the final report. M. Woyshner was the principal investigator of part 2 of the investigation. Project management from Falconbridge was provided by M. Wiseman.

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Piezometer Number	Material	K (cm/s)
FS1-A FS1-B FS1-C	Overburden Overburden Overburden	2.5E-03
FS2-A	Overburden	2.9E-03
FS2-B	Overburden	2.9E-03
FS3-A	Overburden	1.0E-02
FS3-B	Overburden	8.6E-03
FS3-C	Tailings	
FS4-A	Overburden	2.5E-05
FS4-B	Overburden	5.1E-05
FS4-C	Overburden	3.0E-05
FS5-A	Overburden	1.9E-04
FS5-B	Overburden	9.3E-05
FS5-C	Overburden	3.1E-04
FS6-A	Overburden	1.2E-04
FS6-B	Overburden	1.5E-04
FS6-C	Overburden	4.0E-04
FS7-A	Overburden	
FS8-A	Overburden	6.1E-03
FS9-A	Overburden	8.9E-05
FS9-B	Overburden	7.6E-05
FS9-C	Overburden	8.1E-04
FS10-A	Overburden	8.0E-01
FS10-B	Overburden	8.0E-01
FS11-A	Overburden	
FS12-A	Overburden	
FS13-A	Overburden	
FS14-A	Overburden	4.0E-02
FS14-B	Overburden	4.0E-02
FS15-A	Tailings	3.6E-06
FS15-B	Tailings	1.2E-05

Table 1. Measured hydraulic conductivity

1 A 1 B 1 C 2 A 2 B 2 C Tube 3 0-20 Tube 3 0-20 Tube 3 20-40 Tube 3 40-60 Tube 3 80-100 Tube 3 80-100 Tube 3 100-120 4 30-55 4 55-85 4 85-136 7 50-90 13 0-40 13 40-80 18 B 21 C 23 C	0.25-0.40 0.40-0.60 0.60-0.80 0.80-0.95 0.95-1.25 1.25-1.41 3.40-3.60 3.60-3.80 3.80-4.00 4.00-4.20 4.20-4.40 4.40-4.60 5.35-5.50 5.50-5.80 5.80-6.01 10.50-10.95 19.81-20.21 20.21-20.61 27.79-28.19 33.05-33.40 35.60-35.80	0.33 0.50 0.70 0.88 1.10 1.33 3.50 3.70 3.90 4.10 4.30 4.50 5.43 5.70 5.91 10.72 20.01 20.41 27.99 33.23 35.70	19 22 11 11 10 12 14 14 14 14 24 26 16 36 46 32 37 34 32 37 34 32 39 29

Table 2. Measured water content of tailings samplestaken from borehole FS15-A.

Sample	Sample interval (m)	Gravimetric Water Content (%)
Trench 1 T 1-A T 1-B T 1-C T 1-C T 1-D T 1-E T 1-F	0.00-0.10 0.45-0.55 0.95-1.05 1.45-1.55 1.95-2.05 2.45-2.55	39 30 32 29 32 28
Trench 2 T 2-A T 2-B T 2-C T 2-D T 2-E T 2-F T 2-G T 2-J T 2-H T 2-H T 2-I T 2-K T 2-L	0.00-0.05 0.05-0.25 0.25-0.45 0.45-0.50 0.50-0.80 0.80-1.00 1.00-1.20 1.25-1.45 1.20-1.75 2.10 2.40 2.50	8 34 13 24 17 14 13 9 8 29 22 9
Trench 3 T 3-A T 3-B T 3-C T 3-D T 3-E T 3-E T 3-F T 3-G T 3-H	0.00-0.05 0.05-0.25 0.30-0.50 0.55-0.70 0.80-1.00 1.00-1.30 1.35-1.60 2.00-2.20	13 10 10 13 21 13 13 33

 Table 3. Measured water content of tailings

 samples taken from trenches.

Sample	Depth (m)	Dry Density	Gravimetric Moisture Content %	Volumetric Moisture Content %
Trench 1 T1D1 T1D2 T1D3 T1D4 T1D5	0.50 1.00 1.50 2.00 2.50	1.45 1.51 1.39 1.44 1.83	19 35 18 29 17	28 53 26 42 31
Trench 2 T2D1 T2D2 T2D3 T2D4 T2D5	0.50 1.00 1.50 2.00 2.50	2.51 2.58 2.62 2.73 2.66	10 10 7 6 7	25 26 18 15 17
Trench 3 T3D1 T3D2 T3D3 T3D4	0.50 1.00 1.50 2.00	2.11 2.15 2.61 2.42	7 6 3 5	14 13 8 12

Table 4. Measured water content and dry density of tailings taken from trenches.

Sample	Field Temperature (C)	Field REDOX Potential (mV)	Field Electrical Conductivity (mS/cm)	Field pH	Lab pH	Lab Acidity (mg/L CaCO3)
FS1-A	3.2	231	0.519	7.40	8.17	8
FS1-B	3.4	230	0.804	7.40	6.20	< 4
FS1-C	3.5	229	0.207	7.40	8.15	8
FS2-A	4.3	250	0.280	7.30	8.17	8
FS2-B	6.3	251	0.127	7.40	7.77	8
FS3-B	7.0	220	0.980	7.10	7.94	8
FS8-A	4.6	200	0.200	6.20	7.17	12
FS9-A	5.2	240	0.230	5.50	7.95	8
FS9-B	3.8	-40	0.228	6.70	8.16	6
FS9-C	3.2	110	0.098	6.40	7.55	8
FS10-A	5.3	20	0.077	6.60	7.93	8
FS10-B	5.3	160	0.110	6.00	6.68	14

 Table 5. Physico-chemical parameter values for groundwater samples

 taken from piezometers at Fault Lake, December 1993.

	Field measure	ements				Lab measure	ments				
Sample	Temperature (C)	REDOX Potential (mV)	Electrical Conductivity (mS/cm)	рН	Alkalinity * (mg CaCO3/L)	Temperature (C)	REDOX Potential (mV)	Electrical Conductivity (mS/cm)	рH	Alkalinity (mg CaCO3/L)	Acidity (mg/L CaCO3)
FS1-A FS1-B FS1-C FS2-A FS2-B FS3-B	10.5 9.6 11.0 	234 246 231 355 269 180	0.404 0.386 0.366 0.287 0.122 0.984	7.42 64 7.30 6.89 7.48 80 6.51 7.60 7.04 84 6.96		16.4 15.8 16.3 19.0 19.3 19.0	236 224 224 282 270 248	0.522 0.748 0.197 0.289 0.114 1.061	7.88 8.01 5.98 8.18 6.74 7.92	78 80 52 84 20 74	< 2 < 2 < 2 < 2 < 2 < 2 < 2 < 2 < 2
FS4-A FS4-B FS4-C		644 175 221	0.541 0.342 0.290	7.04 84 6.96 7.21		18.0 17.8 18.0	202 242 228	0.527 0.346 0.292	7.85 7.72 8.18	90 58 66	6 8 < 2
FS5-A FS5-B FS5-C		218 178 51	0.420 0.434 0.569	7.27 7.40 6.94	97 	19.0 19.0 19.3	269 257 270	0.404 0.437 0.546	8.30 8.01 7.69	104 132 134	< 2 10 10
FS6-A		207 173	0.328	6.81 4 13	63	19.5 16 3	262	0.350	7.92 5.90	80 8	4
FS9-A FS9-B FS9-C	11.8 12.6 11.1	221 262 243	0.227 0.227 0.080	6.29 6.03 4.56	54 46 	19.2 19.0 19.0	273 160 180	0.229 0.235 0.079	7.71 7.69 5.34	62 66 6	8 8 16
FS10-A FS10-B	9.2 12.8	165 335	0.071 0.134	6.15 3.09	19 	18.7 18.9	254 367	0.074 0.133	6.97 4.72	22 6	8 38
FS13-A FS15-A FS15-B		160 254 166	2.31 1.913 3.33	6.61 7.2 7.23		15.4 16.6 16.0	187 247 241	2.29 1.938 3.38	6.94 8.01 7.88	170 246 268	38 28 12
Lake A Lake B Lake 1 Lake 2 Lake 3 Lake 4	 18 19 19.1 ₎	375 282 199 247 211 181	0.078 0.039 0.194 0.099 0.177 0.235	4.54 5.08 7.8 6.75 7.92 8.05		20 20.3 20.5 20 20 20 20	361 317 262 251 236 250	0.075 0.038 0.203 0.094 0.185 0.245	4.58 5.11 7.8 7.07 8.13 8.08	< 2.0 4 70 32 64 64	14 < 2 < 2 6 < 2 < 2

Table 6. Physico-chemical parameter values for water samples taken from monitoring stations at Fault Lake, May 1994.

Field alkalinities were measured with a HACH hand-held digital titator.

Sample	Field Temperature (C)	Field REDOX Potential (mV)	Field Electrical Conductivity (mS/cm)	Field pH	Lab Acidity (mg/L CaCO3)	Lab Alkalinity (mg/L CaCO3)
FS1-A FS1-B FS1-C	15 17.4 	201 211 	0.615 0.857 	6.83 7.01 	< 2 < 2 	70 75
FS2-A FS2-B	19.7 18	180 186	0.298 0.132	6.89 6.97	2 	75 20
FS3-A	16.6	171	1.17	7.51	5	90
FS3-C	20.2	171	0.84	 6.86	6	155
FS4-A FS4-B FS4-C	16.7 17.4 15	165 168 174	0.595 0.375 0.331	6.14 6.08 6.7	10 5 3	85 50 70
FS5-A FS5-B FS5-C	15.5 15.5 17.5	159 158 158	0.445 0.463 0.62	6.35 6.57 6.55	10 < 2 < 2	130 120 150
FS6-A FS6-B FS6-C	14.1 	154 	0.366 	6.88 	2 	80
FS7-A					<u></u> *	
FS8-A						
FS9-A FS9-B FS9-C	18.7 19.8 	163 164 	0.254 0.274 	5.93 6.11 	< 2 3 	70 100
FS10-A FS10-B	19.7 19	167 171	0.084 0.111	5.32 4.32	< 2 8	30 10
FS11-A						
FS12-A						
FS13-A	16.7	182	1.87	7.2	29	105
FS14-A FS14-B	18.4 	178 - 	0.129	6.31 	< 2 	40
FS15-A FS15-B	 18.4	 197	 4.2	 6.4	27	270
Lake-A Lake-B Lake-1 Lake-2 Lake-3 Lake-4 Lake-5	20.8 23 24 20.7 21.2 21 20.7	167 237 228 156 152 161 156	0.069 0.037 0.181 0.184 0.087 0.241 0.218	4.05 4.06 5.77 6.35 6.21 6.45 6.47	< 2 2 2 5 < 2 3 < 2	5 5 35 55 25 60 40

 Table 7. Physico-chemical parameter values for water samples

 taken from monitoring stations at Fault Lake, August 1994.

Sample	Sample Number	Al (mg/L)		As (mg/L)	Ca (mg/L)		Cd (mg/L)		Cu (mg/L)	(Fe (mg/L)		K (mg/L)	Mg (mg/L)		Mn (mg/L)	Na (mg/L)		Ni (mg/L)		Pb (mg/L)	S (mg/L)		Se (mg/L)		Zn (mg/L)		Fe+3 (mg/L)	Cl (mg/L)
FS1-A FS1-B FS1-C	93227 93229 93231	0.30 0.30 0.46	6) <) <	0.250 0.250 0.250	83.792 130.617 38.565	v v v	0.025 0.025 0.025	~ ~ ~	0.025 0.025 0.025	~ ~ ~	0.025 0.025 0.025	V V V	5.000 5.000 5.000	21.392 31.026 7.086	~ ~ ~	0.005 0.005 0.005	20.704 42.731 18.504	V V V	0.025 0.025 0.025	~ ~ ~	0.250 0.250 0.250	75.242 134.753 22.447	< < <	0.500 0.500 0.500	~ ~ ~	0.025 0.025 0.025	~ ~ ~	0.050 0.050 0.050	5.980 12.100 3.600
FS2-A FS2-B	93233 93235	0.32 0.30	4 <) <	0.250 0.250	38.651 15.778	v v	0.025 0.025	۷ ۷	0.025 0.025	<	0.428 0.025	<	6.974 5.000	9.469 3.957	< <	0.005 0.005	13.743 9.702	< <	0.025 0.025	< <	0.250 0.250	14.625 11.904	< <	0.500 0.500	<	0.025 0.025	<	0.500 0.050	18.200 4.040
FS3-B	93237	0.29	4 <	0.250	198.437	<	0.025	<	0.025		0.031		6.065	23.891		0.282	33.508	<	0.025	<	0.250	187.115	<	0.500	<	0.025	<`	0.050	16.900
. FS8-A	93239	0.31	5 <	0.250	13.625	<	0.025	<	0.025	· :	23.541	<	5.000	3.178		4.770	12.350	<	0.025	<	0.250	15.027	<	0.500	<	0.025		1.150	2.020
FS9-A	93241	0.25	3 <	0.250	31.883	<	0.025	<	0.025		0.204	<	5.000	5.618		0.471	13.394		0.079	<	0.250	10.827	<	0.500		0.049	<	0.050	12.800
FS9-B FS9-C	93243 93245	0.35) < 3 <	0.250	34.260	<	0.025	<	0.025		0.119	< <	5.000	5.239 1.377		0.071 0.874	17.633 13.797	<	0.025 0.418	v v	0.250 0.250	9.825 9.406	< <	0.500	< <	0.025	< <	0.050	10.700
5040 A	000.47	< 0.05		0.050	0.949		0.005		0.005		0.007		5 000	1 500		4.405	40.400		0.055		0.050	0.070		0.500				0.050	
FS10-A FS10-B	93247 93249	< 0.25) < <	0.250	9.343 7.647	< <	0.025	< <	0.025		0.807	< <	5.000	1,582 1,402		1.165 2.576	10.428 14.538		0.055	V . V	0.250	2.076 11.981	< <	0.500	< <	0.025	< <	0.050 0.050	2.270 2.320
				Í																									

Table 8. Major metal and ion concentrations in groundwater samples taken from piezometers at Fault Lake, December 1993.

Table 9. Major metal and ion concentrations in samples taken from surface and ground water monitoring stations at Fault Lake, May 1994.

Sample	Sample Number	Al mg/L	As mg/L	Ca mg/L	Cd mg/L	Co mg/L	Cr mg/L	Cu mg/L	Fe mg/L	K ∙mg/L	Mg mg/L	Mn mg/L	Na mg/L	Ni mg/L	Pb mg/L	S mg/L	Sb mg/L	Se mg/L	Si mg/L	Te mg/L	Ti mg/L	Zn mg/L	Fe+3 mg/L	Ci mg/L	SO4 mg/L
FS1-A FS1-B FS1-C	94747 94748 94749	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	82.180 123.194 22.344	< 0.025 < 0.025 < 0.025	< 0.02 < 0.02 < 0.02	5 < 0.025 5 < 0.025 5 < 0.025 5 < 0.025	< 0.025 < 0.025 < 0.025	< 0.025 < 0.025 < 0.025	< 5.000 < 5.000 < 5.000	20.855 30.174 4.407	< 0.005 < 0.005 < 0.005	15.577 27.891 9.729	< 0.025 < 0.025 < 0.025	< 0.250 < 0.250 < 0.250	71.293 122.466 9.490	< 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500	8,978 8,926 5,447	< 0.100 < 0.100 < 0.100	< 0.250 < 0.250 < 0.250	< 0.025 < 0.025 < 0.025	< 0.050 < 0.050 < 0.050	6.310 10.500 2.150	210.000 318.000 32.400
FS2-A FS2-B	94750 94751	< 0.250 < 0.250	< 0.250 < 0.250	36.874 11.847	< 0.025 < 0.025	< 0.02 < 0.02	5 < 0.025 5 < 0.025	< 0.025 < 0.025	< 0.025 < 0.025	< 5.000 < 5.000	9.318 3.502	< 0.005 < 0.005	10.780 5.366	< 0.025 < 0.025	< 0.250 < 0.250	12.519 8.715	< 0.250 < 0.250	< 0.500 < 0.500	6.336 6.614	< 0.100 < 0.100	< 0.250 < 0.250	< 0.025 < 0.025	< 0.050 < 0.050	14.700 1.770	41.000 29.600
FS3-B	94752	< 0.250	< 0.250	201.518	< 0.025	< 0.02	5 < 0.025	< 0.025	< 0.025	< 5.000	31.425	94.470	32.671	< 0.025	< 0.250	188.403	< 0.250	< 0.500	5.992	< 0.100	< 0.250	< 0.025	< 0.050	23.600	412.000
FS4-A FS4-B FS4-C	94753 94754 94755	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	81.060 42.702 33,927	<pre>< 0.025 < 0.025 < 0.025 < 0.025</pre>	< 0.02 < 0.02 26.07	5 < 0.025 5 < 0.025 5 < 0.025 5 < 0.025	< 0.025 < 0.025 < 0.025	1.527 < 0.025 < 0.025	7.673 7.197 < 5.000	18.956 9.550 6.761	0.444 0.201 36.220	15.002 13.223 16.444	< 0.025 < 0.025 < 0.025	< 0.250 < 0.250 < 0.250	58.286 16.573 10.335	< 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500	6.505 6.716 3.260	< 0.100 < 0.100 0.133	< 0.250 < 0.250 < 0.250	 < 0.025 < 0.025 0.029 	0.440 < 0.050 < 0.050	19.900 42.900 53.700	166.000 50.600 49.800
FS5-A FS5-B FS5-C	94756 94757 94758	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	68.931 58.314 101.736	< 0.025 < 0.025 < 0.025	< 0.02 < 0.02 < 0.02	5 < 0.025 5 < 0.025 5 < 0.025 5 < 0.025	< 0.025 < 0.025 < 0.025	0.068 < 0.025 1.381	5.478 < 5.000 8.216	10.158 9.357 15.750	69.150 0.276 0.195	11.799 33.082 12.516	0.042 0.038 0.068	< 0.250 < 0.250 < 0.250	33.826 35.712 64.199	< 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500	6.621 6.571 5.472	< 0.100 < 0.100 < 0.100	< 0.250 < 0.250 < 0.250	0.051 < 0.025 < 0.025	< 0.050 < 0.050 1.160	6.620 4.760 5.530	97.700 104.000 187.000
FS6-A	94759	< 0.250	< 0.250	40.739	< 0.025	< 0.02	5 < 0.025	< 0.025	< 0.025	< 5.000	8.342	0.128	20.404	< 0.025	< 0.250	12.462	< 0.250	< 0.500	4.711	< 0.100	< 0.250	< 0.025	< 0.050	35.600	35.600
FS8-A	94760	< 0.250	< 0.250	8.114	< 0.025	61.56	< 0.025	< 0.025	7.946	< 5.000	1.867	2.688	7.766	0.544	< 0.250	9.656	< 0.250	< 0.500	10.428	0.114	< 0.250	0.151	0.543	< 0.200	27.000
FS9-A FS9-B FS9-C	94761 94762 94763	< 0.250 < 0.250 0.275	< 0.250 < 0.250 < 0.250	29.000 29.245 4.218	< 0.025 < 0.025 < 0.025	< 0.02 < 0.02 < 0.02	5 < 0.025 5 < 0.025 5 < 0.025 5 < 0.025	< 0.025 < 0.025 < 0.025	< 0.025 < 0.025 0.794	< 5.000 < 5.000 < 5.000	5.868 5.923 1.299	0.113 0.400 0.437	11.168 11.031 8.437	0.066 < 0.025 0.407	< 0.250 < 0.250 < 0.250	9.770 9.959 8.846	< 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500	6.577 6.433 7.097	< 0.100 < 0.100 < 0.100	< 0.250 < 0.250 < 0.250	0.089 < 0.025 0.055	< 0.050 < 0.050 < 0.050	11.400 11.400 < 0.200	27.800 28.100 25.600
FS10-A FS10-B	94764 94765	< 0.250 2.889	< 0.250 < 0.250	5.281 3.146	< 0.025 < 0.025	28.97 0.10	0 < 0.025 4 < 0.025	< 0.025 74,460	0.783 0.686	< 5.000 < 5.000	1.445 1.219	0.865 1.682	7.397 11.205	< 0.025 1.200	< 0.250 < 0.250	2.978 9.235	< 0.250 < 0.250	< 0.500 < 0.500	4.276 8.747	< 0.100 < 0.100	< 0.250 < 0.250	< 0.025 0.271	0.617 0.558	1.910 1.800	9.430 27.600
FS13-A	94766	< 0.250	< 0.250	466.454	< 0.025	< 0.02	5 < 0.025	< 0.025	0.307	72.296	193.768	. 1.191	28.177	0.059	< 0.250	616.943	0.252	< 0.500	4.476	< 0.100	< 0.250	< 0.025	< 0.050	2.710	
FS15-A FS15-B	94767 94768	< 0.250 < 0.250	< 0.250 < 0.250	155.217 334.953	< 0.025 < 0.025	< 0.02 < 0.02	5 < 0.025 5 < 0.025	< 0.025 < 0.025	0.271 3.678	95.339 148.746	136.439 291.771	0.301 0.337	159.479 331.551	0.078 0.139	< 0.250 < 0.250	365.951 859.588	< 0.250 0.308	< 0.500 < 0.500	4.994 6.897	0.135 0.157	< 0.250 < 0.250	< 0.025 < 0.025	< 0.050 < 0.050	38.200 40.600	
Lake A Lake B Lake 1 Lake 2 Lake 3 Lake 4	94741 94742 94743 94744 94745 94745 94746	0.274 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	2.806 < 0.050 22.729 9.011 24.177 30.107	< 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	 < 0.02 	5 <	81.580 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	0.057 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	< 5.000 < 5.000 < 5.000 < 5.000 < 5.000 < 5.000	1.726 0.932 6.133 3.021 4.220 5.606	0.083 0.056 < 0.005 < 0.005 < 0.005 < 0.005 < 0.005	3.867 3.580 8.539 4.662 6.912 11.192	0.273 0.083 < 0.025 < 0.025 < 0.025 < 0.025	< 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	9.985 4.139 7.874 5.116 7.472 8.014	< 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500 < 0.500 < 0.500 < 0.500	0.314 0.406 0.171 0.252 0.071 0.457	< 0.100 < 0.100 < 0.100 < 0.100 < 0.100 < 0.100	< 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	0.051 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	< 0.050 < 0.050 < 0.050 < 0.050 < 0.050 < 0.050		

Table 10. Major metal and ion concentrations in samples taken from surface and ground water monitoring stations at	: Fault La	ike, August 1994
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Sample	Sample	Al	As	Ca	Cd	Co	Cr	Cu	Fe	K	Mg	Mn	Na	Ni	Pb	S	Sb	Se	Si	Te	Ti	Zn	Fe+3	CI	SO4
	Number	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L
FS1-A	941804	0.753	< 0.250	89.665	< 0.025	< 0.025	< 0.025	< 0.025	0.040	< 5.000	21.289	< 0.005	18.737	< 0.025	< 0.250	78,855	< 0.250	< 0.500	9.254	< 0.100	< 0.250	< 0.025	< 0.150	6.200	216.000
FS1-B	941805	0.653	< 0.250	127.567	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	< 5.000	31.143	< 0.005	27.318	< 0.025	< 0.250	118.323	< 0.250	< 0.500	9.246	< 0.100	< 0.250	< 0.025	0.150	9.600	319.000
FS2-A	941806	0.659	< 0.250	40.235	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	< 5.000	9.746	< 0.005	13.246	< 0.025	< 0.250	15.447	< 0.250	< 0.500	6.403	< 0.100	< 0.250	< 0.025	< 0.150	14.700	42.600
FS2-B	941807	0.617	< 0.250	15.681	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	< 5.000	3.501	0.005	8.449	< 0.025	< 0.250	10.903	< 0.250	< 0.500	6.559	< 0.100	< 0.250	< 0.025	< 0.150	2.220	30.300
FS3-A	941808	0.755	< 0.250	201.772	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	6.149	32.081	0.132	34.732	< 0.025	< 0.250	192.051	< 0.250	< 0.500	6.062	< 0.100	< 0.250	< 0.025	0.240	24,100	534.000
FS3-C	941809	0.755	< 0.250	171.705	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	15.346	12.390	0.159	12.367	1.428	< 0.250	111.429	< 0.250	< 0.500	4.700	< 0.100	< 0.250	< 0.025	< 0.150	2.810	303.000
FS4-A	941810	0.533	< 0.250	82.214	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	8.757	19.253	0.479	15.766	< 0.025	< 0.250	62.824	< 0.250	< 0.500	6.196	< 0.100	< 0.250	< 0.025	< 0.150	20.400	190.000
FS4-B	941811	0.662	< 0.250	42.334	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	9.189	9.923	0.253	14.817	< 0.025	< 0.250	18.331	< 0.250	< 0.500	7.367	< 0.100	< 0.250	< 0.025	0.160	44.200	57.200
FS4-C	941812	0.771	< 0.250	40.038	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	5.740	7.625	0.016	16.897	< 0.025	< 0.250	11.499	< 0.250	< 0.500	3.252	< 0.100	< 0.250	< 0.025	< 0.150	39.800	35.600
FS5-A FS5-B FS5-C	941813 941814 941815	0.779 0.622 0.619	< 0.250 < 0.250 < 0.250	74.302 69.935 106.146	< 0.025 < 0.025 < 0.025	< 0.025 < 0.025 < 0.025	< 0.025 < 0.025 < 0.025	< 0.025 < 0.025 < 0.025	0.274 0.474 0.052	6.361 6.106 9.028	10.017 10.558 15.439	0.088 0.376 0.222	9.973 25.042 14.377	< 0.025 < 0.025 < 0.025 < 0.025	< 0.250 < 0.250 < 0.250	37.006 39.779 65.883	< 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500	6.609 6.839 5.434	< 0.100 < 0.100 < 0.100	< 0.250 < 0.250 < 0.250	< 0.025 < 0.025 < 0.025 < 0.025	0.380 0.470 0.210	5.710 4.950 6.920	112.000 120.000 191.000
FS6-A	941816	0.597	< 0.250	46.945	< 0.025	< 0.025	< 0.025	< 0.025	0.107	< 5.000	8.962	0.133	21.689	< 0.025	< 0.250	14.418	< 0.250	< 0.500	4.839	< 0.100	< 0.250	< 0.025	0.280	36.900	37.100
· FS9-A	941817	0.687	< 0.250	39.322	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	< 5.000	6.057	0.067	12.693	0.077	< 0.250	10.637	< 0.250	< 0.500	6.631	< 0.100	< 0.250	0.031	< 0.150	11.800	28.400
FS9-B	941818	0.741	< 0.250	40.210	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	< 5.000	6.387	0.526	14.628	0.084	< 0.250	11.773	< 0.250	< 0.500	6.007	< 0.100	< 0.250	< 0.025	< 0.150	11.700	29.500
FS10-A	941819	0.647	< 0,250	11.289	< 0.025	< 0.025	< 0.025	< 0.025	0.343	< 5.000	1.803	0,869	8.520	0.041	< 0.250	4.204	< 0.250	< 0.500	4.108	< 0.100	< 0.250	< 0.025	0.360	2.470	12.300
FS10-B	941820	1.844	< 0,250	9.009	< 0.025	0.031	< 0.025	0.028	0.168	< 5.000	1.323	1,158	9.969	0.739	< 0.250	7.122	< 0.250	< 0.500	6.967	< 0.100	< 0.250	0.177	0.280	1.970	19.300
FS13-A	941821	0.743	< 0.250	433.734	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	68.121	168.250	1.061	26.772	0,116	< 0.250	558.560	< 0.250	< 0.500	4.368	< 0.100	< 0.250	< 0.025	< 0.150	3.570	276.000
FS14-A	941822	0.562	< 0.250	19.894	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	< 5.000	3.947	0.015	7.314	< 0.025	< 0.250	6,105	< 0.250	< 0.500	0.968	< 0.100	< 0.250	< 0.025	< 0.150	2.410	17.500
FS15-B	941823	0.510	< 0.250	174.047	< 0.025	< 0.025	< 0.025	< 0.025	< 0.025	15.208	12.488	0,167	12.590	1.278	< 0.250	114.374	< 0.250	< 0.500	4.612	< 0.100	< 0.250	< 0.025	< 0.150	39.900	550.000
Lake-A Lake-B Lake-1 Lake-2 Lake-3 Lake-4 Lake-5	941645 941646 941647 941648 941649 941650 941651	0.300 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	5.297 1.560 16.376 24.186 10.073 28.878 24.277	< 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	< 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	< 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	0.065 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	< 0.025 0.217 0.025 0.025 0.025 0.025 0.025 0.025	< 5.000 < 5.000 < 5.000 < 5.000 < 5.000 < 5.000 < 5.000	1.975 0.816 5.776 3.948 2.863 5.231 6.771	0.035 0.012 < 0.005 < 0.005 < 0.005 < 0.005 < 0.005 < 0.005	5.420 5.134 9.952 8.513 6.006 12.444 10.079	0.196 0.028 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	< 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	7.235 2.367 6.872 7.116 4.740 7.643 17.833	< 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500 < 0.500 < 0.500 < 0.500 < 0.500 < 0.500	0.142 0.635 0.177 0.493 0.179 0.401 0.126	< 0.100 < 0.100 < 0.100 < 0.100 < 0.100 < 0.100 < 0.100 < 0.100	< 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250 < 0.250	< 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025 < 0.025	< 0.150 < 0.150 < 0.150 < 0.150 < 0.150 < 0.150 < 0.150 < 0.150	2.580 2.690 15.200 8.850 2.900 15.700 5.640	22.400 9.960 21.600 22.500 16.000 23.300 53.300

Sample	Sample [.] Number	Al mg/L	As mg/L	Ca mg/L	Cd mg/L	Co mg/L	Cr mg/L	Cu mg/L	Fe mg/L	K mg/L	Mg mg/L	Mn mg/L	Na mg/L	Ni mg/L	Pb mg/L	S mg/L	Sb mg/L	Se mg/L	Si mg/L	Te mg/L	Ti mg/L	Zn mg/L	Fe+3 mg/L	Ci mg/L	SO4 mg/L
December 1993					L		L	Ĺł			1			17.7	1	1		L		1	Į				
Prepared spike	-	5.000	2.000	20.000	2.000			5.000	20.000	5.000	20.000	5.000	5.000	5.000	5.000			0.000				20.000	20.000		
Analysis of spike	93253 93254 93255	5.465 5.445 5.335	1.958 1.976 1.905	21.375 21.353 21.405	2.127 2.121 2.134			5.476 5.484 5.472	21.879 21.845 21.845	5.751 < 5.000 5.880	19.900 19.907 19.871	5.069 5.055 5.067	12.421 12.201 12.427	5.630 5.590 5.570	5.469 5.528 5.460	965.970 961.480 963.840		< 0.500 < 0.500 < 0.500				22.092 22.077 22.127	19.600 14.600 14.500	394.000 419.000 421.000	
Analysis of field blanks	93250 93251 93252	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	0.689 0.572 0.528	< 0.025 < 0.025 < 0.025			< 0.025 < 0.025 < 0.025	< 0.025 < 0.025 < 0.025	< 5.000 < 5.000 < 5.000	< 0.500 < 0.500 < 0.500	< 0.005 < 0.005 < 0.005	8.004 7.249 7.186	< 0.025 < 0.025 < 0.025	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	 	< 0.500 < 0.500 < 0.500		 		< 0.025 < 0.025 < 0.025	< 0.050 < 0.050 < 0.050	1.690 1.740 1.760	
May 1994	lay 1994																								
Prepared spike	-	5.000	2.000	20.000	2.000	0.000	0.000	5.000	20.000	5.000	20.000	5.000	5.000	5.000	5.000		0.000	0.000	0.000	0.000	0.000	20.000	20.000		-
Analysis of spike	94784 94785 94786	4.865 4.861 4.931	2.030 1.960 2.034	18.811 19.153 19.040	2.203 2.214 2.211	< 0.025 < 0.025 < 0.025	< 0.025 < 0.025 < 0.025	5.189 5.288 5.224	21.082 21.453 21.288	< 5.000 < 5.000 6.400	19.243 19.411 19.339	5.035 5.098 5.124	10.566 10.467 10.374	5.646 5.750 5.738	5.768 5.796 5.805	927.585 939.676 944.439	< 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500	7.471 7.587 7.550	< 0.100 < 0.100 < 0.100	< 0.250 < 0.250 < 0.250	23.760 23.996 24.086	19.100 18.800 19.100		-
Analysis of field blanks	94781 94782 94783	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	< 0.050 < 0.050 < 0.050	< 0.025 < 0.025 < 0.025	< 5.000 < 5.000 < 5.000	< 0.500 < 0.500 < 0.500	< 0.005 < 0.005 < 0.005	5.769 5.536 5.578	< 0.025 < 0.025 < 0.025	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500	0.215 0.243 0.214	< 0.100 0.124 < 0.100	< 0.250 < 0.250 < 0.250	< 0.025 < 0.025 < 0.025	< 0.050 < 0.050 < 0.050		-				
August 1994											,							·		·				ł	
Prepared spike		5.000	0.000	0.000	0.000	0.000	0.000	0.000	20.000	0.000	0.000	5.000	-	2.000	0.000		0.000	0.000	0.000	0,000	0.000	7.000	20.000		
Analysis of spike	941655 941656	4.863 4.880	< 0.250 < 0.250	< 0.050 < 0.050	< 0.025 < 0.025	< 0.025 < 0.025	< 0.025 < 0.025	< 0.025 < 0.025	20.239 20.213	< 5.000 < 5.000	< 0.500 < 0.500	4.883 4.873	4.364 4.354	2.003 2.007	< 0.250 < 0.250	20.616 20.466	< 0.250 < 0.250	< 0.500 < 0.500	< 0.050 < 0.050	< 0.100 < 0.100	< 0.250 < 0.250	6.997 6.964	19.100 19.200	-	
Equip. blank Equip. blank Travel blank	941652 641653 941654	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	< 0.050 < 0.050 < 0.050	< 0.025 < 0.025 < 0.025	< 5.000 < 5.000 < 5.000	< 0.500 < 0.500 < 0.500	< 0.005 < 0.005 < 0.005	4.241 4.483 4.417	< 0.025 < 0.025 < 0.025	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	< 0.250 < 0.250 < 0.250	< 0.500 < 0.500 < 0.500	< 0.050 0.073 0.059	< 0.100 < 0.100 < 0.100	< 0.250 < 0.250 < 0.250	< 0.025 < 0.025 < 0.025 < 0.025		2.330 2.390 2.300	2.740 2.660 2.610				

Table 11. Quality assurance and quality control of Fault Lake monitoring station water sampling.

Sample	Depth (m)	Pyrrhotite	Chlorite	Mica	Plagioclase	Amphibole	Quartz	Calcite	Gypsum
	0.5	maior	minor	nd	minor	nd	maior	nd	trace
2	1.5	major	minor	nd	minor	nd	major	nd	trace
3	3.0	major	minor	minor	minor	minor	major	trace	trace
4	4.6	minor	maior	minor	minor	minor	major	minor	nd
5	6.2	minor	major	minor	minor	minor	maior	minor	nd
6	7.8	trace	maior	minor	minor	minor	major	minor	nd
7	9.4	trace	major	minor	minor	minor	major	minor	nd
8	11.0	minor	major	minor	minor	minor	major	minor	nd
9	12.6	minor	major	minor	minor	minor	major	minor	nd
10	14.2	minor	major	minor	minor	minor	major	minor	nd
11	15.8	minor	major	minor	minor	minor	major	minor	nd
12	17.4	minor	major	minor	minor	minor	major	minor	nd
13	19.0	minor	major	minor	minor	minor	major	minor	nd
14	20.6	minor	major	minor	minor	minor	major	minor	nd
15	22.2	minor	major	minor	minor	minor	major	minor	nd
16	23.8	minor	major	minor	minor	minor	major	minor	nd
17	25.4	minor	major	minor	minor	minor	major	minor	nd
18	27.0	minor	major	minor	minor	minor	major	minor	nd
19	28.6	minor	major	minor	minor	minor	major	minor	nd
20	30.2	minor	major	minor	minor	minor	major	minor	nd
21	31.8	minor	major	minor	minor	minor	major	minor	nd
22	33.4	minor	major	minor	minor	minor	major	minor	nd
23	35.0	minor	major	minor	minor	minor	major	minor	nd

Table 12. Interpretation of x-ray diffraction patterns on tailings samples recovered from borehole FS15-A.

nd = not detected

Table 13. Interpretation by A tailings samples.	Lakefield Research Labo	oratories of x-ray diffrac	tion patterns on FS15-

Sample	1	. 2	12	14	21	23
Depth (m)	0.5	1.5	17.4	20.6	31.8	35.0
Pyrrhotite, Fe _(1-x) S	major	major	-	-	-	-
Chlorite, (Mg,Fe,Al)₅(Si,Al)₄O ₁₀ (OH) ₈	minor	minor	major	major	major	major
Mica, (K,Na)(Al,Mg,Fe) ₂ (Si,Al)O ₁₀ (OH) ₂	-	-	minor	minor	minor	minor
Plagioclase, (Na,Ca)(Si,Al) ₀₈	minor	minor	minor	minor	minor	minor
Amphibole, (Na,K),Ca ₂ (Fe,Mg) ₅ (Al,Si) ₈ O ₂₂ (OH) ₂	-	-	minor	minor	minor	minor
Quartz, SiO ₂ (free silica determination)	minor (6.8%)	minor (7.9%)	minor (13.9%)	minor (14.0%)	minor (18.6%)	minor (15.3%)
Calcite, CaCO ₃	-	-	minor	minor	minor	minor
Gypsum, CaSO₄●2H₂O	trace	trace	-	-		-

Sample	Depth (m)	Al (%)	As (ug/g)	Ca (%)	Cd (ug/	(g)	Cu (ug/g)	Fe (%)	K (%)	Mg (%)	Mn (%)	Na (%)	Ni (%)	Pb (ug/g)	S (%)	Se (ug/g)	Zn (ug/g)	CI (ug	/g)
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23	0.50 1.50 4.00 5.50 7.50 9.65 10.70 12.00 13.70 15.20 16.00 18.00 20.00 22.00 23.30 24.30 26.30 28.40 29.30 31.00 33.50 34.00 36.80	2.05 2.10 3.11 4.88 4.22 5.18 4.57 5.59 5.39 5.35 4.67 5.72 4.73 4.85 4.83 5.11 5.56 5.42 5.65 5.42 5.63 5.42 5.63 5.65 5.42 5.63 5.63 5.65 5.42 5.63 5.63 5.65 5.42 5.63 5.65 5.42 5.63 5.65 5.42 5.63 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.42 5.65 5.45 5.65 5.42 5.65 5.42 5.65 5.42 5.65	537.82 410.09 460.52 572.88 487.32 544.42 531.10 610.75 554.14 522.45 442.58 571.25 488.64 463.35 479.20 444.03 569.12 603.52 644.58 616.29 576.13 579.49 574.19	0.65 0.98 1.70 2.72 2.05 2.79 2.91 2.92 2.91 2.80 3.04 3.14 3.10 2.83 2.78 2.91 3.55 2.90 2.57 2.80 2.57 2.80 2.57	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	2.00 2.00 2.00 2.00 2.00 2.00 2.00 2.00	965.55 574.16 670.11 520.06 442.83 660.66 430.41 398.94 424.93 238.17 250.05 354.90 249.90 277.30 255.84 200.03 297.49 326.75 412.55 358.66 318.56 202.79 302 39	40.28 37.61 30.82 17.34 19.17 13.89 13.98 13.95 13.24 12.46 11.08 13.94 11.35 11.24 10.85 11.50 13.08 12.74 12.95 12.25 12.83 12.91 13.01	0.12 0.17 0.29 0.45 0.36 0.42 0.38 0.42 0.43 0.43 0.44 0.43 0.44 0.43 0.44 0.43 0.44 0.43 0.44 0.43 0.44 0.43 0.44 0.43 0.44 0.43 0.45 0.42	$\begin{array}{c} 1.04\\ 1.30\\ 1.97\\ 3.07\\ 2.67\\ 3.38\\ 2.92\\ 3.68\\ 3.62\\ 3.59\\ 3.05\\ 3.80\\ 3.13\\ 3.15\\ 3.32\\ 3.40\\ 3.73\\ 3.64\\ 3.67\\ 3.65\\ 3.71\\ 3.63\\ 3.62\end{array}$	0.04 0.05 0.08 0.12 0.10 0.13 0.12 0.14 0.14 0.14 0.12 0.13 0.12 0.13 0.12 0.13 0.12 0.13 0.14 0.14 0.14 0.14 0.14 0.14	$\begin{array}{c} 0.34\\ 0.51\\ 0.49\\ 0.57\\ 0.50\\ 0.53\\ 3.26\\ 0.54\\ 0.52\\ 0.50\\ 0.52\\ 0.53\\ 0.49\\ 0.51\\ 0.52\\ 0.51\\ 0.52\\ 0.51\\ 0.55\\ 0.50\\ 0.51\\ 0.58\\ 0.50\end{array}$	0.87 0.65 0.56 0.29 0.31 0.43 0.26 0.38 0.29 0.17 0.17 0.30 0.15 0.15 0.15 0.15 0.15 0.14 0.23 0.29 0.34 0.26 0.30 0.22 0.31	90.68 108.29 20.00 20.00 20.00 296.66 20.00 296.66 20.00 20.	18.26 17.32 12.58 3.65 4.69 1.35 1.92 1.05 0.81 0.41 0.45 0.86 0.61 0.48 0.56 0.35 0.66 0.72 0.76 0.68 0.69 0.49 0.70	106.37 117.43 80.28 41.59 45.79 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00	120.23 85.02 90.33 90.56 72.57 99.39 87.54 107.54 94.87 82.32 179.99 110.32 64.68 103.30 74.78 74.13 99.26 94.19 100.77 91.27 111.38 89.56 97.25	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 40.00 230.00 40.00 230.00 2379.00 221.00

Table 14. Selected elemental composition of tailings samples taken from borehole FS15-A, December 1993.

.

Sample	Depth (m)	AI %	As ug/g	Ca %	Cd ug/g	Co ug/g	Cr ug/g	Cu ug/g	Fe %	К %	Mg %	Mn %	Na %	Ni %	Pb ug/g	S %	Sb ug/g	Se ug/g	Te ug/g	Ti ug/g	Zn ug/g	Ci %	HPO4 ug/g	SO4 %
Trench 1 T 1-A T 1-B T 1-C T 1-D T 1-E T 1-F	0.05 0.50 1.00 1.50 2.00 2.50	4.894 4.741 5.167 5.261 4.825 4.457	582 558 571 595 525 500	2.081 2.452 2.854 2.930 2.585 2.493	<pre>< 3 < 3 < < 3 </pre>	59 195 177 162 104 129	330 262 390 299 228 324	263 861 845 970 433 660	14.340 15.420 14.130 13.390 12.320 15.760	0.427 0.416 0.504 0.483 0.426 0.400	2.927 3.020 3.277 3.394 3.118 2.936	0.104 0.128 0.141 0.136 0.128 0.120	0.136 0.130 0.164 0.163 0.109 0.117	0.142 0.475 0.445 0.377 0.238 0.326	< 25 < 25 < 25 < 25 < 25 < 25 < 25	1.759 2.084 1.156 0.848 0.994 2.015	178 199 165 200 151 227	< 50 < 50 < 50 < 50 < 50 < 50 < 50	19 19 21 21 20 13	< 25 < 25 < 25 < 25 < 25 < 25 < 25 < 25	157 229 204 199 162 180	4.75 4.59 4.84 3.75 4.99 4.77	< 40 < 40 < 40 < 40 < 40 < 40	6.68 6.32 3.30 3.33 2.96 3.24
Trench 2 T 2-A T 2-B T 2-C T 2-D T 2-E T 2-F T 2-G T 2-J T 2-H T 2-H T 2-H T 2-H T 2-K T 2-L	0.03 0.15 0.35 0.48 0.65 0.90 1.10 1.30 1.48 2.10 2.40 2.50	0.448 0.866 0.684 2.458 1.115 2.614 1.380 0.709 4.039 0.775 3.511 1.057	337 345 397 441 438 473 442 436 444 399 478 461	0.082 0.111 0.763 0.309 0.832 0.308 0.379 1.539 0.349 1.592 0.233	<pre>< 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3</pre>	34 38 188 83 197 191 201 203 213 216 129 184	33 52 53 138 55 135 86 35 251 44 184 56	172 131 511 464 558 1139 733 611 593 638 497 490	32,930 33,570 43,450 30,860 48,250 29,570 43,210 48,930 15,670 46,800 22,340 46,650	0.075 0.189 0.228 0.050 0.225 0.164 0.083 0.334 0.334 0.334 0.300 0.085	0.218 0.478 0.289 1.366 0.618 1.609 0.877 0.436 2.612 0.483 2.267 0.667	0.007 0.020 0.047 0.016 0.064 0.032 0.015 0.012 0.104 0.094 0.092	0.051 0.065 0.067 0.068 0.066 0.060 0.069 0.062 0.082 0.082 0.087 0.083	0.101 0.116 0.576 0.207 0.727 0.757 0.700 0.766 0.260 0.816 0.359 0.731	 < 25 	13.460 7.823 20.630 3.986 20.629 9.089 21.109 28.210 2.870 26.970 8.306 21.029	277 253 325 228 365 251 344 359 166 368 225 373	<pre>< 50 50 50 50 50 50 50 50 50 50 50 50 50 5</pre>	31 27 41 25 35 25 38 34 < 10 43 11 46	 < 25 	79 186 139 99 138 204 133 122 164 122 147 113	4.14 3.91 4.86 5.26 4.80 4.51 5.07 4.44 4.65 4.84 4.40 4.67	< 40 < 40 < 40 < 40 < 40 < 40 < 40 < 40	4.22 13.40 6.79 17.90 6.68 6.71 4.12 5.16 5.52 4.34 6.84 4.24
Trench 3 T 3-A T 3-B T 3-C T 3-C T 3-D T 3-E T 3-F T 3-G T 3-H	0.03 0.15 0.40 0.63 0.90 1.15 1.48 2.10	3.830 1.127 1.346 2.022 3.209 0.612 1.891 4.381	435 409 434 422 445 460 409 541	0.559 0.386 0.571 0.897 1.083 0.165 0.668 2.002	 3 3 3 3 4 3 4 3 4 3 	47 357 184 179 140 226 160 156	260 123 70 94 158 42 126 212	150 4729 512 531 439 649 630 760	14.650 38.430 40.900 31.900 20.620 47.890 32.050 13.520	0.331 0.051 0.134 0.196 0.308 0.066 0.173 0.374	2.385 0.434 0.750 1.249 1.987 0.381 1.248 2.818	0.089 0.011 0.023 0.050 0.081 0.012 0.048 0.114	0.091 0.056 0.069 0.077 0.074 0.057 0.065 0.098	0.068 0.618 0.654 0.579 0.424 0.847 0.557 0.361	< 25 < 25 < 25 < 25 < 25 < 25 < 25 < 25	0.740 9.385 14.726 11.595 5.917 21.637 14.948 1.088	166 281 300 247 190 336 256 135	< 50 < 50 < 50 < 50 < 50 < 50 < 50 < 50	< 10 23 23 23 < 10 48 25 < 10	 25 	120 121 127 145 151 132 161 180	4.79 4.97 4.31 4.60 4.45 4.70 4.18 4.65	< 40 < 40 < 40 < 40 < 40 < 40 < 40 < 40	4.12 8.63 12.00 7.03 6.57 4.94 6.77 5.53

Table 15. Selected elemental composition of tailings samples taken from trenches, May 1994.

Sample	Depth (m)	S(total) %	SO4 %	SO4 as S (SO4)/3 %	Sulphide as S [S(total)]-[SO4 as S] %	Acid Potential (Sulphide as S)x(31.25) (kg CaCO3/tonne)
Trench 1 T 1-A T 1-B T 1-C T 1-C T 1-D T 1-E T 1-F	0.05 0.50 1.00 1.50 2.00 2.50	1.759 2.084 1.156 0.848 0.994 2.015	6.680 6.320 3.300 3.330 2.960 3.240	2.23 2.11 1.10 1.11 0.99 1.08	-0.47 -0.02 0.06 -0.26 0.01 0.93	0 0 2 0 0 29
Trench 2 T 2-A T 2-B T 2-C T 2-D T 2-E T 2-F T 2-G T 2-J T 2-H T 2-H T 2-H T 2-K T 2-L	0.03 0.15 0.35 0.48 0.65 0.90 1.10 1.30 1.48 2.10 2.40 2.50	13.460 7.823 20.630 3.986 20.629 9.089 21.109 26.970 28.210 2.870 8.306 21.029	4.220 13.400 6.790 17.900 6.680 6.710 4.120 4.340 5.160 5.520 6.840 4.240	1.41 4.47 2.26 5.97 2.23 2.24 1.37 1.45 1.72 1.84 2.28 1.41	12.05 3.36 18.37 -1.98 18.40 6.85 19.74 25.52 26.49 1.03 6.03 19.62	377 105 574 0 575 214 617 798 828 32 188 613
Trench 3 T 3-A T 3-B T 3-C T 3-C T 3-D T 3-E T 3-F T 3-G T 3-H	0.03 0.15 0.40 0.63 0.90 1.15 1.48 2.10	0.740 9.385 14.726 11.595 5.917 21.637 14.948 1.088	4.120 8.630 12.000 7.030 6.570 4.940 6.770 5.530	1.37 2.88 4.00 2.34 2.19 1.65 2.26 1.84	-0.63 6.51 10.73 9.25 3.73 19.99 12.69 -0.76	0 203 335 289 116 625 397 0

 Table 16. Sulphate and sulphide composition in tailings

 and stoichiometric estimate of acid potential.

Table II.	major meu		IOIT CONCEI	iuauons a	enacteu	poremater	Hom Dore		-A tanings a	ampies, L	ecember	1993.						
Sample	Depth (m)	Depth (m)	Ag (mg/L)	Al (mg/L)	As (mg/L)	B (mg/L)	Ba (mg/L)	Be (mg/L)	Ca (mg/L)	Cd (mg/L)	CI (mg/L)	Co (mg/L)	Cr (mg/L)	Cu (mg/L)	Fe Total (mg/L)	Fe+3 (mg/L)	K (mg/L)	Li (mg/L)
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 20 21 22 23	$\begin{array}{c} 0.40{-}0.60\\ 0.95{-}1.25\\ 3.80{-}4.00\\ 5.50{-}5.80\\ 7.15{-}7.45\\ 8.75{-}9.15\\ 10.50{-}10.95\\ 12.05{-}12.43\\ 13.57{-}13.97\\ 15.09{-}15.49\\ 16.61{-}16.91\\ 18.08{-}18.43\\ 20.21{-}20.61\\ 21.74{-}22.14\\ 23.17{-}23.47\\ 24.70{-}25.00\\ 26.21{-}26.51\\ 27.79{-}28.19\\ 29.31{-}29.66\\ 30.88{-}31.28\\ 33.05{-}33.40\\ 33.84{-}34.14\\ 36.18{-}36.58\\ \end{array}$	0.50 1.10 3.90 5.70 7.30 8.95 10.72 12.24 13.77 15.29 16.76 18.25 20.41 21.94 23.32 24.35 26.36 27.99 29.49 31.08 33.23 34.00 36.38	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	235.89 6.49 0.52 0.29 0.26 0.31 < 0.25 0.26 0.34 2.01 0.44 0.51 0.38 0.69 0.59 < 0.25 0.39 10.73 0.45 0.49 < 0.25	6.97 5.29 0.25 0.25 </td <td>4.43 60.72 0.30 0.48 0.38 0.36 0.41 0.28 0.34 0.33 0.47 0.44 0.33 0.47 0.44 0.34 0.34 0.37 0.35 < 0.25 < 0.25 < 0.44 0.32 0.32 0.32 0.32</td> <td> 0.05 </td> <td>$\begin{array}{c ccccccccccccccccccccccccccccccccccc$</td> <td>423.26 441.08 1380.00 550.41 596.93 523.38 496.26 525.54 559.95 493.35 1940.00 552.68 577.35 432.92 480.85 348.97 355.54 456.33 279.36 239.54 313.38 284.61 393.19</td> <td>0.88 0.02</td> <td>30.80 21.10 7.13 8.74 29.30 15.00 17.80 61.90 53.60 50.50 69.40 66.60 64.60 57.50 58.70 56.40 44.80 60.60 46.60 51.70</td> <td>75.46 2.16 0.31 0.02 0.03</td> <td>0.17 0.02 <td> 0.025 0.026 0.025 0.026 0.025 0.025 0.025 0.026 0.027 0.025 0.025 0.025 0.025 </td><td>8008.00 12500.00 20.19 16.12 0.23 1.82 10.00 2.13 1.47 5.78 66.25 3.07 2.64 0.07 0.57 0.57 0.30 0.46 0.82 0.17 23.99 1.75 1.14 0.38</td><td>$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$</td><td>52.78 290.97 209.82 148.88 170.02 175.75 182.35 180.75 178.58 146.73 177.23 181.66 144.41 153.61 144.59 125.14 137.07 120.98 100.32 115.57 103.08 119.38</td><td>0.84 < 0.25 < 0.25</td></td>	4.43 60.72 0.30 0.48 0.38 0.36 0.41 0.28 0.34 0.33 0.47 0.44 0.33 0.47 0.44 0.34 0.34 0.37 0.35 < 0.25 < 0.25 < 0.44 0.32 0.32 0.32 0.32	 0.05 	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	423.26 441.08 1380.00 550.41 596.93 523.38 496.26 525.54 559.95 493.35 1940.00 552.68 577.35 432.92 480.85 348.97 355.54 456.33 279.36 239.54 313.38 284.61 393.19	0.88 0.02	30.80 21.10 7.13 8.74 29.30 15.00 17.80 61.90 53.60 50.50 69.40 66.60 64.60 57.50 58.70 56.40 44.80 60.60 46.60 51.70	75.46 2.16 0.31 0.02 0.03	0.17 0.02 <td> 0.025 0.026 0.025 0.026 0.025 0.025 0.025 0.026 0.027 0.025 0.025 0.025 0.025 </td> <td>8008.00 12500.00 20.19 16.12 0.23 1.82 10.00 2.13 1.47 5.78 66.25 3.07 2.64 0.07 0.57 0.57 0.30 0.46 0.82 0.17 23.99 1.75 1.14 0.38</td> <td>$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$</td> <td>52.78 290.97 209.82 148.88 170.02 175.75 182.35 180.75 178.58 146.73 177.23 181.66 144.41 153.61 144.59 125.14 137.07 120.98 100.32 115.57 103.08 119.38</td> <td>0.84 < 0.25 < 0.25</td>	 0.025 0.026 0.025 0.026 0.025 0.025 0.025 0.026 0.027 0.025 0.025 0.025 0.025 	8008.00 12500.00 20.19 16.12 0.23 1.82 10.00 2.13 1.47 5.78 66.25 3.07 2.64 0.07 0.57 0.57 0.30 0.46 0.82 0.17 23.99 1.75 1.14 0.38	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	52.78 290.97 209.82 148.88 170.02 175.75 182.35 180.75 178.58 146.73 177.23 181.66 144.41 153.61 144.59 125.14 137.07 120.98 100.32 115.57 103.08 119.38	0.84 < 0.25 < 0.25
Sample	Depth (m)	Depth (m)	Mg (mg/L)	Mn (mg/L)	Mo (mg/L)	Na (mg/L)	Ni (mg/L)	Pb (mg/L)	S (mg/L)	Sb (mg/L)	Se (mg/L)	Si (mg/L)	Sn (mg/L)	Sr (mg/L)	Te (mg/L)	Ti (mg/L)	TI (mg/L)	Zn (mg/L)
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23	0.40-0.60 0.95-1.25 3.80-4.00 5.50-5.80 7.15-7.45 8.75-9.15 10.50-10.95 12.05-12.43 13.57-13.97 15.09-15.49 16.61-16.91 18.08-18.43 20.21-20.61 21.74-22.14 23.17-23.47 24.70-25.00 26.21-26.51 27.79-28.19 29.31-29.66 30.88-31.28 33.05-33.40 33.84-34.14 36.18-36.58	0.50 1.10 3.90 5.70 7.30 8.95 10.72 12.24 13.77 15.29 16.76 18.25 20.41 21.94 23.32 24.35 26.36 27.99 29.49 31.08 33.23 34.00 36.38	388.04 794.77 521.35 457.52 313.74 409.77 443.10 174.89 146.55 136.88 102.38 149.26 204.27 205.08 246.18 208.97 210.32 194.22 119.29 111.28 137.85 131.50 125.63	$\begin{array}{c} 19.21\\ 101.94\\ 4.03\\ 0.78\\ 0.57\\ 0.68\\ 1.15\\ 0.90\\ 0.55\\ 0.48\\ 10.64\\ 0.53\\ 0.82\\ 0.35\\ 0.47\\ 0.31\\ 0.30\\ 0.39\\ 0.25\\ 0.48\\ 0.46\\ 0.27\\ 0.33\\ \end{array}$	 < 0.25 	12.16 13.61 11.45 14.25 19.51 23.94 34.82 66.22 90.68 114.63 111.18 140.54 173.11 302.93 342.91 280.86 302.62 309.21 194.17 160.27 177.11 171.92 148.38	3200.00 200.73 84.23 0.93 2.48 2.91 0.90 1.84 2.25 2.14 6.70 1.22 1.07 3.09 2.58 1.88 2.96 0.76 2.93 2.26 0.59 3.18 3.00	 < 0.25 	$\begin{array}{c} 12200.00\\ 9280.00\\ 5710.00\\ 1100.00\\ 1240.00\\ 1020.00\\ 1070.00\\ 761.83\\ 768.18\\ 688.88\\ 620.82\\ 769.46\\ 863.41\\ 935.15\\ 1020.00\\ 810.85\\ 850.55\\ 924.85\\ 530.55\\ 924.85\\ 530.55\\ 467.22\\ 568.05\\ 530.66\\ 776.31\\ \end{array}$	0.35 9.25 0.84 0.81 0.49 0.83 0.74 0.44 0.56 0.54 0.37 0.37 0.42 0.59 0.52 0.42 < 0.25	 < 0.25 < 0.25	34.18 19.27 0.84 6.25 2.22 5.50 6.28 6.18 5.31 6.37 7.71 5.95 6.53 4.77 4.50 3.67 2.83 5.23 3.36 14.67 6.05 4.48 4.22	 0.25 3.27 0.25 	 < 2.50 	1.10 1.88 0.25	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	 < 0.25 	19.766 4.648 0.254 0.025

Table 17. Major metal and ion concentrations in exracted porewater from borehole FS15-A tailings samples, December 1993.

Table 18. Extracted porewater physico-chemical parametervalues for borehole FS15-A tailings samples, December 1993

Table '	19. Majo	or metal	and ior	n concentra	itions	in ex	tracted	porewate	r from	trenci	n taili	ngs	sample	s, May	1994.

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Sample	Sample interval (m)	Al mg/L	As mg/L	Ca mg/L	Cd mg/L	Co mg/L	Cr mg/L	Cu mg/L	Fe mg/L	K mg/L	Mg mg/L	Mn mg/L	Na mg/L	Ni mg/L	Pb mg/L	S mg/L	Sb mg/L	Se mg/L	Si mg/L	Te mg/L	TI mg/L	Zn mg/L	Fe+3 mg/L	CI mg/L	SO4 mg/L
Trench 1 T 1-A T 1-B T 1-C T 1-D T 1-E T 1-F	0.00-0.10 0.45-0.55 0.95-1.05 1.45-1.55 1.95-2.05 2.45-2.55	5.43 1.89 0.29 0.29 < 0.25 0.27	 < 0.25 1.06 < 0.25 < 0.25 < 0.25 < 0.25 	564.04 464.12 562.88 500.24 577.88 691.00	< 0.03 < 0.03 < 0.03 < 0.03 < 0.03 < 0.03 < 0.03	2.32 15.51 0.09 < 0.03 0.11 < 0.03	< 0.03 < 0.03 < 0.03 < 0.03 < 0.03 < 0.03 < 0.03	1.34 < 0.03 < 0.03 < 0.03 < 0.03 < 0.03	17.03 1421.00 2.37 0.32 15.49 0.37	50.53 234.23 166.64 240.81 223.04 183.20	27.70 951.08 533.09 845.53 1022.75 1311.09	7.06 240.46 2.21 0.82 2.32 0.72	8.00 16.81 11.63 15.74 13.34 23.20	57.15 730.95 22.26 4.51 19.63 2.17	0.43 0.64 < 0.25 < 0.25 < 0.25 < 0.25	539.84 3042.00 1140.39 1513.94 1907.02 2337.96	< 0.25 1.04 0.96 1.43 1.86 2.13	< 0.50 < 0.50 < 0.50 < 0.50 < 0.50 < 0.50 < 0.50	30.85 14.81 6.65 7.22 2.09 4.79	0.16 0.20 0.10 0.13 0.11 < 0.10	< 0.25 0.30 < 0.25 < 0.25 < 0.25 < 0.25 < 0.25	1.02 5.77 0.12 0.03 0.09 < 0.03	5.50 18.30 0.08 < 0.05 0.24 < 0.05	4.33 12.30 10.60 10.20 13.40 19.30	1530.00 10100.00 3500.00 4900.00 5890.00 5190.00
Trench 2 T 2-A T 2-B T 2-C T 2-D T 2-C T 2-D T 2-F T 2-G T 2-J T 2-H T 2-H T 2-H T 2-H T 2-H T 2-L	0.00-0.05 0.05-0.25 0.25-0.45 0.50-0.80 0.50-0.80 1.00-1.20 1.25-1.45 1.20-1.75 2.10 2.40 2.50		 5.35 13.94 6.51 8.29 4.84 2.58 3.73 3.73 3.64 6.41	272.72 466.25 449.86 443.14 448.14 426.04 494.77 553.60 444.98 439.72		 1.82 35.23 3.70 8.94 2.76 2.29 8.51 5.01 6.14	2.02 			 8.41 27.56 8.54 216.44 151.15 149.99 191.63 203.97 284.42 207.48		 6.82 12.61 9.45 140.99 93.50 71.47 135.40 120.18 151.73 78.63	9.07 	 947.09 574.88 3612.00 293.10 281.26 174.52 757.09 367.02 663.79	 - 0.25 - -<td>4379.00 7340.00 4566.00 7768.00 3259.00 2423.00 6958.00 5534.00 7307.00</td><td>2.94 3.79 3.83 0.62 4.64 2.52 1.46 2.72 4.52 4.57</td><td> < 0.50 < 0.50 <</td><td></td><td>0.39 0.77 0.70 0.78 0.61 0.35 0.11 0.46 0.46 0.73</td><td>< 0.25 - 0.81 0.52 0.68 0.59 0.34 < 0.25 0.51 0.44 0.48</td><td> 2.29 12.13 5.43 23.09 11.41 4.80 1.26 4.92 5.15 6.29</td><td>3040.00 </td><td> 2.33 < 0.50 < 0.50 6.16 5.42 14.10 7.96 10.30 < 0.50</td><td>11300.00 16300.00 12300.00 17100.00 10800.00 9000.00 5790.00 10900.00 13400.00 14200.00</td>	4379.00 7340.00 4566.00 7768.00 3259.00 2423.00 6958.00 5534.00 7307.00	2.94 3.79 3.83 0.62 4.64 2.52 1.46 2.72 4.52 4.57	 < 0.50 <		0.39 0.77 0.70 0.78 0.61 0.35 0.11 0.46 0.46 0.73	< 0.25 - 0.81 0.52 0.68 0.59 0.34 < 0.25 0.51 0.44 0.48	 2.29 12.13 5.43 23.09 11.41 4.80 1.26 4.92 5.15 6.29	3040.00 	 2.33 < 0.50 < 0.50 6.16 5.42 14.10 7.96 10.30 < 0.50	11300.00 16300.00 12300.00 17100.00 10800.00 9000.00 5790.00 10900.00 13400.00 14200.00
Trench 3 T 3-A T 3-B T 3-C T 3-D T 3-E T 3-F T 3-G T 3-H	0.00-0.05 0.05-0.25 0.30-0.50 0.55-0.70 0.80-1.00 1.00-1.30 1.35-1.60 2.00-2.20	 60.10 7.70 9.70 5.34 < 0.25	 13.68 12.81 11.87 6.38 < 0.25	 443.93 440.48 457.21 461.37 704.89	 0.49 0.11 0.11 1.90 < 0.03	 38.82 10.80 7.50 7.42 0.06	 -	 - 0.03 < 0.03 < 0.03 < 0.03 - 0.05	 14530.00 15340.00 14000.00 5926.00 42.80	122.62 320.23 312.66 259.61 240.31	 556.87 1089.96 1488.44 760.48 	 52.50 213.55 213.16 262.93 1.99	8.86 12.86 11.96 10.88 	 2832.00 2329.00 1697.00 2679.00 	- 2.01 1.27 0.95 2.08 - - < 0.25	 13590.00 15370.00 12490.00 6992.00 2316.31	5.71 7.64 8.40 0.62	 	 16.29 18.77 23.15 18.32 2.63	 1.41 1.31 1.13 0.54 < 0.10	 1.39 1.71 1.38 1.04 < 0.25	 15.45 11.39 15.31 20.06 0.09	 294.00 355.00 304.00 87.10 0.20	 52.50 86.50 158.00 33.50 14.00	

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Sample	Depth (m)	рН	REDOX Potential (mV)	Electrical Conductivity (mS/cm)
Trench 1 T 1-A T 1-B T 1-C T 1-D T 1-E T 1-F	0.05 0.50 1.00 1.50 2.00 2.50	4.40 4.68 6.60 6.67 6.23 6.84	301.00 177.00 60.00 165.00 58.00 121.00	2.24 7.78 3.90 5.37 6.16 7.25
Trench 2 T 2-A T 2-B T 2-C T 2-D T 2-E T 2-F T 2-G T 2-J T 2-H T 2-H T 2-H T 2-K T 2-K T 2-L	0.03 0.15 0.35 0.48 0.65 0.90 1.10 1.30 1.48 2.10 2.40 2.50	* 2.13 * 2.75 4.17 3.73 4.35 4.35 4.52 4.32 4.78 4.16 4.19	* 522.00 * 380.00 191.00 222.00 176.00 167.00 169.00 205.00 185.00	* 12.44 * 14.52 11.09 16.86 11.79 11.35 7.93 6.56 11.72 13.01
Trench 3 T 3-A T 3-B T 3-C T 3-C T 3-D T 3-E T 3-F T 3-G T 3-H	0.03 0.15 0.40 0.63 0.90 1.15 1.48 2.10	* 4.05 4.10 4.07 4.12 * 6.12	* 207.00 206.00 198.00 214.00 * -72.00	* 18.98 20.20 17.75 12.13 * 5.72

Table 20. Extracted porewater physico-chemicalparameter values for trench tailings samples, May 1994

* No porewater extractable or sample was hardpan

Sample	nple B.C. Research Initial Test of neu				on potential (NP)	Stoich of acid	iometric estimate I potential (AP)	Acid-base accour	nting
Number	Depth (m)	Paste pH	1N H2SO4 Counsumed (ml/10g)	Acid-consuming Ability (kg H2SO4/tonne)	NP (kg CaCO3/tonne)	Total S (%)	AP (kg CaCO3/tonne)	NNP (NP-AP) (kg CaCO3/tonne)	NP/AP
1	0.50	3.3	0.0	0	0	18.26	571	-571	0.00
2	1.10	3.3	0.0	0	· 0	17.32	541	-541	0.00
3	3.90	5.5	3.6	18	18	12.58	393	-375	0.05
4	5.70	7.2	11.2	55	56	3.65	114	-58	0.49
5	7.30	6.2	7.6	37	38	4.69	147	-109	0.26
6	8.95	7.3	11.3	55	57	1.35	42	14	1.34
7	10.72	7.5	11.9	58	60	1.92	60	0	0.99
8	12.24	7.2	11.0	54	55	1.05	33	22	1.68
9	13.77	7.4	11.6	57	58	0.81	25	33	2.31
10	15.29	7.3	11.6	57	58	0.41	13	45	4.49
11	16.76	7.7	11.4	56	57	0.45	14	43	4.07
12	18.25	7.4	13.0	64	65	0.86	27	38	2.42
13	20.41	7.6	12.7	62	64	0.61	19	44	3.32
14	21.94	7.6	13.0	64	65	0.48	15	50	4.34
15	23.32	7.6	12.0	59	60	.0.56	17	43	3.45
16	24.35	7.8	12.5	61	63	0.35	11	51	5.68
17	26.36	7.5	16.4	80	82	0.66	21	61	3.99
18	27.99	7.2	12.7	62	64	0.72	23	41	2.81
19	29.49	7.3	10.0	49	50	0.76	24	26	2.11
20	31.08	7.4	11.3	55	57	0.68	21	35	2.68
21	33.23	7.5	11.6	57	58	0.69	22	36	2.68
22	34.00	7.4	12.5	61	63	0.49	15	· 47	4.07
23	36.38	7.4	11.5	56	58	0.70	22	36	2.62
Average					52	~~~~	95	-43	0.55

Table 21. Acid-base accounting of samples recovered from borehole FS15-A.

(1) Acid potential was calculated by multiplying the total S% by 31.25 which assums 1 mole pyrrhotite is neutralized by 1 mole CaCO3.
 (2) Neutralization potential is converted from acid-consuming ability by assuming 1 mole H2SO4 is neutralized by 1 mole CaCO3.

able 22. MINTEQA2 saturation indicies of secondary minerals occuring in sulphide rich tailings.							
Sample	Depth (m)	Ferrihydrite Fe(OH)3	Goethite FeOOH	Hematite Fe2O3	Jarosite KFe3(SO4)2(OH)6	Melanterite FeSO4 7H2O	Gypsum CaSO4 2H2O
Borehole ES15-1							
1	0.50	1 29	5 32	15 60	15.26	-0.62	0.53
2	1 10	1.20	5.74	16.00	16.03	-0.02	0.55
2	3 90	1.12 Eo2+ nd	0.74 Eo2+ nd	10.45 Fo24 ad	10.93	-0.52	0.47
4	5.30	Fe3+ nu Fe3+ nu	Fe3+ nd	Fe3+ nd	Fe3+ nd	-3.27	0.96
5	7 30	Fe3+ nd	Fe3t na	Fe3+ nd	Fe3+ nd	-3.00	0.30
5	7.30 9.05	Fe3+ no	Fe3+ nd	Fe3+nd	Fe3+ nd	-5.45	0.39
7	0.95	Fe3+nd	Fe3+nd	Fe3+ nd	Fe3+ nd	-4.61	0.27
<i>'</i>	10.7	Fe3+nd	Fe3+nd	Fe3+ nd	Fe3+ nd	-3.86	0.26
0	12.2	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-4.56	0.25
9	13.8	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-4.72	0.28
10	15.3	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-4.14	0.21
11	16.8	4.09	8.11	21.18	14.04	-3.93	0.54
12	18.3	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-4.40	0.28
13	20.4	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-4.46	0.30
14	21.9	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-6.00	0.22
15	23.3	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-5.09	0.26
16	24.4	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-5.39	0.10
17	26.4	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ nd	-5.20	0.12
18	28.0	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+nd	-4.94	0.23
19	29.5	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+nd	-5 70	-0.05
20	31.1	3.53	7 55	20.06	9.52	-3.60	-1 40
21	33.2	Fe3+ nd	Fe3+ nd	Ee3+ nd	Ee3+ nd	-4.68	0.00
22	34.0	Fe3+ nd	Fo3+ nd	Fo3+ nd	Fe3+ nd	-4.00	0.00
23	36.4	Fo2+ nd	Fe3tind	Fe3t nu	Fe3+ nd	-4.07	-0.05
20	50.4	Fe3+ IId	rest nu	rest nu	rest na	-5.27	0.10
Trench 1		-					
T 1-A	0.05	1.04	5.06	15.08	11.34	-3.76	-1.65
T 1-B	0.50	1.79	5.81	16.59	14.13	-1.58	0.36
T 1-C	1.00	1.41	5.44	15.83	6 50	-4 50	0.31
T 1-D	1.50	Fe3+ nd	Fe3+ nd	Fe3+ nd	Fe3+ od	-5.31	0.30
T 1-E	2 00	1.52	5 54	16.04	8.33	-3.61	0.38
T 1-F	2.50	Fe3+ nd	Fe3+ nd	Fe3+ nd	Ee3+ nd	-5.31	0.39
	2.00	1001114	100.110	1001110	1001114	0.01	0.00
Trench 2							
12-A	0.03					、	
T 2-B	0.15	-1.90	2.12	9.21	8.80	-2.10	-0.01
T 2-C	0.35			·			
T 2-D	0.48	-0.57	3.46	11.88	11.73	-1.06	0.28
T 2-E	0.65	1.88	5.90	16.76	14.48	-0.95	0.33
T 2-F	0.90	1.11	5.13	15.23	14.95	-0.89	0.34
T 2-G	1.10	2.13	6.16	17.27	15.82	-1.01	0.29
T 2-J	1.30	1.53	5.55	16.06	13.56	-1.28	0.30
T 2-H	1.48	0.53	4.56	14.07	10.98	-2.35	0.29
T 2-I	2.10	2.13	6.16	17.28	14.71	-1.16	0.40
T 2-K	2.40	1.54	5.57	16.10	15.10	-1.05	0.35
T 2-L	2.50	1.81	5.84	16.64	15.66	-0.87	0.33
Trench 3	0.00						
1 3-A	0.03						
T 3-B	0.15						
T 3-C	0.40	1.81	5.84	16.64	15.69	-0.63	0.30
T 3-D	0.63	1.95	5.98	16.91	16.28	-0.65	0.27
T 3-E	0.90	1.86	5.88	16.73	16.14	-0.67	0.30
T 3-F	1.15	1.57	5.59	16.15	15.36	-0.94	0.38
T 3-G	1.48						
T 3-H	2.10	1.34	5.36	15.68	7.95	-3.21	0.43

Fe3+ np = Fe3+ ion not detected in porewater

concentration measurements.					
Depth (m)	Dec 1993 (%)	May 1993 (%)			
Station	1, near FS1	5			
0.20	0.40	6.6			
0.25	0.40	5.8			
0.55	0.45	1.05			
1.05	0.40	0.9			
Station 2, near FS5					
0.23	0.65	3.8			
0.30	0.35	1.2			
0.65	0.35	0.8			
1.10	0.35	0.75			

Table 22 D



Figure 1. Site location plan.



Figure 2. Aerial photograph of site in 1946, prior to tailings disposal.



Figure 3. Site plan and topography.

GEOMAP. DRW



Figure 5. Monitoring station location plan.

Figure 6. Trench location plan.

Figure 7. Measurement of pore-gas oxygen concentration..






Figure 9. Cross sectional location plan and water level elevations.



Figure 10. Regional water level elevations along section A-A'.





Figure 11. Water level elevations in the shallowest piezometers along section B-B'.



Figure 12. Equivalent FeS composition, porewater chemistry, acid and neutralization potential, and paste pH distribution by depth at borehole FS15-A.



Figure 13. Iron and sulphur composition of tailings samples taken from the three trenches and borehole FS15-A.



Figure 14. Porewater chemistry for tailings samples taken from the three trenches and borehole FS15-A.

APPENDIX A

Groundwater monitoring station data

PROJECT: Fault Lake Tailings, Falconbridge BOREHOLE ID: F\$15-A PIEZO. INSIDE RADIUS (r): * COORDINATES: N 5161600.09 m GROUND ELEVATION: 321.00 m PIEZO. TIP RADIUS (R): E 515650.20 m CORE SIZE: 0.15 m PIEZO. TIP LENGTH (I): H E CORE DESCRIPTION BOREHOLE FILL DESCRIPTION HYDR CONDUCTION BOREHOLE FILL DESCRIPTION HYDR O Bentonite Seal Borehole Seal	12 mm 8 mm							
HYDR CORE DESCRIPTION BOREHOLE FILL DESCRIPTION HYDR CONDU (cn) Bentonite Seal	Image: BoreHole ID:FS15-APIEZO. INSIDE RADIUS (r):12 mmGROUND ELEVATION:321.00 mPIEZO. TIP RADIUS (R):8 mmCORE SIZE:0.15 mPIEZO. TIP LENGTH (I):26 cm							
Bentonite Seal	AULIC CTIVITY n/s)							
- 5 Sand Backfill								
- 10 Bentonite Seal								
Fine Grey Tailings								
Tailings Closed In (No Backfill)								
■ Bentonite Seal ■ Sand Backfill Piezometer Tip Location 3.6e-	-06							
 40 Bedrock estimated from piezocone testing (UBC)								

NORANDA TECHNOLOGY CENTRE - BOREHOLE AND PIEZOMETER INSTALLATION LOG PROJECT: Fault Lake Tailings, Falconbridge BOREHOLE ID: FS15-B PIEZO. INSIDE RADIUS (r): 12 mm COORDINATES: N 5161604.97 m GROUND ELEVATION: 321.00 m PIEZO. TIP RADIUS (R): 8 mm CORE SIZE: 0.15 m PIEZO. TIP LENGTH (I): 26 cm E 515649.59 m WATER LEVELS HYDRAULIC DEPTH (m) BOREHOLE FILL DESCRIPTION CORE DESCRIPTION CONDUCTIVITY (cm/s) 0 **Bentonite Seal** 5 Sand Backfill 10 **Fine Grey Tailings Bentonite Seal** 15 Tailings Closed In (No Backfill) 20 **Bentonite Seal** Sand Backfill 1.2e-05 **Piezometer Tip Location** 25 30 35 - - - Bedrock estimated from piezocone testing (UBC)- - - - - -40

Table A1. Location of groundwater monitoring stations.

Station	Northing (m)	Easting (m)
FS1 FS2 FS3 FS4 FS5 FS6 FS7 FS8 FS9 FS10 FS11 FS12 FS13 FS14 FS15	5161204.9 5160988.2 5161495.7 5161653.3 5161677.0 5161823.3 5161914.8 5162074.8 5162074.8 5162204.6 5161943.7 5161797.4 5161546.0 5161378.6 5161526.2 5161280.1	516163.5 515777.6 515900.7 515495.0 515577.9 515659.0 515885.8 515886.4 515937.6 515941.2 515748.6 515727.9 515952.5 516041.2 515649.9

ezometer	Elevatio	ns								
	Ground	Top of Piezometer	Tip of Piezometer	Depth of tip below ground	Nov 24, 92 Water	Mar 26, 93 Water	Dec 8, 93 Water	May 21, 94 Water	Jul 16, 94 Water	Sep 3, 9 Water
	(m)	<u>(m)</u>	(m)	(m)	(m)	(m)	(m)	(m)	(m)	(m)
FS1-A	304,468	305,270	287.620	16.848	298.360	298.353	298.470	298,440	298,320	298.47
FS1-B		305.237	292.047	12.422	298.652	298.603	298.447	298.420	298.337	298.44
FS1-C		305.240	294.920	9.549	298.695	298.673	298.450	298.420	298.280	dry
FS2-A	316.112	316.843	300.593	15.518	307.233	307.208	306.873	307.063	306.953	307.04
FS2-B		316.843	305.823	10.288	307.248	307.318	306.873	307.083	306.973	307.04
FS3-A	322.117	322.988	262.988	60.000	299.528	299.525				314.16
FS3-B		322.918	271.903	50.213	299.888	299.630		307.358	299.068	299.04
FS3-C		322.943	302.920	19.197	dry	dry			301.023	299.09
FS4-A	322.635	323.357	291.017	31.618	299.507	291.910		299.097	298.927	299.15
FS4-B		323.394	294.004	28.631	299.454	299.460		299.174	299.004	299.17
FS4-C		323.464	298.654	23.981	299.584	299.510	~~	299.249	299.064	299.08
FS5-A	321.934	322.802	283.522	38.411	299.582	299.582	,	299.322	299.082	299.15
FS5-B		322.741	286.541	35.392	301.766	301.766		299.666	299.101	
FS5-C		322.845	295.645	26.289	299.565	299.565		299.205	299.145	
FS6-A	320.897	321.711	287.541	33.356	300.301	300.301		299.211	299.066	
FS6-B	1	321.419	298.949	21.949	299.519	299.519		dry	dry	dry
FS6-C		321.440	311.720	9.177	311.740	311.740		dry	dry	dry
FS7-A	309.894	310.653	301.703	8.191	dry	dry	dry	dry	dry	dry
FS8-A	302.762	303.508	298.408	4:353	299.288	·	299.098	298,998	298.968	298.60
FS9-A	302.579	303.325	291.025	11.553	299.325		299.125	298.815	298.975	298.93
FS9-B		303.414	292.214	10.365	299.264		299.074	298.969	298.964	298.92
FS9-C		303.469	298.669	3.910	299.369		299.169	299.049	dry	dry
FS10-A	304.194	304.953	294.133	10.061	299.313		299.133	299.053	299.018	299.00
FS10-B		304.959	296.699	7.495	299.349		299.159	299.079	299.039	299.01
FS11-A	320.379	320.602	299.172	21.207	299.352	299.309			dry	dry
FS12-A	320.044	320.044	298.894	21.150	299.054	299.124			dry	300.24
FS13-A	319.282	319.946	296.396	22.886	299.446	299.460			317.251	317.29
FS14-A	322.970	323,193	292.843	30,127	299.243	296.820		298.403	298,963	299.59
FS14-B		324.015	297.365	25.605	299.460	299.280		299,165	299.105	299.14
FS15-A	321.000	321.826	286.826	35.000				306.136	307.096	312.92
FS15-B		321.865	297.865	24,000		·		302.265	306.505	318.96















Groundwater Monitoring Stations FS7 and FS8







Groundwater Monitoring Stations FS11, FS12 and FS13





APPENDIX B

Surface water monitoring station data

Table I	B1. Location	and water	r level elevation	ns of kettle la	kes.		
Lake	Location of UTM Co-orc Northing (m)	gauge linates (m) Easting (m)	Elevation Zero reading on Staff Plate (m)	May 27, 94 Water Level (m)	Jul 16, 94 Water Level (m)	Sep 3, 94 Water Level (m)	May 1984 Topo map Water Level (m)
A B 1 2 3 4	5159502.9 5160048.8 5160519.0 5162039.0 5161866.2 5162328.9	515068.4 515126.3 515396.0 515973.3 516025.1 516035.4	307.126 307.220 307.254 298.545 298.298 298.343	307.736 307.760 307.789 299.165 298.924 298.931	307.690 307.763 307.690 299.165 298.903 298.917	308.386 308.350 308.369 299.745 299.514 299.471	307.540 307.240 307.240 297.480 297.480 297.180

Kettle Lake Monitoring Stations



APPENDIX C

Laboratory Certificates of Analysis

Element	mg/L	Element	mg/L
Ag	0.01	Mn	0.001
AI	0.05	Мо	0.05
As	0.05	Na	0.05
В	0.05	Ni	0.005
Ва	0.01	Pb	0.05
Be	0.01	S	0.05
Са	0.01	Sb	0.05
Cd	0.005	Se	0.05
Co	0.005	Si	0.01
Cr	0.005	Sn	0.05
Cu	0.005	Sr	0.05
Fe	0.005	Те	0.05
к	1	TI	0.05
Li	0.05	Ti	0.1
Mg	0.1	Zn	0.005

Table C1. ICP detection limits in clean aqueous solutions.

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To :	P.Tib	ble M.	.Woyshn	er			PROJET	/ PROJI	ECT: V2-1	T03					Date: 1,	/11/94	
Lab #	I.D.	Description		A	ι		A	S .	Ca	a		C	d		Cı	J	
327	93226	Water						÷									
328	93227	Water		.31	mg/L	<	.25	mg/L	83.79	mg/L	<	.02	mg/L	<	.02	mg/L	
329	93228	Water															
330	93229	Water		.30	mg/L	<	.25	mg/L	130.62	mg/L	<	.02	mg/L	<	.02	mg/L	
331	93230	Water															
332	93231	Water		.46	mg/L	<	.25	mg/L	38.56	mg/L	<	.02	mg/L	<	.02	mg/L	
333	93232	Water															
334	93233	Water		.32	mg/L	<	.25	mg/L	38.65	mg/L	<	.02	mg/L	<	.02	mg/L	
335	93234	Water															
336	93235	Water		.30	mg/L	<	.25	mg/L	15.78	mg/L	<	.02	mg/L	<	.02	mg/L	
337	93236	Water															
338	93237	Water		.29	mg/L	<	.25	mg/L	198.44	mg/L	<	.02	mg/L	<	.02	mg/L	
339	93238	Water															
340	93239	Water		.31	mg/L	<	.25	mg/L	13.62	mg/L	<	.02	mg/L	<	.02	mg/L	
341	93240	Water															
342	93241	Water		.26	mg/L	<	.25	mg/L	31.88	mg/L	<	.02	mg/L	<	.02	mg/L	
343	93242	Water															
344	93243	Water		.35	mg/L	<	.25	mg/L	34.26	mg/L	<	.02	mg/L	<	.02	mg/L	
345	93244	Water															
346	93245	Water		.41	mg/L	<	.25	mg/L	7.81	mg/L		.13	mg/L		.10	mg/L	
347	93246	Water															
348	93247	Water	<	.25	mg/L	<	.25	mg/L	9.34	mg/L	<	.02	mg/L	<	.02	mg/L	
349	93248	Water															
350	93249	Water		.47	mg/L	<	.25	mg/L	7.65	mg/L	<	.02	mg/L	<	.02	mg/L	
351	93250	Water	<	.25	mg/L	<	.25	mg/L	.69	mg/L	<	.02	mg/L	<	.02	mg/L	
352	93251	Water	<	.25	mg/L	<	.25	mg/L	.57	mg/L	<	.02	mg/L	<	.02	mg/L	
353	93252	Water	<	.25	mg/L	<	.25	mg/L	.53	mg/L	<	.02	mg/L	<	.02	mg/L	
354	93253	Water		5.47	mg/L		1.96	mg/L	21.37	mg/L		2.13	mg/L		5.48	mg/L	
355	93254	Water		5.45	mg/L		1.98	mg/L	21.35	mg/L		2.12	mg/L		5.48	mg/L	
356	93255	Water		5.34	mg/L		1.91	mg/L	21.41	mg/L		2.13	mg/L		5.47	mg/L	

Commentaires/ Comments: par ICP. Cl : par IC. Fe+3 : par colorimétrie.

Effectué par/ Work by : J. Groleau D. Thériault B. Legault R. Pelletier *11.11*

Ref.: V2-102-13-56 Date: 1/11/94

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To :	P.Tib	ble	M.Woysl	hner			PROJET	/ PRO	JECT	: V2-1	T03				Date: 1	/11/94
Lab #	I.D.	Descriptio	on	F	e			κ	- <u></u> .	M	g		M	n	N	8
327	93226	Water														
328	93227	Water	<	.02	mg/L	<	5.00	mg/L		21.39	mg/L	<	5.00	ug/L	20.70	mg/L
329	93228	Water														
330	93229	Water	<	.02	mg/L	<	5.00	mg/L		31.03	mg/L	<	5.00	ug/L	42.73	mg/L
331	93230	Water														
332	93231	Water	· <	.02	mg/L	<	5.00	mg/L		7.09	mg/L	<	5.00	ug/L	18.50	mg/L
333	93232	Water														
334	93233	Water		.43	mg/L		6.97	mg/L		9.47	mg/L	<	5.00	ug/L	13.74	mg/L
335	93234	Water									•					
336	93235	Water	<	.02	mg/L	<	5.00	mg/L		3.96	mg/L	<	5.00	ug/L	9.70	mg/L
337	93236	Water														
338	93237	Water		.03	mg/L		6.06	mg/L		23.89	mg/L		.28	mg/L	33.51	mg/L
339	93238	Water														
340	93239	Water		23.54	mg/L	<	5.00	mg/L		3.18	mg/L		4.77	mg/L	12.35	mg/L
341	93240	Water														
342	93241	Water		.20	mg/L	<	5.00	mg/L		5.62	mg/L		.47	mg/L	13.39	mg/L
343	93242	Water														
344	93243	Water		.12	mg/L	<	5.00	mg/L		5.24	mg/L		.07	mg/L	17.63	mg/L
345	93244	Water														
346	93245	Water		1.87	mg/L	<	5.00	mg/L		1.38	mg/L		.87	mg/L	13.80	mg/L
347	93246	Water														
348	93247	Water	•	.81	mg/L	<	5.00	mg/L		1.58	mg/L		1.17	mg/L	10.43	mg/L
349	93248	Water														
350	93249	Water		.86	mg/L	<	5.00	mg/L		1.40	mg/L		2.58	mg/L	14.54	mg/L
351	93250	Water	<	.02	mg/L	<	5.00	mg/L	<	.50	mg/L	<	5.00	ug/L	8.00	mg/L
352	93251	Water	<	.02	mg/L	<	5.00	mg/L	<	.50	mg/L	<	5.00	ug/L	7.25	mg/L
353	93252	Water	<	.02	mg/L	<	5.00	mg/L	<	.50	mg/L	<	5.00	ug/L	7.19	mg/L
354	93253	Water		21.88	mg/L		5.75	mg/L		19.90	mg/L		5.07	mg/L	12.42	mg/L
355	93254	Water		21.85	mg/L	<	5.00	mg/L		19.91	mg/L		5.06	mg/L	12.20	mg/L
356	93255	Water		21.85	mg/L		5.88	mg/L		19.87	mg/L		5.07	mg/L	12.43	mg/L

Commentaires/ Comments: par ICP. Cl : par IC. Fe+3 : par colorimétrie.

Effectué par/ Work by : J. Groleau D. Thériault

B. Legault R. Pelletier

11-1

Ref.: V2-102-13-56

1 -

CENTRE DE TECHNOLOGIE NORANDA

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To :	P.Tib	ble M	1.Woysi	hner		PROJET	/ PROJECT	: V2-1 T03	Ref.: V2-102-13-56 Date: 1/11/94
Lab #	I.D.	Description	n	Fe+	3	C	l		
327	93226	Water				5.98	mg/L		
328	93227	Water	<	.05	mg∕i.				
329	93228	Water				12.10	mg∕L		
330	93229	Water	<	.05	mg/L				
331	93230	Water				3.60	mg∕L		
332	93231	Water	<	.05	mg/L				
333	93232	Water				18.20	mg/L		
334	93233	Water		.50	mg/L				
335	93234	Water				4.04	mg/L		
336	93235	Water	<	.05	mg/L				
337	93236	Water				16.90	mg/L		
338	93237	Water	<	.05	mg/L				
339	93238	Water				2.02	mg/L	,	
340	93239	Water		1.15	mg/L				
341	93240	Water				12.80	mg/L		
342	93241	Water	<	.05	mg/L				
343	93242	Water				10.70	mg/L		
344	93243	Water	<	.05	mg/L				
345	93244	Water				2.24	mg/L		
346	93245	Water	<	.05	mg/L				
347	93246	Water				2.27	mg/L		
348	93247	Water	<	.05	mg/L				
349	93248	Water				2.32	mg/L		
350	93249	Water	<	.05	mg/L				
351	93250	Water	<	.05	mg/L	1.69	mg/L		
352	93251	Water	<	.05	mg/L	1.74	mg/L		
353	93252	Water	<	.05	mg/L	1.76	mg/L		
354	93253	Water		19.60	mg/L	394.00	mg/L		
355	93254	Water		14.60	mg/L	419.00	mg/L		
356	93255	Water		14.50	mg/L	421.00	mg/L		
							-		

Commentaires/ Comments: par ICP. Cl : par IC. Fe+3 : par colorimétrie.

Effectué par/ Work by : J. Groleau D. Thériault B. Legault R. Pelletier

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CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To : P.Tibble

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PROJET / PROJECT: V21 T03 100

Ref.: V2-105-12-56 Date: 6/28/94

Lab #	I.D.	Description		A	ι		A	S		C	a		C	d		C	0	
8090	94741	Water		.27	mg/L	<	.25	mg/L		2.81	mg/L	<	.03	mg/L	_ <	.03	mg/L	
8091	94742	Water	<	.25	mg/L	<	.25	mg/L	<	.05	mg/L	<	.03	mg/L	<	.03	mg/L	
8092	94743	Water	<	.25	mg/L	<	.25	mg/L		22.73	mg/L	<	.03	mg/L	<	.03	mg/L	
8093	94744	Water	<	.25	mg/L	<	.25	mg/L		9.01	mg/L	<	.03	mg/L	<	.03	mg/L	
8094	94745	Water	<	.25	mg/L	<	.25	mg/L		24.18	mg/L	<	.03	mg/L	<	.03	mg/L	
8095	94746	Water	<	.25	mg/L	<	.25	mg/L		30.11	mg/L	<	.03	mg/L	<	.03	mg/L	
8096	94747	Water	<	.25	mg/L	<	.25	mg/L		82.18	mg/L	<	.03	mg/L	<	.03	mg/L	
8097	94748	Water	<	.25	mg/L	<	.25	mg/L		123.19	mg/L	<	.03	mg/L	<	.03	mg/L	
8098	94749	Water	<	.25	mg/L	<	.25	mg/L		22.34	mg/L	<	.03	mg/L	<	.03	mg/L	
8099	94750	Water	<	.25	mg/L	<	.25	mg/L		36.87	mg/L	<	.03	mg/L	<	.03	mg/L	
8100	94751	Water	<	.25	mg/L	<	.25	mg/L		11.85	mg/L	<	.03	mg/L	<	.03	mg/L	
8101	94752	Water	<	.25	mg/L	<	.25	mg/L		201.52	mg/L	<	.03	mg/L	<	.03	mg/L	
8102	94753	Water	<	.25	mg/L	<	.25	mg/L		81.06	mg/L	<	.03	mg/L	<	.03	mg/L	
8103	94754	Water	<	.25	mg/L	<	.25	mg/L		42.70	mg/L	<	.03	mg/L	<	.03	mg/L	
8104	94755	Water	<	.25	mg/L	<	.25	mg/L		33.93	mg/L	<	.03	mg/L		26.07	ug/L	
8105	94756	Water	<	.25	mg/L	<	.25	mg/L		68.93	mg/L	<	.03	mg/L	<	.03	mg/L	
8106	94757	Water	<	.25	mg/L	<	.25	mg/L		58.51	mg/L	<	.03	mg/L	<	.05	mg/L	
8107	94758	Water -	<	.25	mg/L	< '	.25	mg/L		101.74	mg/L	<	.03	mg/L	<	.03	mg/L	
8108	94759	Water	<	.25	mg/L	<	.25	mg/L		40.74	mg/L	<	.03	mg/L	<	.03	mg/L	
8109	94760	Water	<	.25	mg/L	<	.25	mg/L		8.11	mg/L	<	.03	mg/L		61.56	ug/L	
8110	94761	Water	<	.25	mg/L	<	.25	mg/L		29.00	mg/L	<	.03	mg/L	<	.03	mg/L	
8111	94762	Water	<	.25	mg/L	<	.25	mg/L		29.25	mg/L	<	.03	mg/L	<	.03	mg/L	
8112	94763	Water		.28	mg/L	<	.25	mg/L		4.22	mg/L	<	.03	mg/L	<	.03	mg/L	
8113	94764	Water	<	.25	mg/L	<	.25	mg/L		5.28	mg/L	<	.03	mg/L		28.97	ug/L	
8114	94765	Water		2.89	mg/L	<	.25	mg/L		3.15	mg/L	<	.03	mg/L		.10	mg/L	
8115	94766	Water	<	.25	mg/L	<	.25	mg/L		466.45	mg/L	<	.03	mg/L	<	.03	mg/L	
8116	94767	Water	<	.25	mg/L	<	.25	mg/L		155.22	mg/L	<	.03	mg/L	<	.03	mg/L	
8117	94768	Water	<	.25	mg/L	<	.25	mg/L		334.95	mg/L	<	.03	mg/L	<	-03	mg/L	
8118	94775	Water	<	.25	mg/L	<	.25	mg/L		81.14	mg/L	<	.03	mg/L	< _	.05	mg/L	
8119	94//0	Water	<	. 25	mg/L	<	.25	mg/L		81.51	mg/L	<	.03	mg/L	<	.05	ing/L	
8120	94///	Water	<	.25	mg/L	<	.25	mg/L		81.20	mg/L	~	.05	mg/L	2	.05	mg/L	
0121	94110	Water		.25	mg/L	2	.25	mg/L		79.00	mg/L	2	.03	mg/L		.05	mg/L	
9122	947790	Water	2	.25	mg/L	2	.25	mg/L		19.01	mg/L	2	.05	mg/L	2	.05 03	mg/L	
812/	94700	Water	2	.25	mg/L	2	.25	mg/L		00.05	mg/L	2	20.	ma/L	2	20. ZN	mg/L	
8125	0/782	Water	2	.25	ma/L	2	- 25	mg/L	2	.05	ma/l	2	.05	ma/i	Ì	.03	mg/L mg/l	
8126	04702	Water	è.	.25	mg/L	2	.25	ma/L	2	.05	ma/l	2	.03	ma/l	è.	.03	ma/l	
8127	04784	Water	•	4 87	ma/l	•	2.03	ma/L	•	18.81	ma/i	•	2.20	ma/i	<	.03	ma/l	
8128	94785	Water		4.86	ma/l.		1.96	ma/L		19,15	mg/L		2.21	ma/L	<	.03	ma/L	
8129	94786	Water		4.93	ma/L		2.03	ma/L		19.04	ma/L		2.21	ma/L	<	.03	ma/L	
8130	94719	Water																
8131	94720	Water																
8132	94721	Water						•										
8133	94722	Water																
8134	94723	Water																
8135	94724	Water																
8136	94725	Water																
8137	94726	Water																
8138	94727	Water																
8139	94728	Water																
8140	94729	Water																
8141	94730	Water																
8142	94731	Water																
8143	94732	Water																
8144	94733	Water																
8145	94/54	Water																
8146	94/55 0/77/	water																
014/	74130 01777	water																
0140	74/3/ 0/770	water Noter																
0149	94130 01720	Water Voton																
010	74137	Walci																
Commer	taires/	Comments: par	ICP.	Cl et s	504 pa	r IC.	Fe+3	oar co	lori	métrie.								

A. Bouchard J. Groleau

Effectué par/ Work by : D. Thériault R. Pelletier MH

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CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To :	: P.Tib	ble		PROJET / PROJEC		Ref.: V2-105-12-56 Date: 6/28/94	
Lab #	I.D.	Description	AL	As	Ca	Cd	Со
8151	94740	Water		· · · · · · · · · · · · · · · · · · ·			

Commentaires/ Comments: par ICP. Cl et SO4 par IC. Fe+3 par colorimétrie.

Effectué par/ Work by : D. Thériault R. Pelletier A. Bouchard

J. Groleau

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To :	P.Tibble	PROJET / PROJECT:	V21 T03 100	Dat

Ref.: V2-105-12-56 Date: 6/28/94

Lab #	I.D.	Description	Cr	Cu	Fe	К	Mg
8151	94740	Water					

Commentaires/ Comments: par ICP. Cl et SO4 par IC. Fe+3 par colorimétrie.

Effectué par/ Work by : D. Thériault R. Pelletier A. Bouchard J. Groleau

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To : P.Tibble

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PROJET / PROJECT: V21 T03 100

Ref.: V2-105-12-56 Date: 6/28/94

• •	I.D.	Description		Mn		ł	la		Ni			P	b			S
3090	94741	Water		83.09	ug/L	3.87	mg/L		.27	mg/L	<	.25	mg/L		9.98	mg/L
8091	94742	Water		55.77	ug/L	3.58	mg/L		82.76	ug/L	<	.25	mg/L		4.14	mg/L
8092	94743	Water	<	.01	mg/L	8.54	mg/L	<	.03	mg/L	<	.25	mg/L		7.87	mg/L
8093	94744	Water	<	.01	mg/L	4.66	ma/L	<	.03	mg/L	<	.25	mg/L		5.12	mai/L
8094	94745	Water	<	.01	ma/l	6.91	ma/l	<	.03	ma/l	<	.25	ma/1		7.47	mo/l
8005	0/7/6	Ustor	Ż	.01	ma/i	11 10	ma/l	2	.03	ma/i	è	25	ma/l		8 01	ma/l
20075	04747	Untor	2	.01	ma/L	15 59	mm / L	2	03	mg/L	2	- 25	mg/L		71 20	mg/L mg/l
0090	94/4/	Waler		.01	ing/L	13.30	mg/L		.03	mg/L	2	.25	1119/L		11.27	1119/L
0097	94740	water	ς.	.01	mg/L	27.09	mg/L	<	.03	ing/L	•	.23	mg/L		122.41	mg/L
8098	94749	Water	<	.01	mg/L	9.75	mg/L	<	.03	mg/L	<	.25	mg/L		9.49	mg/L
8099	94750	Water	<	.01	mg/L	10.78	mg/L	<	.03	mg/L	<	.25	mg/L		12.52	mg/L
8100	94751	Water	<	.01	mg/L	5.37	mg/L	<	.03	mg/L	<	.25	mg/L		8.71	mg/L
8101	94752	Water		94.47	ug/L	32.67	mg/L	<	.03	mg/L	<	.25	mg/L		188.40	mg/L
8102	94753	Water		.44	mg/L	15.00	mg/L	<	.03	mg/L	<	.25	mg/L		58.29	mg/L
8103	94754	Water		.20	mg/L	13.22	mg/L	<	.03	mg/L	<	.25	mg/L		16.57	mg/L
8104	94755	Water		36.22	ug/L	16.44	ma/L	<	.03	ma/l.	<	.25	ma/L		10.33	ma/L
8105	94756	Water		69.15	ua/1	11.80	ma/l		42.21	ua/1	<	.25	ma/l		33.83	ma/1
8106	94757	Vater		28	ma/l	33 08	ma/l		37 86	ua/l	ح	25	ma/l		35 71	ma/i
0100	74121	•		.20	11-97 -	55.00	ilig/ L		57.00	ug/ L	•	.23	arg/ L		JJ . / 1	mg/ L
8107	94758	Water •		.20	mg/L	12.52	mg/L		67.89	ug/L	<	.25	mg/L		64.20	mg/L
8108	94759	Water		.13	mg/L	20.40	mg/L	<	.03	mg/L	<	.25	mg/L		12.46	mg/L
8109	94760	Water		2.69	mg/L	7.77	mg/L		.54	mg/L	<	.25	mg/L		9.66	mg/L
8110	94761	Water		.11	mg/L	11.17	mg/L		65.72	ug/L	<	.25	mg/L		9.77	mq/L
8111	94762	Water		40	ma/i	11_03	ma/l	<	.03	ma/l	<	25	ma/l		9.96	ma/l
8112	94763	Water		44	mo/l	R 44	mo /1	•	.0J 41	ma/l	< l		ma/1		8 85	mo/i
8117	94.764	Uator		44 84	ma/l	7 /0	mm/L mm/l		17. DZ	mer/i	è		mor/l		2 02	may/⊾ mai/l
914/	0/.74F	Hater		1 20	mg/L	140	1119/L	•	4 00	mg/L	2	. 2.) DE	1119/L		0.70	1179/L
0114	74/00 0/7//	water		1.00	ng/L	11.20	ing/L		1.20	189/L	5	.27	mg/L		7.23	iiig/L
0115	94/00	water		1.19	mg/L	28.18	mg/L		20.72	ug/L	<	.25	mg/L		310.94	mg/L
8116	94767	Water		.30	mg/L	159.48	mg/L		/7.90	ug/L	<	.25	mg/L		365.95	mg/L
8117	94768	Water		.34	mg/L	331.55	mg/L		. 14	mg/L	<	.25	mg/L		859.59	mg/L
8118	94775	Water	<	.01	mg/L	18.41	mg/L	<	.03	mg/L	<	.25	mg/L		73.21	mg/L
8119	94776	Water	<	.01	mg/L	18.67	mg/L	<	.03	mg/L	<	.25	mg/L		72.39	mg/L
8120	94777	Water	. <	.01	mg/L	18.73	mg/L	<	.03	mg/L	<	.25	mg/L		72.87	mg/L
8121	94778	Water	<	.01	mg/L	17.89	ma/L	<	.03	ma/L	<	.25	mg/L		71.83	mg/L
8122	94779	Water		6.58	uα/I	17.97	ma/l	<	70,	ma/l	<	.25	ma/l		70.50	ma/1
8123	94780	Water	<	.01	ma/l	18 20	ma/l	è	נט. דח	mci/l	è.	25	moi/i		72 35	mor/i
812/	0/.791	Uatar	2	.01	ma /1	E 77	ma/i	2	.05	me /i	2	25			2.35 2F	mm / L
0124 0125	01707	Hater Votor	2	.01	mg/L	J.[[E E/	mg/L	2	.03	my/L	2	.23	1119/L	2	- 23	1119/L
0127	74/02	Water		.01	ng/L	5.54	1897L	<u>ج</u>	.03		2	.23	1119/L		. 23	1119/L
0120	Y4/05	water	<	.01	mg/L	5.58	mg/L	<	.03	mg/L	<	· <u></u>	mg/L	<	.25	mg/L
8127	94784	Water		5.03	mg/L	10.57	mg/L		5.65	mg/L		5.77	mg/L		927.59	mg/L
8128	94785	Water		5.10	mg/L	10.47	mg/L		5.75	mg/L		5.80	mg∕L		939.68	mg/L
8129	94786	Water		5.12	mg/L	10.37	mg/L		5.74	mg/L		5.80	mg/L		944.44	mg/L
8130	94719	Water					-									
8131	94720	Water														
8132	94721	Water														
8133	94722	Water														
817/	04727	Vator														
0175	77163	Votor														
0133	74124	water Nata-														
0156	94725	Water														
8157	94726	Water														
8138	94727	Water														
8139	94728	Water														
8140	94729	Water														
8141	94730	Water														•
8142	94731	Water														
8143	94732	Water														
81/./	0/777	Uston														
81/5	0/72/	Water Uston														
0147	74134	water														
8146	94755	Water														
8147	94736	Water														
8148	94737	Water														
A4/A	94738	Water														
8149																

Effectué par/ Work by : D. Thériault R. Pelletier A. Bouchard

J. Groleau
CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To :	: P.Tik	ble		PROJET / PROJEC	r: V21 T03 100		Ref.: V2-105-12-56 Date: 6/28/94	
Lab #	I.D.	Description	Mn	Na	Ni	Pb	S	

8151 94740 Water

Commentaires/	Comments:	par ICP.	Cl et SO4 par	IC. Fe+	3 par colorimétrie.
Effectué par/	Work by :	D. Théria R. Pellet	ult ier	A. J.	Bouchard Groleau

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To : P.Tibble

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PROJET / PROJECT: V21 T03 100

Ref.: V2-105-12-56 Date: 6/28/94

Lab #	I.D.	Descriptio	on	s	b		S	e	Si	i		т	e		Т	ι
8090	94741	Water	<	.25	mg/L	<	.50	mg/L	.31	mg/L	<	.10	mg/L	<	.25	mg/L
8091	94742	Water	<	.25	mg/L	<	.50	mg/L	.41	mg/L	<	.10	mg/L	<	.25	mg/L
8092	94743	Water	<	.25	mg/L	<	.50	mg/L	.17	mg/L	<	. 10	mg/L	<	.25	mg/L
8093	94744	Water	<	.25	mg/L	<	.50	mg/L	.25	mg/L	<	.10	mg/L	<	.25	mg/L
8094	94745	Water	<	.25	mg/L	<	.50	mg/L	70.52	ug/L	<	.10	mg/L	<	.25	mg/L
8095	94746	Water	<	.25	mg/L	<	.50	mg/L	.46	mg/L	<	.10	mg/L	<	.25	mg/L
8096	94747	Water	<	.25	mg/L	<	.50	mg/L	8.98	mg/L	<	.10	mg/L	<	.25	mg/L
8097	94748	Water	<	.25	mg/L	<	.50	mg/L	8.93	mg/L	<	.10	mg/L	<	.25	mg/L
8098	94749	Water	<	.25	mg/L	<	.50	mg/L	5.45	mg/L	<	.10	mg/L	<	.25	mg/L
8099	94750	Water	<	.25	mg/L	<	.50	mg/L	6.34	mg/L	<	.10	mg/L	<	.25	mg/L
8100	94/31	Water	<	.27	mg/L	5	.50	mg/L	0.01	mg/L	<	.10	mg/L	<	.20	mg/L
9103	94122	Water		.25	ng/L	2	.50	mg/L	2.99	ing/L	2	. 10	mg/L	2	.20	mg/L
8102	94755	Ustor		.25	mg/L	2	- 50	mg/L	6.50	mg/L	2	10	mg/L	2	.25	mg/L
8104	94755	Water	č	.25	mg/t	è	50	ma/i	3 26	mg/L		13	mg/∟ mg/l	2	25	mg/L
8105	94756	Water	, ,	.25	ma/l	è.	.50	ma/l	6.62	ma/l	<	.10	mor/l	è	.25	mg/L
8106	94757	Water	<	.25	ma/L	<	.50	ma/l	6.57	ma/i	۲, k	.10	ma/l	۲, k	.25	mor/1
	,								••••					-		
8107	94758	Water	<	.25	mg/L	< -	.50	mg/L	5.47	mg/L	<	.10	mg/L	<	.25	mg/L
8108	94759	Water	<	.25	mg/L	<	.50	mg/L	4.71	mg/L	<	.10	mg/L	<	.25	mg/L
8109	94760	Water	<	.25	mg/L	<	.50	mg/L	10.43	mg/L		.11	mg/L	<	.25	mg/L
8110	94761	Water	<	.25	mg/L	<	.50	mg/L	6.58	mg/L	<	.10	mg/L	<	.25	mg/L
8111	94762	Water	<	.25	mg/L	<	.50	mg/L	6.43	mg/L	<	.10	mg/L	<	.25	mg/L
8112	94763	Water	<	.25	mg/L	<	.50	mg/L	7.10	mg/L	<	.10	mg/L	<	.25	mg/L
8113	94764	Water	<	.25	mg/L	<	.50	mg/L	4.28	mg/L	<	.10	mg/L	<	.25	mg/L
8114	94765	Water	<	.25	mg/L	<	.50	mg/L	8.75	mg/L	<	.10	mg/L	<	.25	mg/L
8115	94/00	Water		.25	mg/L	<	.50	mg/L	4.48	mg/L	<	.10	mg/L	<	.25	mg/L
0110	94101	Water	<	.20	mg/L	~	.50	mg/L	4.99	mg/L		.14	mg/L	< _	.25	mg/L
0117 9119	94700	Water		.31	mg/L	2	.50	mg/L	0.90 9.07	1119/L		- 10	mg/L	2	.25	mg/L
8110	94775	Waler Vator	2	-25	mg/L	2	.50	mg/L	0.93	mg/L	2	10	mg/L	2	.25	mg/L
8120	0%777	Water	, i i i i i i i i i i i i i i i i i i i	.25	ma/L	2	.50	mar/L	9.07	mg/L	2	10	mg/L	2	-25	mg/L mg/l
8121	94778	Water	, k	.25	ma/i	à	50	mor/i	9.04 8.83	mg/L	è	10	ma/L	Ì	.25	mer/i
8122	94770	Water	~	25	ma/i	Ż	50	ma/i	8.89	ma/l	2	10	ma/i	2	- 25	mg/L mg/l
8123	94780	Water	, ,	.25	ma/l	- k	.50	ma/l	8.95	ma/i	~	· .10	ma/t	è	.25	ma/t
8124	94781	Water	۰ ۲	-25	ma/L	< l	.50	ma/i	.21	ma/l	, K	.10	ma/l	ż	.25	ma/i
8125	94782	Water	<	.25	mq/L	<	.50	mq/L	.24	mg/L		.12	ma/L	<	.25	mg/L
8126	94783	Water	<	.25	mg/L	<	.50	mg/L	.21	mg/L	<	.10	mg/L	<	.25	mg/L
8127	94784	Water	<	.25	mg/L	<	.50	mg/L	7.47	mg/L	<	.10	mg/L	<	.25	mg/L
8128	94785	Water	<	.25	mg/L	<	.50	mg/L	7.59	mg/L	<	.10	mg/L	<	.25	mg/L
8129	94786	Water	<	.25	mg/L	<	.50	mg/L	7.55	mg/L	<	.10	mg/L	<	.25	mg/L
8130	94719	Water														
8131	94720	Water														
8132	94721	Water														
8133	94722	Water														
8134	94723	Water														
8135	94/24	Water														
8136	94/25	Water														
015/	94720	Water														
9170	74121	Water														
0139 91/0	74120	water Votor														
81/1	74/ 67 0/ 720	Water							·							
81/2	94730 96731	Water Vator														
8147	94732	Water														
8144	94732	Water														
8145	94734	Water														
8146	94735	Water														
8147	94736	Water														
8148	94737	Water														
8149	94738	Water														
8150	94739	Water														
Commer	ntaires/	Comments: pa	or ICP. C	let	504 pai	r IC.	Fe+3	par col	lorimétrie.							
Effect	ué par/	Work by : D.	Thériaul	t			A. B	ouchard	. E							

J. Groleau

Effectué par/ Work by : D. Thériault R. Pelletier

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To	: P.Til	bble		PROJET / PROJECT:	V21 T03 100		Ref.: V2-105-12-56 Date: 6/28/94	
Lab #	I.D.	Description	Sb	Se	Si	Te	τι	-

8151 94740 Water

Commentaires/ Comment	s: par ICP. Cl et SO4 par IC:	. Fe+3 par colorimétrie.	
Effectué par/ Work by	/: D. Thériault R. Pelletier	A. Bouchard J. Groleau	

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To : P.Tibble

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PROJET / PROJECT: V21 T03 100

Ref.: V2-105-12-56 Date: 6/28/94

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Lab #	I.D.	Descrip	tion		z	n		C	l	SO	4		Fe+	3	
8090	94741	Water			50.67	ug/L						<	.05	mg/L	
8091	94742	Water		< .	.03	ma/L						<	.05	ma/L	
8002	04743	Water		<	.03	ma/1						<	.05	ma/l.	
8093	94744	Water		<	.03	mg/L						<	.05	mg/L	-
8004	94745	Water		<	.03	ma/1						<	.05	ma/L	
8095	94746	Water		<	.03	ma/L						<	.05	mg/L	
8006	94747	Water		<	.03	mo/i						<	.05	ma/L	
8007	94748	Water		<	.03	ma/l						<	.05	ma/L	
8098	94749	Water		<	.03	ma/L						<	.05	mg/L	
8099	94750	Water		<	.03	ma/L						<	.05	ma/L	
8100	94751	Water		<	.03	ma/L						<	.05	ma/L	
8101	94752	Water		<	.03	ma/L						<	.05	mq/L	
8102	94753	Water		<	.03	ma/L							440.00	ug/L	
8103	94754	Water		<	.03	ma/L						<	.05	ma/L	
8104	94755	Water		:	28.77							<	.05	mq/L	
8105	94756	Water		i	51.29	ug/L						<	.05	mq/L	
8106	94757	Water		<	.03	ma/L						<	.05	ma/L	
•••••		•													
8107	94758	Water		<	.03	mg/L							1.16	mg/L	
		-											1.18	mg/L	
8108	94759	Water		<	.03	mg/L						<	.05	mg/L	
8109	94760	Water			. 15	mg/L							543.00		
8110	94761	Water		1	88.83	ug/L						<	.05	mg/L	
8111	94762	Water		<	.03	mg/L						<	.05	mg/L	
8112	94763	Water		1	55.05	ug/L						<	.05	mg/L	
8113	94764	Water		<	.03	mg/L							617.00	ug/L	
8114	94765	Water			.27	mg/L							558.00	ug/L	
8115	94766	Water		<	.03	mg/L						<	.05	mg/L	
8116	94767	Water		<	.03	mg/L						<	.05	mg/L	
8117	94768	Water		<	.03	mg/L						<	.05	mg/L	
8118	94775	Water		<	.03	mg/L						<	.05	mg/L	
8119	94776	Water		<	.03	mg/L						<	.05	mg/L	
8120	94777	Water		<	.03	mg/L						<	.05	mg/L	
8121	94778	Water		<	.03	mg/L						<	.05	mg/L	
8122	94779	Water		<	.03	mg/L						<	.05	mg/L	
8123	94780	Water		<	.03	mg/L						<	.05	mg/L	
8124	94781	Water		<	.03	mg/L						<	.05	mg/L	
8125	94782	Water		<	.03	mg/L						<	.05	mg/L	
8126	94783	Water		<	.03	mg/L						<	.05	mg/L	
8127	94784	Water			23.76	mg/L							19.10	mg/L	
8128	94785	Water			24.00	mg/L							18.80	mg/L	
8129	94786	Water			24.09	mg/L							19.10	mg/L	
8130	94719	Water						6.31	mg/L	210.00	mg/L				
8131	94720	Water						10.50	mg/L	318.00	mg/L				
8132	94721	Water						2.15	mg/L	32.40	mg/L				
8133	94722	Water						14.70	mg/L	41.00	mg/L				
8134	94723	Water						1.77	mg/L	29.60	mg/L				
8135	94724	Water						23.60	mg/L	412.00	mg/L				
8136	94725	Water						19.90	mg/L	166.00	mg/L				
8137	94726	Water						42.90	mg/L	50.60	mg/L				
8138	94727	Water						53.70	mg/L	49.80	mg/L				
8139	94728	Water						6.62	mg/L	97.70	mg/L				
8140	94729	Water						4.76	mg/L	104.00	mg/L				
8141	94730	Water						5.53	mg/L	187.00	mg/L				
8142	94731	Water						35.60	mg/L	35.60	mg/L				
8143	94732	Water					<	.20	mg/L	27.00	mg/L				
8144	94733	Water						11.40	mg/L	27.80	mg/L				
8145	94734	Water						11.40	mg/L	28.10	mg/L				
8146	94735	Water					<	.20	mg/L	25.60	mg/L				
8147	94736	Water						1.91	mg/L	9.43	mg/L				
8148	94737	Water						1.80	mg/L	27.60	mg/L				
8149	94738	Water						2.71	mg/L		-				
				<u></u>											
Commen	taires/	Comments:	par I	ср. С	let	504 par	10.	Fe+3 p	ar colo	rimétrie.					
Effect	ué par/	Work by :	D. Th	ériaul	t			A. Bo	ouchard						
			R. Pe	lletie	۲			J. Gr	roleau						
L						· · · · ·				·					

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To :	P.Tib	ble		PROJET / PROJECT		Ref.: V2-105-12-56 Date: 6/28/94	
Lab #	I.D.	Description	Zn	CL	S04	Fe+3	
8150	94739	Water	·····	38.20 mg/L			
8151	94740	Water		40.60 mg/L			

Commentaires/ Comments: par ICP. Cl et SO4	par IC. Fe+3 par colorimétrie.	
Effectué par/ Work by : D. Thériault R. Pelletier	A. Bouchard J. Groleau	· · · · · · · · · · · · · · · · · · ·

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To : M.Woyshner

PROJET / PROJECT: V21 T03 100

Ref.: T03001-5-53 Date: 10/03/94

Lab #	I.D.	Description	AL	A	s	Ca	a		Co	1	_	C	0	
13960	941784	Fault Lake	<u>.</u>											
13961	941785	Fault Lake												
13962	941786	Fault Lake												
13963	941787	Fault Lake												
13964	941788	Fault Lake												
13965	941789	Fault Lake				÷								
13966	941790	Fault Lake												
13967	941791	Fault Lake												
13968	941792	Fault Lake												
13969	941793	Fault Lake												
13970	941794	Fault Lake												
13971	941795	Fault Lake												
13972	941796	Fault Lake												
13973	941797	Fault Lake												
13974	941798	Fault Lake												
13975	941799	Fault Lake												
13976	941800	Fault Lake												
13977	941801	Fault Lake												
13978	941802	Fault Lake												
13979	941803	Fault Lake												
13980	941804	Fault Lake	.75 mg/L	< .25	mg/L	89.67	mg/L	<	.03	mg/L	<	.03	mg/L	
13981	941805	Fault Lake	.65 mg/L	< .25	mg/L	127.57	mg/L	<	.03	mg/L	<	.03	mg/L	
13982	941806	Fault Lake	.66 mg/L	< .25	mg/L	40.24	mg/L	<	.03	mg/L	<	.03	mg/L	
13983	941807	Fault Lake	.62 mg/L	< .25	mg/L	15.68	mg/L	<	.03	mg/L	<	.03	mg/L	
13984	941808	Fault Lake	.76 mg/L	< .25	mg/L	201.77	mg/L	<	.03	mg/L	<	.03	mg/L	
13985	941809	Fault Lake	.76 mg/L	< .25	mg/L	171.70	mg/L	<	.03	mg/L	<	.03	mg/L	
13986	941810	Fault Lake	.53 mg/L	.< .25	mg/L	82.21	mg/L	<	.03	mg/L	<	.03	mg/L	
13987	941811	Fault Lake	.66 mg/L	< .25	mg/L	42.33	mg/L	<	.03	mg∕L	<	.03	mg/L	
13988	941812	Fault Lake	.77 mg/L	< .25	mg/L	40.04	mg∕L	<	.03	mg/L	<	.03	mg/L	
13989	941813	Fault Lake	.78 mg/L	< .25	mg/L	74.30	mg/L	<	.03	mg/L	<	.03	mg/L	
13990	941814	Fault Lake	.62 mg/L	< .25	mg/L	69.93	mg/L	<	.03	mg/L	<	.03	mg/L	
13991	941815	Fault Lake	.62 mg/L	< .25	mg/L	106.15	mg/L	<	.03	mg/L	<	.03	mg/L	
13992	941816	Fault Lake	.60 mg/L	< .25	mg/L	46.95	mg/L	<	.03	mg/L	<	.03	mg/L	
13993	941817	Fault Lake	.69 mg/L	< .25	mg/L	39.32	mg/L	<	.03	mg/L	<	.03	mg/L	
13994	941818	Fault Lake	.74 mg/L	< .25	mg/L	40.21	mg/L	<	.03	mg/L	<	.03	mg/L	
13995	941819	Fault Lake	.65 mg/L	< .25	mg/L	11.29	mg/L	<	.03	mg/L	<	.03	mg/L	
13996	941820	Fault Lake	1.84 mg/L	< .25	mg/L	9.01	mg/L	<	.03	mg/L		30.56	ug/L	
13997	941821	Fault Lake	.74 mg/L	< .25	mg/L	433.73	mg/L	<	.03	mg/L	<	.03	mg∕L	
13998	941822	Fault Lake	.56 mg/L	< .25	mg/L	19.89	mg/L	<	.03	mg/L	<	.03	mg/L	
13999	941823	Fault Lake	.51 mg/L	< .25	mg/L	174.05	mg/L	<	.03	mg∕L	<	.03	mg/L	

Commentaires/ Comments: By ICP , except Fe+3: by colorimetry. Anions by IC.

Effectué par/ Work by : L. Lavoie B. Legault J. Groleau D. Thériault N.J. for m. M. H.

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To : M.Woyshner

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PROJET / PROJECT: V21 T03 100

Ref.: T03001-5-53 Date: 10/03/94

Lab #	I.D.	Description		C	r		С	u		F	e			ĸ	м	g	
13960	941784	Fault Lake															
13961	941785	Fault Lake															
13962	941786	Fault Lake															
13963	941787	Fault Lake														•	
13964	941788	Fault Lake															
13965	941789	Fault Lake															
13966	941790	Fault Lake															
13967	941791	Fault Lake															
13968	941792	Fault Lake															
13969	941793	Fault Lake															
13970	941794	Fault Lake															
13971	941795	Fault Lake															
13972	941796	Fault Lake															
13973	941797	Fault Lake															
13974	941798	Fault Lake															
13975	941799	Fault Lake															
13976	941800	Fault Lake															
13977	941801	Fault Lake															
13978	941802	Fault Lake															
13979	941803	Fault Lake															
13980	941804	Fault Lake	<	.03	mg/L	<	.03	mg/L		40.33	ug/L	<	5.00	mg/L	21.29	mg/L	
13981	941805	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L	<	5.00	mg/L	31.14	mg/L	
13982	941806	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L	<	5.00	mg/L	9.75	mg/L	
13983	941807	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L	<	5.00	mg/L	3.50	mg/L	
13984	941808	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L		6.15	mg/L	32.08	mg/L	
13985	941809	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L		15.35	mg/L	12.39	mg/L	
13986	941810	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L		8.76	mg/L	19.25	mg/L	
13987	941811	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L		9.19	mg/L	9.92	mg/L	
13988	941812	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L		5.74	mg/L	7.63	mg/L	
13989	941813	Fault Lake	<	.03	mg/L	<	.03	mg/L		.27	mg/L		6.36	mg/L	10.02	mg/L	
13990	941814	Fault Lake	<	.03	mg/L	<	.03	mg/L		.47	mg/L		6.11	mg/L	10.56	mg/L	
13991	941815	Fault Lake	<	.03	mg/L	<	.03	mg/L		52.01	ug/L		9.03	mg/L	15.44	mg/L	
13992	941816	Fault Lake	<	.03	mg/L	<	.03	mg/L		.11	mg/L	<	5.00	mg/L	8.96	mg/L	
13993	941817	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L	<	5.00	mg/L	6.06	mg/L	
13994	941818	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L	<	5.00	mg/L	6.39	mg/L	
13995	941819	Fault Lake	<	.03	mg/L	<	.03	mg/L		.34	mg/L	<	5.00	mg/L	1.80	mg/L	
13996	941820	Fault Lake	<	.03	mg/L		28.25	ug/L		.17	mg/L	<	5.00	mg/L	1.32	mg/L	
13997	941821	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L		68.12	mg/L	168.25	mg/L	
13998	941822	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L	<	5.00	mg/L	3.95	mg/L	
13999	941823	Fault Lake	<	.03	mg/L	<	.03	mg/L	<	.03	mg/L		15.21	mg/L	12.49	mg/L	

Commentaires/ Comments: By ICP , except Fe+3: by colorimetry. Anions by IC.

Effectué par/ Work by : L. Lavoie B. Legault J. Groleau D. Thériault

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

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PROJET / PROJECT: V21 T03 100

Ref.: T03001-5-53 Date: 10/03/94

.ab #	I.D.	Description		M	n	N	8		N	i		P	b	:	S	
13960	941784	Fault Lake			-											
13961	941785	Fault Lake														
13962	941786	Fault Lake														
13963	941787	Fault Lake							٤							
13964	941788	Fault Lake														
13965	941789	Fault Lake														
13966	941790	Fault Lake														
13967	941791	Fault Lake														
13968	941792	Fault Lake														
13969	941793	Fault Lake														
13970	941794	Fault Lake														
13971	941795	Fault Lake														
13972	941796	Fault Lake														
13973	941797	Fault Lake														
13974	941798	Fault Lake														
13975	941799	Fault Lake														
13976	941800	Fault Lake														
13977	941801	Fault Lake														
13978	941802	Fault Lake														
13979	941803	Fault Lake														
13980	941804	Fault Lake	<	.01	mg/L	18.74	mg/L	<	.03	mg/L	<	.25	mg/L	78.86	mg/L	
13981	941805	Fault Lake	<	.01	mg/L	27.32	mg/L	<	.03	mg/L	<	.25	mg/L	118.32	mg/L	
13982	941806	Fault Lake	<	.01	mg/L	13.25	mg/L	<	.03	mg/L	<	.25	mg/L	15.45	mg/L	
13983	941807	Fault Lake		5.44	ug/L	8.45	mg/L	<	.03	mg/L	<	.25	mg/L	10.90	mg/L	
13984	941808	Fault Lake	•	.13	mg/L	34.73	mg/L	<	.03	mg/L	<	.25	mg/L	192.05	mg/L	
13985	941809	Fault Lake		.16	mg/L	12.37	mg/L		1.43	mg/L	<	.25	mg/L	111.43	mg/L	
13986	941810	Fault Lake		.48	mg/L	15.77	mg/L	<	.03	mg/L	<	.25	mg/L	62.82	mg/L	
13987	941811	Fault Lake		.25	mg/L	14.82	mg/L	<	.03	mg/L	<	.25	mg/L	18.33	mg/L	
13988	941812	Fault Lake		16.32	ug/L	16.90	mg/L	<	.03	mg/L	<	.25	mg/L	11.50	mg/L	
13989	941813	Fault Lake		88.44	ug/L	9.97	mg/L	<	.03	mg/L	<	.25	mg/L	37.01	mg/L	
13990	941814	Fault Lake		.38	mg/L	25.04	mg/L	<	.03	mg/L	<	.25	mg/L	39.78	mg/L	
13991	941815	Fault Lake		.22	mg/L	14.38	mg/L	<	.03	mg/L	<	.25	mg/L	65.88	mg/L	
13992	941816	Fault Lake		.13	mg/L	21.69	mg/L	<	.03	mg/L	<	.25	mg/L	14.42	mg/L	
13993	941817	Fault Lake		66.65	ug/L	12.69	mg/L		77.01	ug/L	<	.25	mg/L	10.64	mg/L	
13994	941818	Fault Lake		.53	mg/L	14.63	mg/L		83.61	ug/L	<	.25	mg/L	11.77	mg/L	
13995	941819	Fault Lake	•	.87	mg/L	8.52	mg/L		40.92	ug/L	. <	.25	mg/L	4.20	mg/L	
13996	941820	Fault Lake		1.16	mg/L	9.97	mg/L		.74	mg/L	<	.25	mg/L	7.12	mg/L	
13997	941821	Fault Lake		1.06	mg/L	26.77	mg/L		.12	mg/L	<	.25	mg/L	558.56	mg/L	
13998	941822	Fault Lake		15.27	ug/L	7.31	mg/L	<	.03	mg/L	<	.25	mg/L	6.10	mg/L	
13999	941823	Fault Lake		.17	mg/L	12.59	mg/L		1.28	mg/L	<	.25	mg/L	114.37	mg/L	

Commentaires/ Comments: By ICP , except Fe+3: by colorimetry. Anions by IC.

Effectué par/ Work by : L. Lavoie B. Legault J. Groleau D. Thériault

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

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PROJET / PROJECT: V21 T03 100

Ref.: T03001-5-53 Date: 10/03/94

Lab #	I.D.	Description		Sb		S	e	S	i		Te	•		T	l	
13960	941784	Fault Lake			····											
13961	941785	Fault Lake														
13962	941786	Fault Lake														
13963	941787	Fault Lake													×	
13964	941788	Fault Lake														
13965	941789	Fault Lake														
13966	941790	Fault Lake														
13967	941791	Fault Lake														
13968	941792	Fault Lake														
13969	941793	Fault Lake														
13970	941794	Fault Lake														
13971	941795	Fault Lake														
13972	941796	Fault Lake														
13973	941797	Fault Lake														
13974	941798	Fault Lake														
13975	941799	Fault Lake														
13976	941800	Fault Lake														
13977	941801	Fault Lake														
13978	941802	Fault Lake														
13979	941803	Fault Lake									÷					
13980	941804	Fault Lake	<	.25 mg	;/L <	.50	mg/L	9.25	mg/L	<	.10	mg/L	<	.25	mg/L	
13981	941805	Fault Lake	<	.25 mg	j/L <	.50	mg/L	9.25	mg/L	<	.10	mg/L	<	.25	mg/L	
13982	941806	Fault Lake	<	.25 mg	j/L <	.50	mg/L	6.40	mg/L	<	.10	mg/L	<	.25	mg/L	
13983	941807	Fault Lake	<	.25 mg	j/L <	.50	mg/L	6.56	mg/L	<	.10	mg/L	<	.25	mg/L	
13984	941808	Fault Lake	<	.25 mg	j/L <	.50	mg/L	6.06	mg/L	<	.10	mg/L	<	.25	mg/L	
13985	941809	Fault Lake	<	.25 mg	j/L <	.50	mg/L	4.70	mg/L	<	.10	mg/L	<	.25	mg/L	
13986	941810	Fault Lake	<	.25 mg	j/L <	.50	mg/L	6.20	mg/L	<	.10	mg/L	<	.25	mg/L	
13987	941811	Fault Lake	<	.25 mg	g/L <	.50	mg/L	7.37	mg/L	<	.10	mg/L	<	.25	mg/L	
13988	941812	Fault Lake	<	.25 mg	g/L <	.50	mg/L	3.25	mg/L	<	.10	mg/L	<	.25	mg/L	
13989	941813	Fault Lake	<	.25 mg	j/L <	.50	mg/L	6.61	mg/L	<	.10	mg/L	<	.25	mg/L	
13990	941814	Fault Lake	<	.25 mg	;/L <	.50	mg/L	6.84	mg/L	<	.10	mg/L	<	.25	mg/L	
13991	941815	Fault Lake	<	.25 mg	j/L <	.50	mg/L	5.43	mg/L	<	.10	mg/L	<	.25	mg/L	
13992	941816	Fault Lake	<	.25 mg	g/L <	.50	mg/L	4.84	mg/L	<	.10	mg/L	<	.25	mg/L	
13993	941817	Fault Lake	<	.25 mg	g/L <	.50	mg/L	6.63	mg/L	<	.10	mg/L	<	.25	mg/L	
13994	941818	Fault Lake	<	.25 mg	3/L <	.50	mg/L	6.01	mg/L	<	.10	mg/L	<	.25	mg/L	
13995	941819	Fault Lake	<	.25 mg	j/L <	.50	mg/L	4.11	mg/L	<	.10	mg/L	<	.25	mg/L	
13996	941820	Fault Lake	<	.25 mg	g/L <	.50	mg/L	6.97	mg/L	<	.10	mg/L	<	.25	mg/L	
13997	941821	Fault Lake	<	.25 m	g/L <	.50	mg/L	4.37	mg/L	<	.10	mg/L	<	.25	mg/L	
13998	941822	Fault Lake	<	.25 mg	;/L <	.50	mg/L	.97	mg/L	<	.10	mg/L	< ∘	.25	mg/L	
13999	941823	Fault Lake	<	.25 mg	3/L <	.50	mg/L	4.61	mg/L	<	.10	mg/L	<	.25	mg/L	

Commentaires/ Comments: By ICP , except Fe+3: by colorimetry. Anions by IC.

Effectué par/ Work by : L. Lavoie B. Legault J. Groleau D. Thóriault

D. Thériault

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To : M.Woyshner

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PROJET / PROJECT: V21 T03 100

Ref.: T03001-5-53 Date: 10/03/94

Lab #	I.D.	Description		Zn		C	Cl			N(NO2))	SO	S04	
13960	941784	Fault Lake				6.20	mg/L	<	.12	mg/L		2.74	mg/L	216.00	mg/L	
13961	941785	Fault Lake				9.60	mg/L	<	.12	mg/L		5.26	mg/L	319.00	mg/L	
13962	941786	Fault Lake				14.70	mg/L	<	.12	mg/L		1.75	mg/L	42.60	mg/L	
13963	941787	Fault Lake				2.22	mg/L	<	.12	mg/L		1.20	mg/L	30.30	mg/L	
13964	941788	Fault Lake				24.10	mg/L	<	.12	mg/L	<	.05	mg/L	534.00	mg/L	
13965	941789	Fault Lake				2.81	mg/L	<	.12	mg/L		1.26	mg/L	303.00	mg/L	
13966	941790	Fault Lake				20.40	mg/L	<	.12	mg/L	<	.05	mg/L	190.00	mg/L	
13967	941791	Fault Lake				44.20	mg/L	<	.12	mg/L	<	.05	mg/L	57.20	mg/L	
13968	941792	Fault Lake				39.80	mg/L	<	.12	mg∕L	<	.05	mg/L	35.60	mg/L	
13969	941793	Fault Lake				5.71	mg/L	<	.12	mg/L		.99	mg/L	112.00	mg/L	
13970	941794	Fault Lake				4.95	mg/L	<	.12	mg/L	<	.05	mg/L	120.00	mg/L	
13971	941795	Fault Lake				6.92	mg/L	<	.12	mg/L		1.34	mg/L	191.00	mg/L	
13972	941796	Fault Lake				36.90	mg/L	<	.12	mg/L		2.99	mg/L	37.10	mg/L	
13973	941797	Fault Lake				11.80	mg/L	<	.12	mg/L		2.97	mg/L	28.40	mg/L	
13974	941798	Fault Lake				11.70	mg/L	<	.12	mg/L		2.06	mg/L	29.50	mg/L	
13975	941799	Fault Lake				2.47	mg/L	<	.12	mg/L	<	.05	mg/L	12.30	mg/L	
13976	941800	Fault Lake				1.97	mg/L	<	.12	mg/L		3.25	mg/L	19.30	mg/L	
13977	941801	Fault Lake				3.57	mg/L	<	.12	mg/L		.24	mg/L	276.00	mg/L	
13978	941802	Fault Lake				2.41	mg/L	<	.12	mg/L		.75	mg/L	17.50	mg/L	
13979	941803	Fault Lake				39.90	mg/L	<	.12	mg/L	<	.05	mg/L	550.00	mg/L	
13980	941804	Fault Lake	<	.03	mg/L											
13981	941805	Fault Lake	<	.03	mg/L											
13982	941806	Fault Lake	<	.03	mg/L											
13983	941807	Fault Lake	<	.03	mg/L											
13984	941808	Fault Lake	<	.03	mg/L											
13985	941809	Fault Lake	<	.03	mg/L											
13986	941810	Fault Lake	<	.03	mg/L											
13987	941811	Fault Lake	<	.03	mg/L											
13988	941812	Fault Lake	<	.03	mg/L											
13989	941813	Fault Lake	<	.03	mg/L											
13990	941814	Fault Lake	<	.03	mg/L											
13991	941815	Fault Lake	<	.03	mg/L											
13992	941816	Fault Lake	<	.03	mg/L											
13993	941817	Fault Lake		31.13	ug/L											
13994	941818	Fault Lake	<	.03	mg/L											
13995	941819	Fault Lake	<	.03	mg/L											
13996	941820	Fault Lake		. 18	mg/L											
13997	941821	Fault Lake	<	.03	mg/L											
13998	941822	Fault Lake	<	.03	mg/L											
13999	941823	Fault Lake	<	.03	mg/L											

Commentaires/ Comments: By ICP , except Fe+3: by colorimetry. Anions by IC.

Effectué par/ Work by : L. Lavoie J. Groleau B. Legault D. Thériault

CERTIFICAT D'ANALYSE/ CERTIFICATE OF ANALYSIS

A/To : M.Woyshner

PROJET / PROJECT: V21 T03 100

Ref.: T03001-5-53 Date: 10/03/94

Lab #	I.D.	Description		Fe+3				F				 	
13960	941784	Fault Lake				<	.20	mg/L	 -				
13961	941785	Fault Lake				< '	.20	mg/L					
13962	941786	Fault Lake				<	.20	mg/L					
13963	941787	Fault Lake				<	.20	mg/L					
13964	941788	Fault Lake				<	.20	mg/L					
13965	941789	Fault Lake				<	.20	mg/L					
13966	941790	Fault Lake				<	.20	mg/L					
13967	941791	Fault Lake				<	.20	mg∕L					
13968	941792	Fault Lake				<	.20	mg/L					
13969	941793	Fault Lake				<	.20	mg/L					
13970	941794	Fault Lake				<	.20	mg/L					
13971	941795	Fault Lake				<	.20	mg/L					
13972	941796	Fault Lake				<	.20	mg/L					
13973	941797	Fault Lake				<	.20	mg/L					
13974	941798	Fault Lake				<	.20	mg/L					
13975	941799	Fault Lake				<	.20	mg/L					
13976	941800	Fault Lake				<	.20	mg/L					
13977	941801	Fault Lake				<	.20	mg/L					
13978	941802	Fault Lake				< l	.20	mg/L					
13979	941803	Fault Lake				<	.20	mg/L					
13980	941804	Fault Lake	<	.15	mg/L								
13981	941805	Fault Lake		. 15	mg/L								
13982	941806	Fault Lake	<	.15	mg/L								
13983	941807	Fault Lake	<	. 15	mg/L								
13984	941808	Fault Lake		.24	mg/L								
13985	941809	Fault Lake	<	. 15	mg/L								
13986	941810	Fault Lake	<	.15	mg/L								
13987	941811	Fault Lake		. 16	mg/L								
13988	941812	Fault Lake	<	. 15	mg/L								
13989	941813	Fault Lake		.38	mg/L								
13990	941814	Fault Lake		.47	mg/L								
13991	941815	Fault Lake		.21	mg∕L								
13992	941816	Fault Lake		.28	mg/L						-		
13993	941817	Fault Lake	< .	.15	mg/L								
13994	941818	Fault Lake	<	.15	mg/L								
13995	941819	Fault Lake		.36	mg/L								
13996	941820	Fault Lake		.28	mg/L								
13997	941821	Fault Lake	<	.15	mg/L								
13998	941822	Fault Lake	<	.15	mg/L								
13999	941823	Fault Lake	<	. 15	mg/L					•			

Commentaires/ Comments: By ICP , except Fe+3: by colorimetry. Anions by IC.

APPENDIX D

Piezocone data taken near monitoring station FS15 (UBC).



APPENDIX E

Investigation of the porous envelope effect at the Fault Lake tailing site, Part 1, draft report.

INVESTIGATION OF THE POROUS ENVELOPE EFFECT

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AT THE FAULT LAKE TAILINGS SITE

PRELIMINARY DRAFT REPORT

submitted to:

Falconbridge Limited Sudbury Operations

and

The Ontario Ministry of Northern Development and Mines

by

Noranda Technology Centre Mineral Sciences Laboratory Environmental Program

September 1993

Executive Summary

A porous envelope effect may occur in groundwater systems when mine tailings of low permeability are placed within high permeability soils. If the permeability contrast between the tailings and the natural soil is large, groundwater will flow around the tailings mass rather than through it, and metal leaching may be small.

The present hydrogeological study suggests that conditions for porous envelope containment may be occurring at the Falconbridge Fault Lake tailings site. The tailings have been deposited in a kettle lake formed within glacial outwash sand and gravel.

Water quality sampling in monitoring wells outside the tailings did not show any evidence of above-background metal concentrations, which suggests that leaching of metals from the tailings would be minimal. This is also suggested by the results of water sampling in nearby lakes, and by groundwater flow models.

Factors which contribute to limit metal concentrations downgradient of the Fault Lake tailings are:

- high hydraulic conductivity contrast between the tailings and the surrounding sediments,
- low position of the water table relative to the tailings bottom,
- limited infiltration through the surface of the tailings,
- dilution of metals flushed from the tailings by water flowing around and below the tailings,
- chemical attenuation of metals in the tailings and overburden

These factors could probably be present at other locations near mine sites. Tailings deposition could be done at these sites with little effect on groundwater quality pending that thorough site evaluations are performed and that appropriate control is done at the time of deposition.

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1. INTRODUCTION

- 1.1 Background
- 1.2 Surficial geology

2. INVESTIGATIVE METHODOLOGY

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1. INTRODUCTION

In 1992, Noranda Technology Centre (NTC) undertook a hydrogeological investigation of the Fault Lake tailings site. The site is unique in that, theoretically, a "porous envelope effect" may occur. If this is the case, flow through the tailings mass is low enough, relative to the surrounding more permeable till, that impact to the groundwater by tailings oxidation is insignificant at the regional scale.

The specific objectives of the investigation were to analyze the chemical and physical hydrogeology of the site, to delineate areas affected by acid mine drainage (AMD) generated from the tailings, and to verify the presence of the porous envelope effect.

1.1 Background

The Fault Lake tailings site is located northwest of the Falconbridge Sudbury operations, approximately 3 km north of Falconbridge and 0.5 km east of the Sudbury Airport (Fig. 1). The tailings were deposited between 1965 and 1978 and were produced from the milling of nickel ore in the Sudbury area. Approximately 6.45 million tonnes of tailings containing as much as 50% pyrrhotite were deposited in a depression of a maximum depth of approximately 30 m. The tailings were contained by dams to the north and south of the site. The deposit has an approximate volume of $3.36 \times 10^6 \text{ m}^3$ and a surface area of 22.2 ha (55 acre). It sits in a 55 ha (136 acre) closed watershed (Fig. 1a).

During the spring and fall, ponding occurs at the north dam, south dam and various berms. The water slowly infiltrates into the tailings and evaporates from the ponds. During the summer months, extensive ponding has not been observed. Tailings in areas where ponding has occurred are soft and grey, while the rest of the tailings are hard and crusty, showing orange traces of oxidation.

In 1971, while deposition was active, the analysis of groundwater in one well located 2 km downgradient (northeast) of the Fault Lake tailings site indicated abovebackground sulphate levels of 382 mg/L, suggesting influence from tailings oxidation International Water Supply, 1971). Groundwater and surface water monitoring and analysis done at later dates, though, showed improvements in water quality and suggested that impact to groundwater by tailings oxidation would have stopped.

1.2 Surficial Geology

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Overburden thickness varies within the studied area, from 36 m to more than 60 m. Overburden mainly consists of coarse to fine glacial outwash sands and gravels with some large boulders and silt lenses.

Kettles, fluvial terraces, discontinuous crevasse fillings, and eskers within the Fault Lake tailings area are evidence of a glacial meltwater channel, partly choked with stranded ice blocks. The small round kettle lakes were formed after the late melting of the stranded ice blocks which were caught among the mass of glacial sediments. The sediment are assembled in longitudinal formations which follow a northeasterly direction, eventually leading into Bowlands Bay, part of Lake Wanipitie. Figure 2 shows the main overburden materials and their orientation in the area of the site.

2. INVESTIGATIVE METHODOLOGY

2.1 Installation of Groundwater Monitoring Stations

The routes by which acid water could be transported from the Fault Lake tailings site were identified by field observations and supported by geophysical data prior to the drilling of the monitoring wells (Geomar, 1991, 1992). Probable seepage routes were identified leaving the tailings site at the base of the north and south dams.

Thirteen groundwater monitoring stations were located to sample the groundwater in the sediments directly below the tailings deposit and along the seepage routes. In addition, one station (FS-2) was located upgradient of the tailings to characterize background conditions. Figure 3 shows the locations of all the stations and also shows the outline of the original kettle lake as determined from aerial photographs taken prior to tailings deposition. Bedrock in the vicinity of the tailings site was not instrumented as it was not believed to have an important influence on groundwater flow in the area.

Drilling of the groundwater monitoring stations at the Fault Lake tailings site was conducted using a 15 cm ID hollow stem auger. A total of 14 holes were drilled in the overburden between September 15 and November 8, 1992. Diamond core attachments were required where boulders were encountered. At stations where the water table was deep below the surface, water injection was required in order to prevent excess friction on the drill stem. The water served as a lubricant and coolant, as a means for removing drill cuttings, and as a way to clean the holes after drilling was completed. The water was supplied from the Falconbridge mill, and was modified by the addition of rhodamine-D, a non-adsorbing photo-degradable tracer. This enabled the possibility of recognizing the presence of drilling water which could have remained in the boreholes at the time of groundwater sampling. Samples of the overburden and tailings were obtained using a split spoon sampler. Grain size analysis was conducted on most of the samples collected and the results are displayed in Appendix A.

Piezometers / monitoring wells installed at all the stations of the Fault Lake Tailings site are 1.9 cm (0.75 inch) ID, schedule 80 PVC pipe with a 0.3 m (1 ft) PVC screened tip. These were installed by placing the pipe inside the hollow stem auger at the required depth. The auger was then raised approximately 1.5 m at which time clean silica sand was packed around the PVC screen. A bentonite seal was placed above the sand to insure hydraulic isolation of the well. Bentonite seals and steel

casings equipped with locking covers were also installed at the surface to protect the wells and prevent infiltration of water from surface.

After installation, all monitoring wells were labelled and surveyed. The labels were numbered according to station location and depth (e.g., at location FS-1, well FS-1-A is deeper than well FS-1-B). Boreholes logs and corresponding well tip locations and station coordinates appear in Appendix B.

Each monitoring well was purged of initial drilling water. Water in the well was allowed to re-equilibrate and level readings were then obtained using an electrical water level meter. Water level data is also included in Appendix B.

2.2 Water Sampling and Analysis

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Groundwater was sampled from monitoring wells in December 1992 and in March 1993. Before sampling, the depth to water level was measured and three well volumes were purged to eliminate standing water and drill water. After the water had been sufficiently recovered, depth to water level was re-measured and water samples collected.

The samples were collected using a peristaltic pump, a nitrogen-driven positive displacement pump, or the Waterra system. The groundwater samples were filtered using a 0.45 μ m (ACRO 50A) disposable in-line filter. Field measurements of pH, temperature, oxidation reduction potential (Eh), and electrical conductance were recorded. Half of each sample was acidified in the field using reagent grade (2% v/v) hydrochloric acid (HCI) for metal preservation prior to analysis. All electrodes were calibrated before use and between samples. All sampling equipment was rinsed with distilled water before each sample was collected.

Water samples were transported to a field laboratory within six hours from collection. In the laboratory, measurements of pH were repeated on the non-acidified portion of the samples along with titration for acidity and alkalinity.

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In the NTC analytical laboratory, the acidified portion of each sample was analyzed for dissolved metal and major ions and the non-acidified portion was analyzed for chloride. Potassium (K) was analyzed using flame atomic emission, ferrous iron (Fe²⁺) by colorimetry/volumetry, chloride (Cl⁻) by turbidimetry, and all other elements by inductively coupled plasma spectrophotometry (ICP). All certificates of analysis appear in Appendix C.

Tailings pore water was sampled by squeezing tailings samples using a pneumatic squeeze apparatus at 0.8 MPa in a stainless steel loading cell. Due to the low water contents of the samples, it was required to saturate the samples with distilled water in order to retrieve enough volume for analysis (saturation extract procedure). Initial and final moisture contents were determined to calculate a dilution factor, which was used to correct the metal concentrations determined on the extract solution by the same analytical methods as described above.

Quality testing for all sample batches was performed using replicate and standard samples. The samples were collected to evaluate reproducibility and accuracy of the analytical procedure and to assess the cleanliness of the equipment during sampling. Coefficient of variation and relative error for the samples selected were low, indicating good quality data. The results of the quality assurance testing are included in Appendix C.

Five kettle lakes in the Fault Lake tailings area (Fig. 3) were sampled in October 1992 by Falconbridge personnel. Samples were collected at selected depths from each of the ponds and analyzed using the same procedure as for the groundwater samples. The five ponds were numbered according to Fig.3.

2.3 Hydraulic Conductivity Measurements

Measurements of *in-situ* hydraulic conductivity were conducted at most of the monitoring stations using the "falling head test". The test is performed below the water table. An instantaneous water level rise in the piezometer is induced with a slug and water level recovery is recorded with time. Water level recovery was monitored either manually, using a watch and water level indicator, or automatically, using a Shape SH3500 submersible pressure transducer (which also served as a slug) and a Lakewood meter/logger.

Interpretation of the water level versus time data was conducted using the Hvorslev (1951) method for point piezometers. As described in Freeze and Cherry (1979), hydraulic conductivity (K) is determined using the following equation:

$$K = \frac{r^2 \ln \frac{L}{R}}{2LT_o}$$

(1)

where, T_{o} is the time lag or time that would be required for the complete equalization of the head differences if the original rate of inflow were maintained, **L** is the length of the piezometer intake or screen, and **R** is the radius of the piezometer, and **r** is the radius of the borehole.

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3. RESULTS

3.1 Physical Hydrogeology

Measurements indicated that, in general, the water table is 0 to 2 m below the base of the tailings, except at station FS-6 where the water table is within the tailings (about 8 m above the base of the tailings). At that station, the water level may be high due to surface ponding or perched on fine-grained material. At station FS-12, which is the closest to the centre of the original kettle lake, the measured water level was approximately at the same level as the base of the tailings may be slightly deeper than the water table in the small area of the tailings between stations FS-12 and FS-5. The fact that most of the tailings are above the water table is surprising because water was visible in the kettle lake before tailings deposition. Reasons for this apparent water level change are discussed further.

A water table contour map (Fig. 4) was constructed for the Fault Lake tailings area by using the data collected in November 1992 and March 1993 to show average groundwater conditions. Since groundwater flow in hummocky terrain is generally controlled by topography, the water table contours reflect topographic contours. Regional groundwater flow from the tailings watershed is northeast toward the small kettle lakes and southeast toward the new tailings area, as illustrated by the arrows on Fig.4.

Hydraulic gradients were calculated for both vertical and horizontal directions. Vertical gradients were near zero at most of the monitoring stations surrounding the tailings deposit. Beneath the tailings deposit, vertical gradients were significant and indicate vertical percolation. At stations FS-4, FS-5 and FS-6 vertical gradients were respectively 2.5, 0.7 and 1.0. At FS-5, an upward gradient also appears, which, when

coupled with the high (>1) gradient at FS-4, suggests partially confined conditions and/or localized flow paths below the tailings (likely from the region around FS-6). Horizontal gradients were very small. At the north dam, between stations FS-6 and FS-8 the horizontal gradient was 0.0002. The horizontal gradient at the south dam, between monitoring station FS-1 and FS-13, was 0.007.

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Table 1 shows the results of the hydraulic conductivity tests for each piezometer. Measured hydraulic conductivities in the natural overburden units were highly variable, ranging between 8x10⁻¹ cm/s (at FS-10) and 2.5x10⁻⁵ cm/s (at FS-4). The large variations in hydraulic conductivity are explained by the variability in soil types typical of ice-contact deposits which include silts, sands, gravels, and boulders. The higher values of hydraulic conductivity (such as at FS-10 and FS-14) would occur where fast meltwater flows would have formed accumulations of well-sorted sands and gravels. The lower hydraulic conductivities occur where glacial abrasion and slow meltwater flows would have left finer silts. The hydraulic conductivity values also suggest that silts may be present within void spaces between boulders (at FS-4, for example).

The geometric average of all hydraulic conductivity measurements is 1.6×10^{-3} cm/s. This value would be representative of a clean to silty medium sand, and is considered to be representative of the overall effective hydraulic conductivity of the ice-contact deposits in which lie the tailings.

The average linear groundwater velocity (v) in the overburden north of the tailings can be estimated by the Darcy equation:

$$r = K i / n \tag{2}$$

Using the average hydraulic conductivity (K) of 1.6x10⁻³ cm/s, a hydraulic gradient (i) of 0.0002, and an estimated porosity (n) of 0.30, the calculated velocity north of the tailings is approximately 30 cm/year. With a horizontal gradient of 0.007, the

calculated average darcy velocity south of the tailings is 6 m/year. This velocity is only approximate, and actual local velocities may vary by a factor of 10.

The hydraulic conductivity of the tailings could not be measured with the same field techniques used for measurements in the natural overburden because the tailings were above the water-table. The hydraulic conductivity of the tailings was therefore estimated using the *Kozeny-Carman* equation (Bear, 1972):

$$K = \frac{d_m^2}{180} \frac{n^3}{(1-n)^2}$$
(3)

where d_m is a representative particle diameter and n is the soil porosity. Table 2 shows the results of these calculations using an estimated porosity of 0.45 and the median particle diameter, or d_{50} . The resulting estimated tailings hydraulic conductivities averaged 1.2×10^{-5} cm/s, which is consistent with measurements at other sites (Yanful and St-Arnaud, 1991, for example). For comparative purposes, calculations of tailings hydraulic conductivities using various porosities and particle diameters are presented in Appendix D. The measured hydraulic conductivities of tailings and overburden materials also agree with previous estimates done by Geocon (1985).

3.2 Chemical Hydrogeology

Results of the analysis of tailings pore water, groundwater, and water from surface bodies are shown in Tables 3 through 8. Concentrations of nickel, sulphate, and total iron characterize the general water quality found at the site and are discussed in most detail. Other physico-chemical parameters and chemical concentrations were determined and are listed in the Tables.

Water extractions were performed on 6 selected samples collected in November 1992 at each of the monitoring stations located on the tailings (Table 3). Nickel

concentrations varied from 4 mg/L (FS-11 at 6.1 m depth) to 644 mg/L (FS-4 at 6.1 m depth). Sulphate concentrations varied from 3041 mg/L (FS-11 at 6.1 m depth) to more than 84 g/L (84,600 mg/L at FS-4). Total iron concentrations were between 0.5 mg/L to 466 mg/L. These values indicate the presence of high metal concentrations due to sulphide oxidation within portions of the tailings deposit. Metal concentrations in the pore water are strongly influenced by downward water movement and chemical precipitation and dissolution reactions which occur in the tailings to concentrations of 5 to 8 mg/L (FS-12, depth 16 m; FS 13, depth 9 m in Table 3). Thermodynamic calculations done on porewater quality data from FS-4 and FS-6 using the MINTEQ program (Felmy et. al, 1984) suggested that the pore water could be near saturation with respect to nickel sulphate mineral species.

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Variability in measured metal concentrations could also be caused by variations in the intensity of oxidation across the surface of the tailings. Visual inspection of the tailings shows the development of cracks and crusty layers at the surface which could locally influence water and oxygen entry and the resulting production of acid. Thorough investigations of the geochemical sources and evolution of metal concentrations have been investigated for other sulphide tailings sites (Blowes et. al, 1988, for example) and for the Falconbridge main pyrrhotite tailings site adjacent to the Fault Lake site (Nicholson and David, 1991). The investigation of the geochemistry of the Fault Lake tailings was not part of the objectives of the present study, and was therefore not pursued further than described above.

Background groundwater monitoring station FS-2 showed a pH near 7 and nickel concentrations near 0.01 mg/L, iron near 0.03 mg/L, and sulphate near 30 mg/L (Tables 4 to 7). Although nickel levels at FS-2 were slightly lower in the second sampling round (<0.005 mg/L), iron and sulphate values were the same as in the first round. The metal concentrations measured in the first sampling round at FS-2 can thus be accepted as background concentrations for groundwater at the site.

Background pH could be lower than that measured at FS-2 (as low as 6) due, for example, to the infiltration of acidic rainwater.

Groundwater sampled around the tailings site in December 1992 and March 1993 had pH values above 6 (Tables 4 to 7). Above-background concentrations of nickel were measured in wells FS-3A, FS-3B, FS-9C, and FS-10B; the highest of these concentrations was 0.5 mg/L (at FS-10), and was measured during only one of the sampling rounds. Only at FS-3 were the higher nickel levels associated with abovebackground sulphate concentrations of near 240 mg/L. Above-background sulphate concentrations were also encountered at station FS-1 (max 339 mg/L), but were not associated with any above-background metal concentrations.

Sampling results suggest that metal concentrations are not high enough to affect groundwater quality. This is also suggested by the results of surface water quality sampling (Table 8), which do not show the presence of any metal above background concentrations.

The presence of the tracer-labelled drilling water was encountered during both of the sampling rounds at some of the monitoring stations. Wells FS-3C, -6B, -6C, -14A, - 14B were therefore not sampled. Samples should be obtained from these stations at a later date, after all the drill water evacuates. Monitoring at all stations should also be pursued.

4.0 GROUNDWATER FLOW MODELLING

4.1 Modelling Procedure

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Groundwater flow around the tailings was simulated using the saturated-unsaturated flow modelling program SEEP/W (GEO-SLOPE International, 1992). SEEP/W requires the definition of a domain (finite-element grid), of soil hydraulic conductivities, and of boundary conditions to determine the flownet.

Two models were defined in order to obtain a quasi-3-dimensional perspective of the groundwater flows in the Fault Lake tailings area. Model 1 was a plan model, and was defined as a rectangle with the long edges parallel to the main flow direction inferred from the equipotential map of Fig.4. Model 1 was conceptual and represented the flow of water directly beneath the soil surface as affected by the hydraulic conductivity contrast between the tailings and the surrounding sediments. Model 1 did not incorporate the effect of hydraulic potential variations due to topographical effects.

Model 2 was a cross-section across stations FS-4 to FS-9 (**A-A'** in Fig 3). The model domain started 305 m (1000 ft) southwest of FS-4, extended 105 m (344 ft) northeast of FS-9 and passed through FS-6 and FS-8. Surface elevations were taken from the monitoring well data where possible; otherwise, the topographic map was used. Tailings and bedrock depths were determined using drilling data. Model 2 was extended 5 m into the bedrock by assuming that the top part of the bedrock was fractured and was not hydraulically isolated from the overburden.

Representative hydraulic conductivities (K) for the soils and tailings were derived from the geometric mean of the field measurements, as described in section 3.1. The modelling also required the input of soil characteristic functions describing the decrease in hydraulic conductivity with water content in the unsaturated zone. These

characteristic functions were not specifically determined for the Fault Lake soils. Instead, functions for soils similar to those at the Fault Lake site reported in Yanful at. al (1993) were used and were considered accurate enough for the present study.

Boundary Conditions

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Constant head boundary conditions were defined at both ends of the model domains. In Model 1, the constant heads were set equal to the elevation of the nearest lakes: lake #1 (306.7 m) in the south and lake #3 (297.6 m) in the north. The elevations of these lakes are marked on the topographic map (Fig 1) and the numbering of the lakes is marked in Figure 3. Since Model 1 simulated horizontal flow only, precipitation and evaporation effects were not considered.

In Model 2, a water table elevation slightly higher than that measured at FS-4 was used at **A** (300 m), and for **A'**, the elevation of lake #2 was taken off the topographic map (297.6 m). Two top boundary conditions were used: with infiltration (Model 2A) and without infiltration (Model 2B). In Model 2A, the infiltration flux across the top boundary was determined using previous estimates from Yanful et. al (1993) obtained using the HELP (Hydrologic Evaluation of Landfill Performance) computer program of Shroeder et. al (1984). HELP is a deterministic water balance program which uses climatic, soil and design data to calculate infiltration. Different fluxes were used in the SEEP/W modelling depending on the slope and nature of the ground surface. The tailings surface infiltration was set to 200 mm/year, sloped till 250 mm/year and flat till 350 mm/year. It was determined that infiltration into the till would be higher than that into the tailings since the hydraulic conductivity is higher and the water table is low. This would cause precipitation to be absorbed by the till and transferred away from the surface (to the water table), thereby limiting evaporation. On the tailings, evaporation is enhanced by surface ponding and infiltration is reduced compared to the till.

4.2 Modelling Results

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Flow vectors for Model 1 describing groundwater flow without topographical effects are shown in Figure 5(A). The flow vectors represent the direction of flow, with their length being proportional to the flow quantities. Figure 5(A) clearly demonstrates that the water flows mainly in the till and around the tailings because of the higher hydraulic conductivity of the till. The hydraulic gradients (in m of water) are illustrated in Figure 5(B). Higher gradients (closer contours) developed across the tailings, as opposed to around the tailings, because of the restriction to flow.

Figure 6 depicts the flow characteristics for Model 2A (section with infiltration). The figure indicates that the horizontal flow is predominantly in the till and does not enter the tailings. In addition, all water that infiltrates the surface of the tailings flows vertically through the tailings and to the right (northeast). Therefore, water that flows through the tailings is only that water which infiltrates into the tailings from precipitation. The model also shows that the water infiltrating into the till left of the tailings flows to the left (west). In this case where the hydraulic gradient and flow velocities are low, the direction of flow in the west side of the section may be an artifact of the model. In any case, the hydraulic gradient and flow velocities generally agree with the water table contour map (Fig. 4), and suggest that flow from the west is not likely to enter the tailings.

Figure 7 displays the results of Model 2B (without infiltration). As in Model 1 (plan view), water flow bypasses the tailings. Essentially, no water flows into or out of the tailings; the calculated fraction of water flowing through the tailings, as opposed to around it, is less than 0.4%. The head contours below the tailings show that an increased gradient exists in that part of the flow section. This is simply due to the smaller cross-sectional area available in the flow domain. A dilution of contaminated water leaking from the tailings proportional to the groundwater discharge rate is

expected to occur in that part of the flow section. Higher gradients are also evident at the right of the grid where the cross-sectional area of the till is again reduced.

To simulate conditions that would occur if the water table was to rise into the tailings and saturate the bottom portion of the impoundment, the constant head boundary functions at either end of the section were modified (Model 2C). The results of the simulation, which includes infiltration, are displayed in Figure 8. Although the bottom 5 metres of the tailings are saturated, the flow vectors within the saturated tailings are insignificantly small, as in Model 2A. The result shows that the flow is predominantly below the impoundment and suggests that the increase in metal loading in the groundwater due to the water table rise could not be significant.

The use of a 2-D flow model to assess the conditions on the Fault Lake site has inherent limitations. In particular, any quantification of flow volumes is affected by the fact that all the water is forced to move within the 2-D reference plane, while, in reality, water also moves across the plane. This would reduce the porous envelope effect in the model compared to reality. Another limitation is that, in reality, metal velocities usually slower than water velocities, and metal concentrations usually decrease as the water moves through the tailings and soils. This chemical attenuation is not taken into account by the model, so model predictions based only on groundwater flow can lead to overestimation of chemical loadings. The flow model is however very useful for predicting the worse-case scenario where no chemical attenuation would take place. Predictions which account for chemical attenuation are complex and are not part of the objectives of the present study.

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5. DISCUSSION

The glacial sediments surrounding the Fault Lake tailings site are characterized by their elongated formation and relatively high bulk hydraulic conductivities. This creates a flow system with a relatively flat and deep water table. Several favourable factors contribute to limit the observed metal concentrations downgradient of the tailings: (1) the hydraulic conductivity contrast between the tailings and the surrounding sediments, as described conceptually by Model 1 in the previous section; (2) the low position of the water table relative to the tailings bottom, which has the effect of limiting the volume of tailings in saturated conditions, as described in Model 2; (3) the limited infiltration through the surface of the tailings, (4) the dilution of metals flushed from the tailings by water flowing around and below the tailings; (5) chemical attenuation of metals, which probably plays a large role both inside the tailings mass and in the surrounding sediments.

All the factors outlined above have a chance to occur simultaneously at other sites. The probability of occurrence for each of the factors are as follows:

(1) hydraulic conductivity contrast:

Sediments of high hydraulic conductivities such as sands and gravels do occur commonly in Canada. The ice-contact deposits surrounding the Fault Lake area possess a high average hydraulic conductivity, but are also characterized by a high variability due to the process by which they were deposited which produced a mix of particle sizes. Other sand and gravel units may be more uniform and have less variability in hydraulic properties; bulk hydraulic conductivities may be higher (near 10^{-2} cm/s, for example).

As for tailings, measurements at several sites (for example by Blowes et. al, 1988; Yanful and St-Arnaud, 1992) have yielded values close to 10⁻⁵ cm/s, as measured at the Fault Lake tailings. Large hydraulic conductivity contrasts between tailings and surrounding sediments are therefore commonly possible.

(2) Deep water table

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The water table in well-drained sand and gravel formations in temperate climates will usually be deep below the ground surface. The depth to water table is also largely affected by topographical features. The hummocky terrain surrounding the Fault Lake site is largely controlled by the occurrence of the kettle lakes. In other similar glacial outwash areas, kettle lakes may not occur; however, depending on bedrock topography, the coarse glacial deposits are commonly elevated and produce deep water tables.

At the Fault Lake site, the water table was expected to be within the tailings because the area of deposition was once a kettle lake. The observed water table, however, was lower, probably because of man-made changes in the watershed. One hypothesis is that the tailings deposit may divert hillside runoff which had once entered the lake, to the edges of the tailings where it infiltrates into the till. Once within the till, the water is less likely to flow into the tailings. Therefore, as illustrated in Model 2 in the previous section, only the precipitation that infiltrates into the tailings surface will reach the water table below the tailings. This has the effect of reducing the watershed supplying the water table below the tailings (or the original lake) from 55 ha to 22 ha (the surface area of the tailings). Other changes to watersheds due to quarry excavation may also contribute to the change in water level.

(3) limited infiltration

Infiltration of water through the surface of the tailings surfaces is usually less than through most natural soils. This is due to the relatively low bulk hydraulic conductivity of the tailings, the high potential for evaporation at the tailings surface, and the formation of dense crusts at the tailings surface which can further reduce the tailings conductivity. Infiltration is promoted by water ponding on the tailings surface. At the Fault Lake tailings site, infiltration could be reduced compared to present conditions by preventing the ponding of water along the dams.

(4) dilution by regional groundwater flow

Large glacial outwash sediment formations are propitious to high groundwater discharges which are less susceptible to degradation by point contamination sources. In the case of the Fault Lake site, results from Model 2 suggest that flow upgradient of the tailings site would not be very large. The occurrence of much larger regional flow systems at other sites is possible.

(5) chemical attenuation

Some amount of chemical attenuation in the form of precipitation and adsorption reactions occur in most tailings. The degree at which these reactions take place depends on geochemical and mineralogical factors, in particular those which influence the neutralisation potential of the tailings. Chemical attenuation in the Fault Lake tailings is suggested by the tailings pore water data, and could also occur within the natural overburden deposits.
All five of the factors outlined above would contribute to create the porous envelope effect; these factors could probably be present at other locations near mine sites. Tailings deposition in these types of environments could be done with little effect on groundwater quality, pending that thorough site evaluations are performed and that appropriate control is done at the time of deposition.

5. CONCLUSIONS

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- 1) The piezometric elevations throughout the Fault Lake site, combined with lake elevations, suggest that the regional direction of subsurface flow is toward the northeast, along with the alignment of kettle lakes. Some sub-regional flow systems could be moving groundwater in other directions.
- 2) The base of the tailings are at the same level or higher than the water table across most of the site. This is surprising because water was visible in the kettle lake before tailings deposition. Low water infiltration in the tailings and changes in watershed configurations due to nearby quarry excavation are suggested as causes for the apparent water level change.
- 3) The average bulk hydraulic conductivity of the glacial outwash soil material surrounding the tailings is estimated at 1.6 x 10⁻³ cm/s. The hydraulic conductivity of the tailings was estimated using grain size correlations at 1 x 10⁻⁵ cm/s. These values agree with previous estimates.
- 4) Analysis of tailings pore water showed elevated values of nickel, iron, and sulphate indicating the presence of sulphide oxidation products within portions of the tailings deposit. Metal concentrations are attenuated in the deeper parts of the tailings. Apparent high variability in measured metal concentrations could

be caused by variations in the intensity of oxidation across the surface of the tailings due to surface effects such as drying and cracking.

- 5) Water quality sampling in monitoring wells outside the tailings did not show any evidence of above-background metal concentrations, which suggests that leaching of metals from the tailings would be minimal. This is also suggested by the results of water sampling in nearby lakes.
- 6) Two-dimensional groundwater flow models showed that groundwater flow is diverted around the tailings mass due to the hydraulic conductivity contrast between the tailings and the surrounding sediments. The models also showed that flushing of the tailings mass by groundwater should not contribute significantly to the regional groundwater flow system under present water table conditions, as well as under conditions of moderate rise in water table level.
- Factors which contribute to limit metal concentrations downgradient of the Fault Lake tailings are:
 - the large hydraulic conductivity contrast between the tailings and the surrounding sediments,
 - the low position of the water table relative to the tailings bottom,
 - the limited infiltration through the surface of the tailings,

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- the dilution of metals flushed from the tailings by water flowing around and below the tailings
- chemical attenuation of metals in the tailings and overburden

These factors could probably be present at other locations near mine sites. Tailings deposition could be done at these sites with little effect on groundwater quality pending that thorough site evaluations are performed and that appropriate control is done at the time of deposition.

CLOSURE

Field work and preliminary data analysis for this work were performed by S. Aiken. Numerical modelling was done by B. Aubé. M. Woyshner and P. Tibble assisted in the review and preparation of the final report. L. St-Arnaud coordinated the project and reviewed the final report.

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Piezometer	K
No.	(cm/s)
FS-1-A	No Test
FS-1-B	2.5E-03
FS-1-C	No Test
FS-2-A	2.9E-03
FS-2-B	2.9E-03
FS-3-A	1.0E-02
FS-3-B	8.6E-03
FS-3-C	No Water
FS-4-A	2.5E-05
FS-4-B	5.1E-05
FS-4-C	3.0E-05
FS-5-A	1.9E-04
FS-5-B	9.3E-05
FS-5-C	3.1E-04
FS-6-A	1.2E-04
FS-6-B	1.5E-04
FS-6-C	4.0E-04
FS-7	No Water
FS-8	6.1E-03
FS-9-A	8.9E-05
FS-9-B	7.6E-05
FS-9-C	8.1E-04
FS-10-A	8.0E-01
FS-10-B	8.0E-01
FS-11	3.6E-03 *
FS-12	1.0 E-04 *
FS-13	2.0E-02 *
FS-14-A	4.0E-02
FS-14-B	4.0E-02
* Estimate - Fro	m Grain Size Data

Table 1. Hydraulic Conductivity of Natural Soils

Table 2.Estimated Tailings ConductivitiesUsing Modified Kozeny-Carman

Sample	Depth (m)	d50 (mm)	K (cm/s)
FS-4 FS-4 FS-5 FS-5 FS-5 FS-5 FS-5	Surface 6.10 - 6.70 7.6 - 8.2 Surface 3.05 - 3.65 4.57 - 5.18 7 6 - 8 2	0.0185 0.044 0.017 0.058 0.071 0.0245 0.011	5.73E-07 3.24E-06 4.84E-07 5.63E-06 8.44E-06 1.00E-06 2.03E-07
FS-6 FS-6 FS-6 FS-6 FS-6 FS-6	Surface 3.05 - 3.65 4.5 - 5.2 6.1 - 6.71 7.62 - 8.07	0.017 0.024 0.0082 0.16 0.21	4.84E-07 9.64E-07 1.13E-07 4.28E-05 7.38E-05
FS-8	1.5 - 2.1	0.051	4.35E-06
FS-11 FS-11 FS-11 FS-11 FS-11 FS-11 FS-11 FS-11	1.5 - 2.1 6.1 - 6.71 7.6 - 8.2 9.1 - 9.8 15.2 - 15.8 16.8 - 17.4 18.3 - 18.9 19.8 - 20.4	0.04 0.12 0.021 10 6.4 0.91 0.48 0.18	2.68E-06 2.41E-05 7.38E-07 1.67E-01 6.85E-02 1.39E-03 3.86E-04 5.42E-05
FS-12 FS-12	16.76 - 17.37 19.81 - 20.42	0.12 0.18	2.41E-05 5.42E-05
FS-13	3.1 - 3.7	0.0195	6.36E-07

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Table 3.

Metal concentrations in tailings pore water

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Sample	Depth (m)	Al (mg/L)	As (mg/L)	Ca (mg/L)	Cd (mg/L)	Cu (mg/L)	Fe (mg/L)	K (mg/L)	Mg (mg/L)
FS-4 FS-6 FS-11 FS-12a FS-12b FS-13	6.1 - 6.7 6.1 - 6.7 6.1 - 6.7 16.8 - 17.4 19.8 - 20.4 9.1 - 9.8	3.88 0.95 <0.25 <0.25 <0.25 <0.25 <0.25	<0.25 <0.25 <0.25 <0.25 <0.25 <0.25 <0.25	8536.00 3197.70 890.50 1056.00 1316.00 1395.00	<0.02 0.09 <0.02 <0.02 <0.02 <0.02 <0.02	0.15 0.11 <0.02 0.04 <0.02 <0.02	465.60 109.77 0.79 0.42 0.41 0.52	1188.25 319.77 143.85 211.20 291.40 281.25	3773.30 602.14 173.99 226.80 248.16 879.75
Sample	Depth (m)	Mn (mg/L)	Na (mg/L)	Ni (mg/L)	Pb (mg/L)	SO4 (mg/L)	Se (mg/L)	Zn (mg/L)	
FS-4 FS-6 FS-11 FS-12a FS-12b FS-13	6.1 - 6.7 6.1 - 6.7 6.1 - 6.7 16.8 - 17.4 19.8 - 20.4 9.1 - 9.8	96.03 4.30 0.79 1.37 1.00 2.45	81.48 153.34 87.68 79.56 141.38 97.88	645.05 30.48 4.00 5.63 4.79 8.21	<0.25 <0.25 <0.25 <0.25 <0.25 <0.25 <0.25 <0.25	84681.00 29340.30 3041.40 7200.00 4230.00 6750.00	<0.50 <0.50 <0.50 <0.50 <0.50 <0.50 <0.50	3.69 3.22 0.10 0.11 0.38 0.27	

	Temp.	pН	pН	Eh	Conductivity	Acidity	Alkalinity
Sample	C	(field)	(lab)	(mv)	(uS/cm)	(mg/L CaCO3)	(mg/L CaCO3)
FS-1-A	8.1	7.15	7.10	542	470	<50	50
FS-1-B	8.7	8.12	7.95	526	580	<50	150
FS-1-C	6.6	7.96	7.90	521	212	<50	100
FS-2-A	4.5	7.94	7.90	432	188	<50	100
FS-2-B	5.5	7.12	7.01	413	209	<50	<50
FS-3-A	7.4	6.36	6.40	370	90	<50	<50
FS-3-B	<			Trace	r in Water	>	
FS-3-C	<			Trace	r in Water	>	
FS-4-A	<			Trace	r in Water	>	· · · ·
FS-4-B	<			Trace	r in Water	>	
FS-4-C	<			Trace	r in Water	>	
FS-5-A	<			Trace	r in Water	>	
FS-5-B	<	*****		Trace	r in Water	>	
FS-5-C	<			Trace	r in Water '	>	
FS-6-A	<			Trace	r in Water	>	
FS-6-B	<	الد فدخة في عن ال		Irace	r in Water	>	
FS-0-C	<	*****		Irace	r in Water	•••••••>	
FS-7	<		·	No	Water	>	
FS-8	7.3	6.02	5.91	486	150	<50	<50
FS-9-A	6.8	6.82	6.80	455	175	<50	<50
FS-9-B	7.8	6.75	6.70	472	254	<50	<50
FS-9-C	7.5	6.39	6.51	419	158	<50	<50
FS-10-A	6.3	6.52	6.40	409	149	<50	<50
FS-10-B	6.7	5.87	6.10	390	123	<50	<50
FS-11	<			No	Water	>	
FS-12	<			No	Water	>	
FS-13	<			No	Water	>	
FS-14-A	<			Trace	r in Water	>	
FS-14-B	<			Trace	r in Water	>	

Table 4. Physico-Chemical Parameter Values Fault Lake, December 1992

Table 5. Metal and Major Ion Concentrations in Groundwater Sampled Fault Lake Tailings, December 1992

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Sample	Al (mg/L)	As (mg/L)	Ca (mg/L)	Cd (ug/L)	Co (ug/L)	Cr (ug/L)	Cu (ug/L)	Fe (ug/L)	K (mg/L)	Mg (mg/L)	Mn (ug/L)	Na (mg/L)	Ni (ug/L)	Pb (mg/L)	S (mg/L)	Se (mg/L)	Sn (mg/L)	Zn (ug/L)	CI (mg/L)	NO3 (mg/L)	SO4 (mg/L)	Fe2+ (mg/L)
FS-1-A FS-1-B FS-1-C	0.07 0.05 0.07	<0.05 <0.05 <0.05	84.44 84.86 18.42	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	<5.00 <5.00 5.04	10.9 13.9 49.4	2.39 1.65 1.30	21.68 21.02 3.35	21.7 6.6 3.86	14.48 37.44 23.68	<5.00 <5.00 <5.00	<5.00 <5.00 0.07	73.74 85.69 11.82	<0.05 <0.05 <0.05	<0.05 <0.05 <0.05	<5.00 <5.00 <5.00	4.02 6.47 3.97	12.9 24.9 8.37	210 239 31,9	-
FS-2-A FS-2-B	<0.05 <0.05	<0.05 <0.05	33.68 21.54	<5.00 <5.00	<5.00 <5.00	<5.00 <5.00	<5.00 7.16	27.2 17.7	1.09 1.73	8.17 3.64	22.7 95.4	6.49 7.89	11.9 9.75	<5.00 <5.00	12.2 12.2	<0.05 <0.05	<0.05 <0.05	<5.00 <5.00	13.5 3.70	8.22 5.25	33.5 32:4	-
FS-3-A FS-3-B FS-3-C	0.08	<0.05 	9.52	<5.00 	<5.00	<5.00	0.08	140	<1.00	2.42	16.6	2.84	14.1	<5.00	4.63 > >	<0.05	<0.05	150	2.07	2.08	11.2	-
FS-4-A FS-4-B FS-4-C															> >					.		
FS-5-A FS-5-B FS-5-C		 													> >							
FS-6-A FS-6-B FS-6-C		 													> >			·				
FS-7	No Wate	ır				••••••									>							
FS-8	0.07	<0.05	11.66	<5.00	<5.00	<5.00	<5.00	1310	1.92	2.58	6310	6.95	6.68	<5.00	12.89	<0.05	<0.05	15.4	1.56	6.75	27.4	1.05
FS-9-A FS-9-B FS-9-C	0.08 0.08 0.08	<0.05 <0.05 <0.05	28.71 22.81 6.18	<5.00 <5.00 <5.00	<5.00 <5.00 7.23	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	23.2 42.5 55.4	1.84 2.45 1.20	3.83 3.33 1.11	73.9 120 1510	5.15 22.33 11.83	9.75 <5.00 110	<5.00 <5.00 <5.00	7.93 7.80 10.03	<0.05 <0.05 <0.05	<0.05 <0.05 <0.05	<5.00 <5.00 <5.00	5.20 9.51 3.06	2.30 2.38 2.61	16.6 20.7 29.3	-
FS-10-A FS-10-B	0.08 <0.05	<0.05 <0.05	12.07 6.17	<5.00 <5.00	<5.00 24.00	<5.00 <5.00	<5.00 <5.00	34.2 710	1.59 1.00	2.04 1.28	1.51 5.56	8.29 6.85	5.78 9.03	<5.00 <5.00	3.14 6.65	<0.05 <0.05	<0.05 <0.05	<5.00 <5.00	1.74 1.74	1.32 <0.20	17.4 17.3	0.68
FS-11	No Wate	r		******	••••····										>							
FS-12	No Wate	ur -	· ·												>							
FS-13	No Wate	r		*******	· · ·										>							
FS-14-A FS-14-B															> >			•				
	Not Sam	pled										-										

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-	Temp.	Eh	Conductivity	Acidity	Alkalinity
Sample	С	(mv)	(uS/cm)	(mg/L CaCO3)	(mg/L CaCO3)
,				((g /
		-			
FS-1-A	6.2	495	660	< 0.05	< 0.05
FS-1-B	6.8	491	943	< 0.05	< 0.05
FS-1-C	6.3	493	205	< 0.05	< 0.05
					:
FS-2-A	6.3	492	310	< 0.05	< 0.05
FS-2-B	6.1	492	176	< 0.05	< 0.05
FS-3-A	7.6	498	730	< 0.05	< 0.05
FS-3-B	7.6	483	996	< 0.05	< 0.05
FO (A)		450	500		
FS-4-A	8	452	523	< 0.05	< 0.05
FS-4-B	9.9	458	507	< 0.05	< 0.05
FS-4-C	10.9	464	365	< 0.05	< 0.05
	0.4	070	545		. 0.05
F0-0-A	9.4	379	515	< 0.05	< 0.05
F0-0-B	10.3	3/1	591	< 0.05	< 0.05
F3-5-C	11.7	305	504	< 0.05	< 0.05
ES-6-A	0.8	366	401	< 0.05	< 0.05
FS-6-B	9.0	Nowater	401	< 0.05	< 0.05
FS-6-C	<	No water			~ ~ ~
13-0-0	<	INC Waler	* *** ***********		
FS-7	<	No water			
		no nator			-
FS-8	8.9	236	221	< 0.05	< 0.05
					0.00
FS-9-A	10.3	247	261	< 0.05	< 0.05
FS-9-B	11.7	255	263	< 0.05	< 0.05
FS-9-C	12.1	307	126	< 0.05	< 0.05
	· · · · ·				
FS-10-A	7.3	281	155	< 0.05	< 0.05
FS-10-B	9.2	292	111	< 0.05	< 0.05
FS-11	<	No water			>
FS-12	<	No water			>
FS-13	<	No water			>
FS-14-A	<	No water			>
FS-14-B	<	No water	۲۰۰۰ میرون می		>
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Table 6. Physico-Chemical Parameter Values Fault Lake Tailings, March 1993

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Table 7. Metal and Major Ion Concentrations in Groundwater Sampled Fault Lake Tailings, March 1993

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Sample	Al (mg/L)	As (mg/L)	Ca (mg/L)	Cd (ug/L)	Co (ug/L)	Cr (ug/L)	Cu (ug/L)	Fe (ug/L)	K (mg/L)	Mg (mg/L)	Mn (ug/L)	Na (mg/L)	Ni (ug/L)	Pb (mg/L)	S (mg/L)	Se (mg/L)	Sn (mg/L)	Zn (ug/L)
FS-1A FS-1B FS-1C	0.06 <0.05 0.08	<0.05 <0.05 <0.05	80.94 95.42 16.55	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	2.19 1.29 <1	21.16 24.18 3.20	13.40 2.12 <1	15.34 53.95 14.72	<5.00 <5.00 <5.00	<0.05 <0.05 <0.05	72.48 113.43 11 16	<0.05 0.07 <0.05	<0.05 <0.05 <0.05	<5.00 <5.00 <5.00
FS-2A FS-2B	0.08 0.12	<0.05 <0.05	28.60 18.36	<5.00 <5.00	<5.00 <5.00	<5.00 <5.00	5.19 <5.00	<5.00 33.10	1.22 1.01	7.11 4.01	1.91 47.20	6.36 3.99	<5.00 <5.00	<0.05 <0.05	11.01 11.11	<0.05 <0.05	<0.05 0.06	<5.00 <5.00
FS-3A FS-3B FS-3C	0.08 0.08	<0.05 <0.05 	73.76 101.29	<5.00 <5.00	<5.00 <5.00	<5.00 <5.00	<5.00 <5.00	<5.00 <5.00	3.18 3.34	16.33 18.37	76.00 318.00	25.33 27.81	138.00 38.80	<0.05 <0.05	49.52 82.23 >	<0.05 <0.05	<0.05 ' <0.05	<5.00 <5.00
FS-4A FS-4B FS-4C	0.07 0.10 0.18	<0.05 <0.05 <0.05	52.60 62.05 18.25	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	1.34 14.10 56.80	5.88 6.52 3.37	11.89 12.17 2.89	248.00 176.00 24.70	18.42 20.43 48.16	<5.00 11.70 <5.00	<0.05 <0.05 <0.05	35.45 34.84 8.49	<0.05 <0.05 <0.05	<0.05 <0.05 <0.05	<5.00 <5.00 <5.00
FS-5A FS-5B FS-5C	0.17 0.17 0.23	<0.05 <0.05 <0.05	63.02 47.01 53.37	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	5.29 6.95 6.02	192.00 356.00 514.00	5.64 3.03 8.04	10.86 9.09 8.50	274.00 265.00 146.00	13.95 42.64 31.79	<5.00 <5.00 6.01	<0.05 <0.05 <0.05	33.78 33.58 32.07	<0.05 <0.05 <0.05	<0.05 <0.05 <0.05	<5.00 <5.00 <5.00
FS-6A FS-6B FS-6C	0.26	<0.05 	41.37	<5.00	<5.00	<5.00	6.95	<5.00	3.82	9.15	64.70	17.62	<5.00	<0.05	14.66 > >	<0.05	<0.05	<5.00
FS-7							•• •• • • • • • • • • • • • • • • • •						*		>			
FS-8A	0.09	<0.05	11.38	<5.00	5.29	<5.00	<5.00	19.32	2.05	2.39	5.44	4.84	<5.00	<0.05	12.54	<0.05	<0.05	<5.00
FS-9A FS-9B FS-9C	0.07 0.12 0.31	<0.05 <0.05 <0.05	29.03 26.81 6.38	<5.00 <5.00 <5:00	<5 <5 8.30	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	91.00 139.00 746.00	1.91 2.15 <1	4.64 4.35 1.13	156.00 110.00 1.12	6.16 11.01 11.87	<5.00 <5.00 46.30	<0.05 <0.05 <0.05	8.79 8.61 8.54	<0.05 <0.05 <0.05	<0.05 <0.05 <0.05	<5.00 <5.00 <5.00
FS-10A FS-10B	0.41 0.16	<0.05 <0.05	15.37 6.13	<5.00 <5.00	7.03 84.80	<5.00 <5.00	5.30 13.40	1.68 2.54	1.62 <1	2.09 1.16	1.58 1.92	8.24 2.87	19.40 532.00	<0.05 <0.05	3.79 8.37	<0.05 <0.05	<0.05 <0.05	7.50 0.14
FS-11							******								> ·			
FS-12		'												*******	>			
FS-13						· · · · · · · · · · · · · · · · · · ·									>			
FS-14-A FS-14-B								· · · · · · · · · · · · · · · · · · ·							>			
	Not Sam	pled								-11	-			······				

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Sample	Depth (m)	As (mg/L)	Cd (mg/L)	Cu (mg/L)	Fe(T) (m g/L)	Mn (mg/L)	Na (mg/L)	Ni (mg/L)	Pb (mg/L)	Zn (mg/L)	SO4 (mg/L)	рH
Pond #1	0.00	0.003	0.0003	<0.02	0.07	0.01	7.3	0.03	0.003	0.007	27	7.20
Pond #1	2.74	0.003	0.0002	<0.02	0.08	0.01	7.3	0.02	0.003	<0.005	28	7.40
Pond #1	4.57	0.003	0.0005	<0.02	0.06	0.01	7.3	0.02	0.002	0.009	28	7.60
Pond #2	0.00	0.007	0.0002	<0.02	0.05	0.01	5.3	<0.02	0.003	<0.005	26	7.60
Pond #2	4.57	0.006	0.0002	<0.02	0.06	0.01	4.9	<0.02	0.004	<0.005	25	7.60
Pond #2	7.62	0.010	0.0002	<0.02	0.17	0.03	5.0	<0.02	0.002	<0.005	24	7.20
Pond #3	0.00	<0.003	0.0002	<0.02	0.05	0.01	1.8	0.02	0.001	<0.005	18	7.40
Pond #3	9.14	<0.003	0.0002	<0.02	0.05	0.01	2.0	0.02	0.002	0.011	18	7.00
Pond #3	15.24	0.007	0.0002	<0.02	0.60	0.36	2.0	0.03	0.002	0.006	15	6.70
Pond #4	0.00	0.006	0.0002	<0.02	0.03	0.01	.9	<0.02	<0.002	0.013	25	7.40
Pond #4	4.57	0.006	0.0002	<0.02	0.04	0.01	9	<0.02	0.003	<0.005	25	7.80
Pond #4	8.23	0.005	0.0003	<0.02	0.05	0.01	8.9	<0.02	0.003	0.007	25	7.80
Pond #5	0.00	0.003	0.0002	<0.02	0.02	0.01	7.2	0.02	0.001	<0.005	67	7.80
Pond #5	9.14	0.002	0.0002	<0.02	0.02	0.01	7.2	0.02	0.003	0.009	65	7.80
Pond #5	15.24	0.007	0.0003	<0.02	0.28	0.57	7.8	0.02	0.002	0.007	69	7.00

Table 8. Chemical Quality of Surface Water Fault Lake Tailings, October 1992













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	301.3 300.9 300.5 300.1 299.7 299.7 299.1	P 288.5 207.7
CENTRE DE TECHNOLOGIE NORANDA ENVIRONMENTAL PROGRAM	Figure 6. Flow Vectors and Head Contours With Infiltration Prepared by: BA MAR 1993	Fault Lake Tailings Site Porous Envelope Conceptual Modelling Results

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APPENDIX E.1

Grain Size Distribution Curves






























































APPENDIX E.2

Monitoring well Data

Piezometer	Northing (m)	Easting (m)
FS-1	5161204.9	516063.5
FS-2	5160988.2	515777.6
FS-3	5161495.7	515900.7
FS-4	5161653.3	515495.0
FS-5	5161677.0	515577.9
FS-6	5161823.3	515659.0
FS-7	5161914.8	515885.8
FS-8	5162074.8	515886.4
FS-9	5162204.6	515937.6
FS-10	5161943.7	515941.2
FS-11	5161797.4	515748.6
FS-12	5161546.0	515727.9
FS-13	5161378.6	515952.5
FS-14	5161526.2	516041.2

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Table B1. Locations of Monitoring Stations

		Elevat	tions (m)	······································		
Piezometer	Ground	Top of Piezometer	Tip of Piezometer	Depth of tip below ground	Nov 24/92 Water	Mar 26/93 Water
FS-1-A FS-1-B FS-1-C	304.47	305.27 305.24 305.24	287.62 292.05 294.92	16.85 12.42 9.55	298.36 298.65 298.69	298.35 298.60 298.67
FS-2-A FS-2-B	316.11	316.84 316.84	300.59 305.82	15.52 10.29	307.23 307.25	307.21 307.32
FS-3-A FS-3-B FS-3-C	322.12	322.99 322.92 322.94	271.90 302.92	50.21 19.20	299.53 299.89	299.52 299.63
FS-4-A FS-4-B FS-4-C	322.63	323.36 323.39 323.46	291.02 294.00 298.65	31.62 28.63 23.98	299.51 299.45 299.58	291.91 299.46 299.51
FS-5-A FS-5-B FS-5-C	321.93	322.80 322.74 322.85	283.52 286.54 295.65	38.41 35.39 26.29	299.58 301.77 299.57	299.58 301.77 299.57
FS-6-A FS-6-B FS-6-C	320.90	321.71 321.42 321.44	287.54 298.95 311.72	33.36 21.95 9.18	300.30 299.52 311.74	300.30 299.52 311.74
FS-7	309.89	310.65	301.70	8.19	·	
FS-8	302.76	303.51	298.41	4.35	299.29	
FS-9-A FS-9-B FS-9-C	302.58	303.33 303.41 303.47	291.03 292.21 298.67	11.55 10.36 3.91	299.33 299.26 299.37	
FS-10-A FS-10-B	304.19	304.95 304.96	294.13 296.70	10.06 7.49	299.31 299.35	
FS-11	320.38	320.60	299.17	21.21	299.35	299 .31
FS-12	320.04	320.04	298.89		299.05	299.12
FS-13	319.28	319.95	296.40	22.89	299.45	299.46
FS-14-A FS-14 -B	322.97	323.19 324.02	292.84 297.37	30.13 25.60	299.24 299.46	296.82 299.28

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Table B2. Fault Lake Piezometer Data

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PROJECT COORDIN DEPTH C	NE Ti FAULT LAKE TA IATES: N5161852.0 IF CASING:]RANDA T NILINGS, FALCON DRE515879.487		Y CENTRE HOLE NO: FS-7 ELEVATION: 309 DIP:	- BOREHOL 9.890 n	LOG TOTAL DEPTH: 8.20 m AZIMUTH: CORE SIZE:
DEPTH (n)	CIRE DESCRIPTION			PIEZDMETER LDCATIDNS	VATER LEVELS	HYDRAULIC CONDUCTIVITY (cm/s)
- 0 m 		Sand and betu Eni	LDERS D OF HOLE			NCI WATER
- 10 m						
▶		··· ··· ··· <u>-</u> ··· ·		JND	<u></u>	
- 				BENTONITE SEAL		
- 				BACKFILL AND/OR	SAND	
- - 15 n				SAND AND PIEZON	eter tip	















APPENDIX E.3

Laboratory certificates of analyses

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CENTRE DE TECHNOLOGIE NORANDA

CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : 1.St-Arnaud

PROJET / PROJECT: V2-1 T03

Date: 2/26/93

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.ab #	ab # 0.0. Description			A!			As		Ca		Cđ			Co		
2663	FS-1-4	EAU		. 07	mg/L	<	. 35	 mg/L	84.44		(5.00	ua/L	(5.00	uah
2664	FS-1-8	EAU		. 05	mg/L	(. 05	ng/L	84.86	ng/L	<	5.00	94/1	1	5.00	20/1 20/1
2665	FS-1-C	EAU		. 07	mg/L	<	. 35	mg/L	18.42	πg/'_	(5.00	19/1	ί.	5.00	20/L
2666	FS-2-A	EAU	<	. 05	ng/L	(. 05	mq/L	33.68	mq/L	(5.00	19/1	1	5.00	14/5
2667	FS-2-8	EAU	<	.05	mg/L	{	. 05	mg/L	21.54	mg/L	<	5.00	-g/L	;	5.00	19/5 - 19/5
2668	FS-3-A	EAU		. 38	mg/L	(.05	ng/L	9.52	mg/L	<	5.00	ug/L	(5.00	19/1
2669	= S - 8	EAU		. 37	mg/L	<	.05	mg/L	11.66	mg/L	(5.00	29/2	(5.00	-9/1
2670	FS-9-A	EAU		. 08	mg/L	<	. 05	mg/L	28.71	ng/L	(5.00	ug/L	<	5.00	29/L
2671	=S-9-8	EAU		. 38	mg/L	<	.05	mg/L	22.81	mg/L	(5.00	ug/L	(5.00	19/L
2672	FS-9-C	EAU		. 38	mq/L	<	. 05	mq/L	6.18	mg/L	<	5.00	29/1		7.23	10/L
2673	=S-9-S2	EAU		20.62	mg/L		1.04	mg/L	101.88	ng/L		1.02	na/L	<	5.00	-9/- 29/1
2674	FS-10-A	EAU		. 08	mg/L	<	. 05	mg/L	12.07	ng/L	(5.00	34/E	<	5.00	19/L
2875	FS-10-8	EAU	<	. 05	mg/L	<	.05	mg/L	6.17	mg/L	<	5.00	ua/L		24.00	JOL
2676	FS-10-R1	EAU		.07	ng/L	<	. 05	ng/L	9.62	ng/L	<	5.00	39/1		21.60	uq/1
2677	FS-10-R2	EAU		. 37	mg/L	<	.05	mg/L	9.64	mg/L	<	5.00	29/1		22.70	29/1
2678	FS-10-93	EAU		. 36	ng/L	<	. 35	ng/i	4.76	ng/L	(5.00	19/1		19.00	ua/L
2679	FS-10-S1	EAU		20.39	mg/L		1.02	ng/L	101.25	mg/L		1.01	mg/L	<	5.00	19/L
2680	s-15-0	EAU		.05	ng/L	<	. 05	mg/L	. 15	ng/L	(5.00	19/2	<	5.00	ug/L
2681	FS-15-S3	EAU		20.85	ag/L		1.06	πg/:_	103.14	mg/L		1.33	mg/L	(5.00	49/L

Commentaires/ Comments: Par ICP, sauf Cl, NO3, SO4: par IC. Fe(+2) et Fe(+3): par colorimétrie.

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Effectué_par/ Work by _: R._ Pelletier _ _ _ _ D.. Thériault - _- _- _-
CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : L.St-Arnaud

PROJET / PROJECT: V2-1 T03

Date: 2/26/93

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Lab #	1.0.	Descript	ton	C	r		3	, i	;	ê			(Y	9
2663	FS-1-A	EAU	(5.00	µg/L	(5.00	29/1	10.90	ug/L		2.39	mg/1		21.58	 πα/:
2664	FS-1-8	EAU	(5.00	ug/L	(5.00	Jq/L	13.90	Jq/L		1.65	ng/L		21.02	
2665	FS-1-C	EAU	(5.00	ug/L		5.04	uq/5	49.40	uq/L		1,30	ma/L		3.35	37- ma/:
2666	FS-2-A	EAU	(5.00	uq/L	(5.00	uq/L	27.20	uq/L		1.09	mg/L		8.17	
2667	FS-2-3	EAU	(5.00	ug/L		7.16	ug/L	17.70	uq/L		1.73	ma/L		3.64	ma/:
2668	FS-3-A	EAU	<	5.00	ag/L		. 08	ug/1	. 14	nq/l	(1.00	ng/L		2.47	3/- "d/
2669	FS-8	EAU	(5.00	ug/L	(5.00	ug/L	1.31	mg/L		1,92	ma/L		2.58	- 3r -
2670	FS-9-A	EAU	<	5.00	ug/L	<	5.00	ug/L	23.20	ug/L		1.84	ma/L		3.83	na/L
2671	FS-9-8	EAU	<	5.00	ug/L	(5.00	ug/L	42.50	ug/L		2.45	mq/L		3.33	na/1
2672	FS-9-C	EAU	<	5.00	39/L	(5.00	Ja/L	55,40	59/L		1.20	.ng/L		1.11	
2673	FS-9-52	EAU	<	5.00	ug/L		11.02	mg/L	30.32	mg/L	(1.00	mq/L	(10	ng/1
2674	FS-10-A	EAU	<	5.00	ug/L	(5.00	ug/L	34.20	ug/L		1.59	ng/L		2.04	ng/L
2675	FS-10-B	EAU	<	5.00	ug/L	<	5.00	aq/L	.71	mg/L	(1.00	ma/L		1.28	ng/L
2676	FS-10-R1	EAU	<	5.00	ug/L	(5.00	ug/L	. 56	mg/L		1.11	mq/L		1.14	mg/L
2677	FS-10-R2	EAU	<	5.00	ug/L	<	5.00	ug/L	. 56	mg/L		1.09	mg/L		1.14	mg/L
2678	FS-10-R3	EAU	<	5.00	ug/L	(5.00	ug/L	1.60	חg/L	(1.00	79/1		. 79	mg/L
2679	FS-10-S1	EAU	<	5.00	29/1		10.86	mg/L	30.04	mg/L	<	1.00	mg/L	<	. 10	ח פ /L
2680	FS-15-0	EAU	<	5.00	ug/L	(5.00	ug/L	. 15.00	ug/L	(1.00	nq/L	<	. 10	na/L
2581	=S-15-S3	EAU	<	5.00	19/L		11.15	mg/L	30.68	mg/L	<	1.00	mg/L	(. 10	mg/L

Commentaires/ Comments: Par ICP, sauf Cl, NO3, SO4: par IC. Fe(+2) et Fe(+3): par colorimétrie.

Effectué par/ Work by : R. Pelletier - - - D.- Thériault

Thériault -- -

CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : 1.St-Arnaud

PROJET / PROJECT: V2-1 T03

Date: 2/25/93

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_ab #	1.0.	Description	Mo	}	X	â		N	•		ρ	'b		S
2563	FS-1-4	EAU	21.70	29/L	1.48	mq/L	(5.00		<u>.</u>	. 35	74/2	13.74	na/:
2664	=S-1-B	EAU	5.60	۱۹/۲	37.44	ng/L	(5.00	ua/1	(. 35	ma/L	85.69	מן/ניי זעמר
2865	5-1-C	EAU	3.86	ug/1	23.68	mq/L	{	5.00	24/1		. 07	ភាជ/៤	11.82	ng/1
2666	FS-2-A	EAU	22.70	ug/L	6.49	mq/L		11.90	uq/L	(. 05	קר. קר/L	12.20	- 10/:
2667	FS-2-8	EAU	95.40	uq/L	7.89	mg/L		9.75	uq/L	{	. 35	ng/L	12.20	ng/4
2668	FS-3-A	EAU	16.60	34/L	2.84	mg/L		14.10	24/1	(. 35	ng/L	4.63	- 37 - - 37
2669	FS-8	EAU	6.31	ng/L	6.95	ma/L		6.68	uq/L	(. 05	ma/L	12 99	nal:
2670	FS-9-A	EAU	73.90	ug/L	5.15	ng/L		9.75	10/1	Ċ	. 35	ma/L	7.93	
2571	=S-9-3	EAU	. 12	ma/L	22.33	ma/L	(5.00	uq/L	<	.05	mg/L	7.80	na/1
2672	5-9-C	EAU	1.51	mg/L	11.83	mg/L		.11	ma/L	<	.35	mg/L	10.03	na/-
2573	FS-9-52	EAU	20.86	ma/L	1.04	mq/L		10.18	ng/L	(. 05	ma/L	35.74	- 10 na/'
2674	FS-10-A	EAU	1.51	ng/L	8.29	ng/L		5.78	uqil	(.05	тя/L	3.14	πα/1
2575	FS-10-8	EAU	5.56	mg/L	6.85	ng/L		9.03	uq/L	<	. 05	ng/L	6.65	mg/2
2676	FS-10-81	EAU	4.28	ng/L	6.94	ng/L		11.80	uq/L	{	. 35	ng/L	6.80	ng/1
2677	FS-10-R2	EAU	4.25	ma/L	6.86	ma/L		10.20	aq/L	<	. 35	ng/L	6.76	ma /!
2678	FS-10-R3	EAU	4.17	mg/L	4.41	ng/L		7.16	ag/t	(.35	ma/L	4.05	mg/1
2679	FS-10-51	EAU	20.60	mg/L	1.02	ma/L		10.09	ma/!	(.05	ma/L	35.26	ng/1
2680	=S-15-0	EAU	1.44	ad/L	1.10	ma/L	(- 5.30	adít	(.05	ma/:	. 18	ma/!
2681	s-15-53	EAU	20.88	ng/L	1.03	ag/L		10.30	mg/L	<	. 35	mg/L	35.79	ng/1

Commentaires/ Comments: Par ICP, sauf Cl, NO3, SO4: par IC. Fe(+2) et Fe(+3): par colorimétrie.

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Effectué par/ Work by : R. Pelletier ---- D. Thériault

CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : L.St-Arnaud

PROJET / PROJECT: V2-1 T03

Date: 2/28/93

.ab #	.).	Description		S	e		S	n		2	In
2663	=S-1-A	EAU	<	. 35	ng/L	(. 35	mg/L	(5.00	
2664	FS-*-3	EAU	(. 05	ng/L	1	. 35		(5.00	Ja/L
2665	=S-'-C	EAU	!	.05	ng/1	(.05	ng/L	(5.00	ua/1
2668	=S-2-A	EAU	(. 05	mg/L	{	. 35	mg/L	<	5.00	39/L
2667	FS-2-8	EAU	(.05	mg/L	<	.05	mg/L	(5.00	19/1
2668	=S-3-A	EAU	<	. 05	mg/L	(. 05	mg/L		. 15	na/L
2669	FS-8	EAU	{	.05	mg/L	(. 05	mg/L		15.40	ug/L
2670	5-9-A	EAU	<	. 05	mg/L	{	. 05	mg/L	(5.00	ug/L
2671	FS-9-8	EAU	<	. 05	mg/L	<	. 05	mg/L	<	5.00	µg/L
2672	FS-9-C	EAU	<	.05	ng/L	<	. 05	mg/L		27.00	ug/L
2673	FS-9-52	EAU	(.05	mg/L	<	.05	mg/L		21.15	mg/L
2674	FS-10-A	EAU	<	.05	mg/L	(. 05	ng/L	(5.00	ug/L
2675	FS-10-8	EAU	<	. 05	mg/L	(. 05	mg/L	<	5.00	ug/L
2676	FS-10-R1	EAU	(. 05	mg/L	(. 05	mg/L	(5.00	19/L
2677	FS-10-R2	EAU	<	. 05	mg/L	<	. 05	mg/L	<	5.00	ug/L
2678	FS-10-93	EAU	(. 05	mg/L	(.05	ng/L	(5.00	ug/L
2679	FS-10-S1	EAU	<	. 05	mg/L	(. 05	mg/L		21.18	mg/L
2680	FS-15-D	EAU	(. 05	mg/L	<	.05	ng/L	(. 5.00	ug/L
2681	FS-15-53	EAU	<	.05	mg/L	<	. 05	ag/L		21.39	mg/L

Commentaires/ Comments: Par ICP, sauf Cl, NO3, SC4: par IC. Fe(+2) et Fe(+3): par colorimétrie.

CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : L.St-Arnaud

PROJET / PROJECT: V2-1 T03

Oate: 2/26/93

Lao #	.).	Description	Cl			NO	3	SO	4	Fe(+2	}	Fe(+3	}
2663	=3- ° -4	Water	4.02 m	ng/L		12.90	mg/1	210.00	ma/1			····	
2564	=\$-1- <u>3</u>	Nater	5.4 7 m	g/L		24.90	ng/L	239.00	ng/L				
2665	FS-1-0	Water	3.97 m	ng/L		3.37	mg/1	31.90	ng/L				
2666	=5-2-A	Water	13.50	g/L		8.22	ng/L	33.50	aq/L				
2867	=5-2 -8	Water	3.70 m	ng/L		5.25	mg/L	32.40	na/L				
2668	=S-3-4	Nater	2.07 m	ig/L		2.08	mg/L	11.20	mg/L				
2669	= 5 - 8	Water	1.56 m	ig/L		5.75	mg/L	27.40	mq/_	1.05	ng/1		
2670	=5-9-A	Water	5.20 m	ig/L		2.30	mg/L	16.60	ng/L		- - -		
2671	=S-9-8	Water	9.51 m	ig/L		2.38	mg/L	20.70	mg/L				
2672	FS-9-C	Water	3.06 m	g/L		2.61	ng/L	29.30	ng/L				
2673	=S-9-S2	Water	3.62 m	ng/L	<	. 20	mg/L	8.35	ng/L				
2674	FS-10-A	Nater	1.74	ig/L		*.32	ng/L	17,10	ng/L				
2675	=S-10-B	Water	1.74 m	ig/L	:	. 20	mg/l	17.30	mg/L	. 68	ng/1		
2676	=S-10-91	Water		•••			•	18.00	mg/L		•		
2677	FS-10-R2	Water						18.50	mg/L				
2678	FS-10-83	Water						7.98	mg/L				
2679	FS-10-S1	Water						97.00	mg/L			29.10	ng/L
2680	=S-15-D	Water						2.57	mg/L				••
2681	=\$-15-\$3	Water						96.80	mg/L			29.10	mg/L

Commentaires/ Comments: Par ICP, sauf Cl, NO3, SO4: par IC. Fe(+2) et Fe(+3):par colorimétrie.

Effectué par/ Work by-: R. Pelletier - D. Thériault -

CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

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A/To :	L.St-/	Arnaud B.Aub	é		PROJET	: / PROJEC	ST: V2 13	21 01					0	ate: 3/	31/93
Lab #	1.0.	Description	A	1	ļ	5	C	a		(d		C	ių	
3350 3351	MLW #1 BLW #2	Tailing H2O Tailing H2O	12.30	mg/L mg/L (1.97 .25	mg/1 mg/1	.42 4.05	g/L g/L	(. 21	mg/L mg/L	(3.78	mg/l ng/l	
3352 3353	S #1 S #2	lake water lake water													
3354 3355 3356 3357	FS-4 FS-6 FS-11 FS-12a	Tailing H2O Tailing H2O Tailing H2O 〈 Tailing H2O 〈 Tailing H2O 〈	.80 .51 :25 .25	mg/L < mg/L < mg/L < mg/L <	. 25 . 25 . 25 . 25	ng/L mg/L mg/L mg/L	1,76 1,71 .65 .88	9/L 9/L 9/L 9/L	< < <	.02 .05 .02 .02	mg/L mg/L mg/L mg/L	<	.03 .06 .02 .03	mg/L mg/L mg/L mg/L	
3358 3359	FS-126 FS-13	Tailing H2O (Tailing H2O (. 25 . 25	mg/L ⟨ mg/L ⟨	. 25 . 25	mg/l mg/l	. 70 . 52	g/L g/L	((. 92 . 02	mg/L mg/L	< <	.02 .02	mg∕L mg∕L	

Commentaires/ Comments: par ICP.

Effectué par/ Work by : D. Thériault B.Legault -

CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To :	L.St-/	Arnauc B.Au	Dé		PROJET	/ PROJE	CT: V2 13	121 01				C	ate: 3/31	/93
Lao \$	¦.D.	Description	F	e		lg	K	 In	N	la		P	b	
3350 3351	MLW #1 BLW #2	Tailing H2O Tailing H2O	305.00 167.00	mg/L mg/L	176.00 40.60	mg/L mg/L	51.70 89.50	mg/L mg/L	69.70 72.90	mg/L mg/l		18.20 42.50	mg/L mg/L	
3352 3353	S #1 S #2	Lake water Lake water												
3354	FS-4 FS-6	Tailing H20	96.00	mg/L	778.00	mg/L	19.80	mg/L	16.80	mg/L	<	. 25	mg/L	
3356	FS-11	Tailing H20	.58	mg/L mg/L	322.00	mg/L mg/l	2.30	mg/L ma/l	82.00 64.00	mg/L ma/l	((.25	mg/L mg/l	
3357	FS-12a	Tailing H2O	. 35	mg/L	189.00	mg/L	1.14	mg/L	66.30	mg/L	(. 25	mg/L	
3358	FS-12b	Tailing H2C	. 22	mg/L	132.00	mg/L	. 53	mg/L	75.20	mg/L	<	. 25	mg/L	
3359	ES-13	Tailing H2C	. 23	mg/L	391.00	mg/L	1.09	mg/L	43.50	mg/L	<	. 25	mg/L	

Commentaires/ Comments: par ICP.

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Effectué par? Work by : D. Thériault 3.1egault J. Groleau

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CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To :	L.St-/	Arnaud B.Aub	é			PROJET	' / PROJE	CT: V2 13	821 01			C	ate: 3/31/93
Lab #	I.D.	Description	. <u> </u>	S		S	e	2	In	<u>, , ,</u>	K		1
3350 3351	MLW #1 3LW #2	Tailing H2O Tailing H2O	.61 .11	g/L g/L	< <	. 50 . 50	mg/L mg/L	317.00 133.00	mg/L mg/L	< 5.00 < 5.00	mg/L mg/L		
3352 3353	S #1 S #2	Lake water Lake water						. 95 . 93	g/L g/L			1.22 1.23	g/L g/l
3354 3355 3356 3357 3358 3359	FS-4 FS-6 FS-11 FS-12a FS-12b FS-13	Tailing H2C Tailing H2O Tailing H2O Tailing H2O Tailing H2O Tailing H2O Tailing H2O	5.82 5.23 .74 2.00 .75	9/L 9/L 9/L 9/L 9/L	< < < < <	.50 .50 .50 .50 .50	mg/L mg/L mg/L mg/L mg/L	.76 1.72 .07 .09 .20	mg/L mg/L mg/L mg/L mg/L	245.00 171.00 105.00 176.00 155.00	mg/L mg/L mg/L mg/L mg/l		

Commentaires/ Comments: par ICP.

Effectué par/ Work by : D. Thériault 8.Legault --

J. Groleau

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CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To :	L.St-/	Arnaud B.Au	ibé	PROJET / PROJECT: V2 1321 01	Date: 3/31/93
Lab #	1.0.	Description	Ni		
3350	MLW #1	Tailing H2O			
3351	BLW #2	Tailing H2O			
3352	S #1	Lake water			
3353	S #2	Lake water			
3354	FS-4	Tailing H2O	133.00 mg/L		
3355	FS-6	Tailing H2O	16.30 mg/L		
3356	FS-11	Tailing H2O	2.92 mg/L		
3357	FS-12a	Tailing H2O	4.69 mg/L		
3358	FS-12b	Tailing H2O	2.55 mg/L		
3359	FS-13	Tailing H2O	3.65 mg/L		

Commentaires/ Comments: par ICP.

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Effectué par/ Work by : D. Thériault

J. Groleau

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CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : L.St-Arnaud

PROJET / PROJECT: V2-1 T03

Date: 4/30/93

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Lab #	1.0.	Description			1		Å	.5	C	1		C	d		C	0
6563	FS-1A	Water		.06	ng/L	<	.05	mg/L	80.94	ng/L	(5.00	µg/L	<	5.00	ug/L
6564	FS-18	Water	<	.05	ng/L	(.05	mg/L	95.42	ng/L	<	5.00	µg/L	(5.00	ug/L
6565	FS-1C	Water		.08	ag/L	<	.05	ng/L	16.55	ag/L	(5.00	Mg/L	<	5.00	ug/L
6566	FS-2A	Water		.08	mg/L	(.05	ng/L	28.60	ng/L	<	5.00	ug/L	<	5.00	ug/L
6567	FS-2B	Water		. 12	ag/L	<	.05	ng/L	18.36	as/L	<	5.00	ug/L	<	5.00	ug/L
6568	FS-2C	Vater	<	.05	mg/L	<	.05	mg/L	.11	mg/L	(5.00	ug/L	<	5.00	ug/L
6569	FS-3A	Nater		.08	mg/L	<	.05	ag/L	73.76	Mg/L	<	5.00	ug/L	(5.00	ua/L
6570	FS-38	Water		.08	mg/L	<	.05	ag/L	101.29	Mg/L	<	5.00	ug/L	(5.00	ug/L
6571	FS-3C	Nater		20.60	ag/L		1.01	ag/L	100.50	ma/L		1.02	mg/L	ć	5.00	ua/L
6572	FS-4A	Water		.07	mg/L	(.05	ag/L	52.60	ng/L	<	5.00	ug/L	(5.00	ug/L
6573	FS-48	Vater		. 10	mg/L	(.05	ng/L	62.05	ag/L	<	5.00	ug/L	<	5.00	ua/L
6574	FS-4C	Water		. 18	mg/L	(.05	mg/L	18.25	mg/L	(5.00	ug/L	<	5.00	uq/L
6575	FS-5A	Vator		.17	ag/L	(.05	mg/L	63.02	ag/L	<	5.00	ug/L	(5.00	ue/L
6576	FS-58	Water		.17	mg/L	<	.05	sg/L	47.01	mg/L	<	5.00	ug/L	ć	5.00	ug/L
6577	FS-5C	Vater		.23	ng/L	<	.05	ma/L	53.37	ma/L	<	5.00	uq/L	(5.00	ug/L
6578	FS-6A	Water		. 26	mg/L	(.05	mg/L	41.37	mg/L	ć	5.00	uq/L	(5.00	ua/L
6579	FS-6B	Water		21.04	mg/L		1.07	ad/L	101.23	mg/L	-	1.03	na/L	Ì	5.00	ua/L
6580	FS-8A	Nater		.09	mg/L	(.05	mg/L	11.38	mq/L	(5.00	ue/L	•	5.29	u o/ 1
6581	FS-88	Vater		20.88	mg/L		1.07	nd/L	100.82	ng/L	•	1.03	mg/L	(5.00	ua/L
6582	FS-9A	Vater		.07	ag/L	<	.05	mg/L	29.03	mg/L	(5.00	ua/L	i	5.00	40/L
6583	FS-9B	Vatar		. 12	ng/L	<	.05	ag/L	25.81	mq/L	ć	5.00	ua/L	ć	5.00	ua/L
6584	FS-9C	Tater		.31	mg/L	ć	.05	mg/L	6.38	mg/L	(5.00	ug/L	•	8.30	ua/L
6585	FS-10A	Vater		.41	mg/L	ć	.05	mg/L	15.37	mg/L	Ċ	5.00	uq/L		7.03	ug/L
6586	FS-108	Tater		. 16	ne/L	ć	.05	mg/L	6.13	ng/L	(5.00	uo/L		84.80	ug/1
6587	PREP.\$1	Water		9.17	ag/L	-	.35	ng/L	253.30	ma/L	•	10.70	uq/L		48.40	ua/L
6588	SURF #1	Tater		.47	mg/L	(.05	mg/L	71.49	ng/L	(5.00	ца/!		7.60	ua/L
6589	SURF #2	Vater		.47	ma/L	(.05	mg/L	70.16	mg/L	(5.00	ua/L		7.18	ua/L

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Commentaires/ Comments: Par ICP.

Effectué par/ Work by : R. Pelletier

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CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : L.St-Arnaud

PROJET / PROJECT: V2-1 T03

Date: 4/30/93

Lab #	1. D .	Description		Č)r		C)u		F				K		N	9	
6563	FS-1A	Vater	<	5.00	µg/L	<	5.00	ug/L	<	5.00	us/L		2.19	ag/L		21.15	ng/L	
6564	FS-18	Water	<	5.00	µg/L	<	5.00	µg/L	(5.00	Ha/L		1.29	mg/L		24.18	ng/L	
6565	FS-1C	Water	(5.00	Hg/L	<	5.00	µg/L	<	5.00	J/E	<	1.00	ng/L		3.20	ng/L	
6566	FS-2A	Water	(5.00	µg/L		5.19	µg/L	(5.00	ug/L		1.22	Mg/L		7.11	mg/L	
6567	FS-2B	Water	<	5.00	µg/L	<	5.00	µg/L		33.10	Hg/L		1.01	ag/L		4.01	ag/L	
6568	FS-2C	Vater	(5.00	µg/L	<	5.00	µg/L	(5.00	Ha/L	۲	1.00	mg/L	(. 10	ng/L	
6569	FS-3A	Nater	<	5.00	ug/L	<	5.00	ug/L	(5.00	ug/L		3.18	mg/L		16.33	ng/L	
6570	FS-38	Vater	(5.00	µg/L	(5.00	µg/L	(5.00	µg/L		3.34	mg/L		18.37	ng/L	
6571	FS-3C	Vater	(9.00	ug/L		11.08	mg/L		29.75	M/L	<	1.00	ag/L		.11	mg/L	
6572	FS-4A	Water	<	5.00	µg/L	(5.00	µg/L		1.34	mg/L		5.88	mg/L		11.89	mg/L	
6573	FS-4B	Water	(5.00	µg/L	(5.00	ug/L		14.10	us/L		6.52	ng/L		12.17	mg/L	
6574	FS-4C	Vater	<	5.00	µg/L	(5.00	µg/L		56.80	Hg/L		3.37	mg/L		2.89	mg/L	
6575	FS-5A	Water	<	5.00	ug/L		5.29	us/L		192.00	J/E		5.64	Mg/L		10.86	ag/L	
6576	FS-5B	Water	(5.00	µg/L		6.95	µg/L		356.00	He/L		3.03	mg/L		9.09	mg/L	
6577	FS-5C	Water	(5.00	Hg/L		6.02	Ha/L		514.00	J/L		8.04	ag/L		8.50	ag/L	
6578	FS-6A	Water	(5.00	µg/L		6.95	ug/L	(5.00	HO/L		3.82	mg/L		9.15	mg/L	
6579	FS-68	Water	<	5.00	Hg/L		11.34	ag/L		30.21	M/L	<	1.00	mg/L		. 13	mg/L	
6580	FS-8A	Water	(5.00	µg/L	(5.00	µg/L		19.32	Mg/L		2.05	Mg/L		2.39	mg/L	
6581	FS-8B	Water	(5.00	Hg/L		11.25	as/L		30.08	Mg/L	<	1.00	ng/L	(. 10	mg/L	
6582	FS-9A	Water	(5.00	µg/L	(5.00	µg/L		91.00	µg/L		1.91	mg/L		4.64	mg/L	
6583	FS-9B	Water	(5.00	ug/L	<	5.00	µg/L		139.00	Hg/L		2.15	ng/L		4.35	ng/L	
6384	FS-9C	Water	(5.00	µg/L	<	5.00	µg/L		746.00	µg/L	<	1.00	mg/L		1.13	mg/L	
6585	FS-10A	Water	(5.00	ug/L		5.30	µg/L		1.68	mg/L		1.62	ng/L		2.09	mg/L	
6586	FS-108	Water	(5.00	µg/L		13.40	µg/L		2.54	mg/L	(1.00	mg/L		1.16	ng/L	
6587	PREP.#1	Water	<	5.00	µg/L		144.00	µg/L		908.70	mg/L		5.99	∎g/L		238.66	eg/L	
6588	SURF #1	Water	(5.00	µg/L		6.44	µg/L		58.27	ng/L		2.39	mg/L		19.74	mg/L	
6589	SURF #2	Water	(5.00	µg/L		8.21	µg/L		57.19	mg/L		2.11	mg/L		19.24	mg/L	

Commentaires/ Comments: Par ICP.

Effectué par/ Work by : R. Pelletier

CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : L.St-Arnaud

PROJET / PROJECT: V2-1 T03

Date: 4/30/93

Lab #	1.D.	Description		H	ก	Ň	1		X	i		P	b		s
6563	FS-1A	Water		13.40	µg/L	15.34	ng/L	<	5.00	µg/L	<	.05	ag/L	72.48	ng/L
6564	FS-18	Water		2.12	µg/L	53.95	ng/L	<	5.00	µg/L	۲	.05	ag/L	113.43	ng/L
6565	FS-1C	Water	۲	1.00	µg/L	14.72	MJ/L	(5.00	µg/L	(.05	ag/L	11.16	mg/L
6566	FS-2A	Water		1.91	µg/L	6.36	mg/L	<	5.00	µg/L	۲	.05	mg/L	11.01	ng/L
6567	FS-28	Water		47.20	µg/L	3.99	mg/L	(5.00	µg/L	(.05	∎g/L	11.11	mg/L
6568	FS-2C	Water	(1.00	µg/L	1.15	mg/L	(5.00	µg/L	(.05	mg/L	. 27	ng/L
6569	FS-3A	Water		76.00	µg/L	25.33	mg/L		138.00	µg/L	۲	.05	mg/L	49.52	sg/L
6570	FS-3B	Water		318.00	µg/L	27.81	mg/L		38.80	µg/L	<	.05	mg/L	82.23	ag/L
6571	FS-3C	Water		20.22	mg/L	1.21	mg/L		9.90	mg/L	(.05	mg/L	36.00	mg/L
6572	FS-4A	Water	1	248.00	µg/L	18.42	mg/L	{	5.00	µg/L	(.05	ng/L	35.45	mg/L
6573	FS-48	Vater		176.00	µg/L	20.43	MJ/L		11.70	ug/L	(.05	ag/L	34.84	mg/L
6574	FS-4C	Water		24.70	µg/L	48.15	ag/L	(5.00	µg/L	(.05	mg/L	8.49	mg/L
6575	FS-5A	Water	:	274.00	µg/L	13.95	ag/L	(5.00	J/gu	(. 05	mg/L	33.78	mg/L
6576	FS-58	Water	1	265.00	µg/L	42.64	mg/L	<	5.00	µg/L	<	.05	mg/L	33.58	ng/L
6577	FS-5C	Water		146.00	µg/L	31.79	ng/L		6.01	µg/L	<	.05	mg/L	32.07	mg/L
6578	FS-6A	Water		64.70	µg/L	17.62	mg/L	(5.00	µg/L	<	.05	mg/L	14.66	mg/L
6579	FS-68	Water		20.50	ag/L	1.44	mg/L		10.03	ng/L	<	.05	ag/L	36.29	mg/L
6580	FS-8A	Water		5.44	mg/L	4.84	ag/L	<	5.00	µg/L	<	.05	mg/L	12.54	mg/L
6581	FS-88	Water		20.47	ag/L	1.29	ng/L		10.00	ag/L	<	.05	mg/L	35.72	mg/L
6582	FS-9A	Water	•	156.00	µg/L	6.16	mg/L	(5.00	µg/L	<	.05	mg/L	8.79	mg/L
6583	FS-98	Vater		110.00	µg/L	11.01	ag/L	(5.00	µg/L	(.05	∎g/L	8.61	ag/L
6584	FS-9C	Water		1.12	mg/L	11.87	mg/L		46.30	µg/L	<	.05	mg/L	8.54	mg/L
6585	FS-10A	Water		1.58	mg/L	8.24	ag/L		19.40	µg/L	(.05	mg/L	3.79	mg/L
6586	FS-10B	Water		1.92	mg/L	2.87	mg/L		532.00	µg/L	۲	.05	ng/L	8.37	ng/L
6587	PREP.#1	Water		29.80	mg/L	7.94	ag/L		28.20	µg/L	<	.05	ag/L	*****	mg/L
6588	SURF \$1	Water		2.19	ng/L	3.30	mg/L	<	5.00	Hg/L	(.05	mg/L	118.67	mg/L
6589	SURF #2	Water		2.13	ng/L	3.25	mg/L	(5.00	µg/L	<	.05	ng/L	116.96	mg/L

Commentaires/ Comments: Par ICP.

Effectué par/ Nork by : R. Pelletier

CERTIFICAT D'ANALYSE / CERTIFICATE OF ANALYSIS

A/To : L.St-Arnaud

PROJET / PROJECT: V2-1 T03

Date: 4/30/93

Lab #	I.D.	Description	1	S	ie		S	in		2	'n
6563	FS-1A	Vater	<	.05	mg/L	<	.05	ng/L	<	5.00	us/L
6564	FS-18	Water		.07	mg/L	<	.05	ng/L	(5.00	ug/L
6565	FS-1C	Water	<	.05	mg/L	(. 05	Rg/L	Ì	5.00	ug/L
6566	FS-2A	Water	<	. 05	mg/L	(.05	mg/L	ć	5.00	uq/L
6567	FS-28	Water	<	.05	mg/L		.05	MS/L	ć	5.00	ug/L
6568	FS-2C	Water	<	.05	mg/L	<	.05	Rg/L	ć	5.00	ug/L
6569	FS-3A	Water	<	. 05	mg/L	<	.05	mg/L	<	5.00	ug/L
6570	FS-38	Water	<	.05	mg/L	<	.05	MA/L	<	5.00	ug/L
6571	FS-3C	Water		.05	mg/L	<	.05	IS/L	-	21.08	MS/L
6572	FS-4A	Water	<	.05	mg/L	<	.05	ag/L	<	5.00	ue/L
6573	FS-48	Water	(.05	ag/L	(.05	ss/L	(5.00	ug/L
6574	FS-4C	Water	<	.05	mg/L	<	.05	ng/L	(5.00	ug/L
6575	FS-5A	Water	(.05	ng/L	(.05	ag/L	<	5.00	µg/L
6576	FS-5B	Water	(.05	mg/L	<	.05	mg/L	<	5.00	ug/L
6577	FS-5C	Vater	<	.05	ag/L	<	.05	es/L	< l	5.00	ug/L
6578	FS-6A	Water	<	.05	ma/L	(.05	ma/L	ć	5.00	ue/L
6579	FS-68	Water	(.05	mq/L	ć	.05	ac/L	•	21.33	
6580	FS-8A	Water	<	.05	mq/L	<	.05	ng/L	(5.00	uq/L
6581	FS-88	Water	(.05	ng/L	Ś	.05	mg/L	•	21.16	MA/L
6582	FS-9A	Water	<	.05	mg/L	ć	.05	ng/L	(5.00	ue/L
6583	FS-9B	Water	, (.05	ng/L	į	.05	ng/L	i	5.00	ua/L
6584	FS-9C	Water	ì	.05		ì	.05		į	5.00	ue/i
6585	FS-10A	Water	Ì	.05	Ad/L	ì	.05	ne/L	•	7.50	ua/L
6586	FS-108	Water	, (.05	ma/1	į	.05	ma/l		.14	ma/l
6587	PREP.\$1	Water	•	.15	na/1	•	.31	ng/1		13.17	
6588	SURF \$1	Water	(.05	ma/l	(.05	ma/1		.39	
6589	SURF #2	Water	, (.05	ma/L	ì	.05	ma/L		.40	ad/L

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Commentaires/ Comments: Par ICP.

Effectué par/ Work by : R. Pelletier

	certificate	e date: feb	26, 1993		·													
	Al (mg/L)	As (mg/L)	Ca (mg/L)	Cđ (ug/L)	Co (ug/L)	Cr (ug/L)	Cu (ug/L)	Fe (ug/L)	K (mg/L)	Mg (mg/L)	Mn (ug/L)	Na (mg/L)	Ni (ug/L)	Pb (mg/L.)	S (mg/L)	Se (mg/L)	Sn (mg/L)	Zn (ug/L)
Standard sample	20.00	1.00	100.00	1.00	-	-	10.00	30.00	-	-	20.00	-	10.00	-	-	-	-	20.00
Laboratory Analysis	20.62 20.39 20.85	1.04 1.02 1.06	101.88 101.26 103.14	1.02 1.01 1.03	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	11.02 10.86 11.15	30.32 30.04 30.68	<1.00 <1.00 <1.00	<0.10 <0.10 <0.10	20.86 20.60 20.88	1.04 1.02 1.03	10.18 10.09 10.30	<0.05 <0.05 <0.05	35.74 35.26 35.79	<0.05 <0.05 <0.05	- -	21.15 21.18 21.39
X (mg/L) S.D. (mg/L) C.V. (%)	20.62 0.19 0.91	1.04 0.02 1.57	102.09 0.78 0.77	1.02 0.01 0.80	-	- -	11.01 0.12 1.08	30.35 0.26 0.86	-	- -	20. 78 0.13 0.61	1.03 0.01 0. 79	10.19 0.09 0.84	- - -	35.60 0.24 0.67	- -	-	21.24 0.11 0.50
Replicate	0.07 0.07 0.06	<0.05 <0.05 <0.05	9.62 9.64 4.76	<5.00 <5.00 <5.00	21.60 22.70 19.00	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	0.56 0.56 1.60	1.11 1.09 1.00	1.14 1.14 0.79	4.28 4.25 4.17	6.94 6.86 4.41	11.80 10.20 7.16	<0.05 <0.05 <0.05	6.80 6.76 4.05	<0.05 <0.05 <0.05	<0.05 <0.05 <0.05	<5.00 <5.00 <5.00
X (mg/L) S.D. (mg/L) C.V. (%)	0.06 0.00 7.54	<0.05 0.00 0.00	8.01 2.30 28.67	<5.00 0.00 0.00	21.10 1.55 7.35	<5.00 0.00 0.00	<5.00 0.00 0.00	0.91 0.49 54.07	1.07 0.05 4.49	1.02 0.16 16.12	4.23 0.05 1.10	6.07 1.17 19.35	9.72 1.92 19.80	<0.05 0.00 0.00	5.87 1.29 21.93	<0.05 <0.05 0.00	<0.05 <0.05 0.00	<5.00 <5.00 0.00
Lab Blank	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Laboratory Analysis	<0.05	<0.05	0.15	<5.00	<5.00	<5.00	<5.00	15	<1.00	<0.10	1.44	1.1	<5.00	<0.05	0.16	<0.05	<0.05	<5.00

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X = Arithmetic mean S.D. = Standard deviation

C.V. = Coefficient of variation

Fs2-26chem.wb1

	certificat	certificate date: April 30, 1993																
	Ai (mg/L)	As (mg/L)	Ca (mg/L)	Cd (ug/L)	Co (ug/L)	Cr (ug/L)	Cu (ug/L)	Fe (ug/L)	K (mg/L)	Mg (mg/L)	Mn (ug/L)	Na (mg/L)	Ni (ug/L)	Pb (mg/L)	S (mg/L)	Se (mg/L)	Sn (mg/L)	Zn (ug/L)
Standard sample	20.00	1.00	100.00	1.00	-	-	10.00	30.00			20.00		10.00	-	-	-	-	20.00
Laboratory Analysis	20.60 21.04 20.88	1.01 1.07 1.07	100.50 101.23 100.82	1.02 1.03 1.03	<5.00 <5.00 <5.00	<5.00 <5.00 <5.00	11.08 11.34 11.26	29.75 30.21 30.08	<1.00 <1.00 <1.00	0.11 0.13 0.10	20.22 20.50 20.47	1.21 1.44 1.29	9.90 10.03 10.00	<0.05 <0.05 <0.05	36.00 36.29 35.72	0.05 <0.05 <0.05	<0.05 <0.05 <0.05	21.08 21.30 21.16
X (mg/L) S.D. (mg/L) C.V. (%)	20.84 0.18 0.87	1.05 0.03 2.69	100.85 0.30 0.30	1.03 0.00 0.46	- - -	- -	11.23 0.11 0.97	30.01 0.19 0.65	-	-	20.40 0.13 0.62	1.31 0.10 7.26	9.98 0.06 0.56	-	36.00 0.23 0.65	- - -	-	21.18 0.09 0.43
Lab Blank	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Laboratory Analysis	<0.05	<0.05	0.11	<5.00	<5.00	<5.00	<5.00	<5.00	<1.00	<0.10	<1.00	1.15	<5.00	<0.05	0.27	<0.05	<0.05	<5.00

X = Arithmetic mean

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Fs4-30chem.wb1

S.D. = Standard deviation C.V. = Coefficient of variation

APPENDIX E.4

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Calculations of tailings hydraulic conductivity.

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Natural Soil Conductivities

	Hole	Hydraulic		1
Piezometer	Depth	Conductivity	Log K	
	(m)	(cm/s)		
FS-4A	32.34	2.50E-05	-4.602	4
FS-4B	29.39	5.10E-05	-4.292	1
FS-9B	11.20	7.60E-05	-4.119	1
FS-9A	12.30	8.90E-05	-4.051	1
FS-5B	36.20	9.30E-05	-4.032	1
FS-6A	34.17	1.20E-04	-3.921	1
FS-5A	39.28	1.90E-04	-3.721	
FS-5C	27.20	3.10E-04	-3.509	
FS-9C	4.80	8.11E-04	-3.091	
FS-1B	13.19	2.50E-03	-2.602	<median< td=""></median<>
FS-2B	11.02	2.90E-03	-2.538]
FS-2A	16.25	2.90E-03	-2.538	
FS-8	5.10	6.10E-03	-2.215	
FS-3B	51.02	8.60E-03	-2.066	
FS-3A		1.00E-02	-2.000	
FS-14A	30.35	4.00E-02	-1.398	
FS-14B	26.65	4.00E-02	-1.398	
FS-10A	FS-10A 10.82		-0.097	
FS-10B	8.26	8.00E-01	-0.097	
		Log Average:	-2.752	-
		Average:	1.77E-03	
		Median:	2.62E-03	

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Estimated Tailings Conductivities Using Modified Kozeny-Carman

Sample #	Depth	d50	k50	Log K	7
	(m)	(mm)	(cm/s)	1	1
FS-6	4.5 - 5.2	0.0082	1.13E-07	-6.949	1
FS-5	7.6 - 8.2	0.011	2.03E-07	-6.694	1
FS-6	Surface	0.017	4.84E-07	-6.315	7
FS-4	7.6 - 8.2	0.017	4.84E-07	-6.315	1
FS-4	Surface	0.0185	5.73E-07	-6.242	
FS-13	3.1 - 3.7	0.0195	6.36E-07	-6.196]
FS-11	7.6 - 8.2	0.021	7.38E-07	-6.132	1
FS-6	3.05 - 3.65	0.024	9.64E-07	-6.016	
FS-5	4.57 - 5.18	0.0245	1.00E-06	-5.998	
FS-11	1.5 - <u>2</u> .1	0.04	2.68E-06	-5.572	1
FS-4	6.10 - 6.70	0.044	3.24E-06	-5.489]
FS-8	1.5 - 2.1	0.051	4.35E-06	-5.361	<median< td=""></median<>
FS-5	Surface	0.058	5.63E-06	-5.250	1
FS-5	3.05 - 3.65	0.071	8.44E-06	-5.074]
FS-12	16.76 - 17.37	0.12	2.41E-05	-4.618	
FS-11	6.1 - 6.71	0.12	2.41E-05	-4.618	····
FS-6	6.1 - 6.71	0.16	4.28E-05	-4.368	1
FS-11	19.8 - 20.4	0.18	5.42E-05	-4.266	
FS-12	19.81 - 20.42	0.18	5.42E-05	-4.266]
FS-6	7.62 - 8.07	0.21	7.38E-05	-4.132	
FS-11	18.3 - 18.9	0.48	3.86E-04	-3.414	
FS-11	16.8 - 17.4	0.91	1.39E-03	-2.858	
FS-11	15.2 - 15.8	6.4	6.85E-02	-1.164	
FS-11	9.1 - 9.8	10	1.67E-01	-0.776	
			Log Average:	-4.920	-
			Average:	1.20E-05	
-	··· -		Median:	4.35E-06	-

Modified Kozeny-Carman

Sample #	Depth	d50	d10	d30	n	k50	k10	k30
····	(m)	<u>(mm)</u>	<u>(mm)</u>	<u>(mm)</u>		(cm/s)	(cm/s)	_(c m/s)
FS-4	Surface	0.0185	0.004	0.0125	0.45	5.73E-07	2.68E-08	2.61E-07
FS-4	6.10 - 6.70	0.044	0.013	0.022	0.45	3.24E-06	2.83E-07	8.10E-07
FS-4	7.6 - 8.2	0.017	0.0038	0.0115	0.45	4.84E-07	2.42E-08	2.21E-07
FS-5	Surface	0.058	0.005	0.027	0 45	5.63E-06	4 18F-08	1 225-06
FS-5	3.05 - 3.65	0.071	0.0082	0.029	0.45	8.44E-06	1 13E-07	1.415-06
FS-5	4.57 - 5.18	0.0245	0.0068	0.016	0.45	1.00E-06	7 74E-08	4 28E-07
FS-5	7.6 - 8.2	0.011	0.0024	0.0056	0.45	2.03E-07	9.64E-09	5.25E-07
FS-6	Surface	0.017	0.0041	0.01	0.45	4.84E-07	2.81E-08	1.67E-07
FS-6	3.05 - 3.65	0.024	0.0056	0.015	0.45	9.64E-07	5.25E-08	3.77E-07
FS-6	4.5 - 5.2	0.0082	0.0021	0.0043	0.45	1.13E-07	7.38E-09	3.09E-08
FS-6	6.1 - 6.71	0.16	0.082	0.13	0.45	4.28E-05	1.13E-05	2.83E-05
FS-6	7.62 - 8.07	0.21	0.017	0.13	0.45	7.38E-05	4.84E-07	2.83E-05
FS-8	1.5 - 2.1	0.051	0.007	0.02	0.45	4.35E-06	8.20E-08	6.69E-07
FS-11	1.5 - 2.1	0.04	0.024	0.03	0.45	2.68E-06	9.64E-07	1.51E-06
FS-11	6.1 - 6.71	0.12			0.45	2.41E-05		
FS-11	7.6 - 8.2	0.021	0.0105	0.018	0.45	7.38E-07	1.85E-07	5.42E-07
FS-11	9.1 - 9.8	10	0.3	1.6	0.45	1.67E-01	1.51E-04	4.28E-03
FS-11	15.2 - 15.8	6.4	0.6	2.65	0.45	6.85E-02	6.02E-04	1.18 E-02
FS-11	16.8 - 17.4	0.91	0.33	0.69	0.45	1.39E-03	1.82E-04	7.97E-04
FS-11	18.3 - 18.9	0.48	0.13	0.3	0.45	3.86E-04	2.83E-05	1.51E-04
FS-11	19.8 - 20.4	0.18	0.072	0.12	0.45	5.42E-05	8.68E-06	2.41E-05
FS-12	16.76 - 17.37	0.12	0.075	0.091	0.45	2.41E-05	9.41E-06	1.39E-05
FS-12	19.81 - 20.42	0.18	0.08	0.13	0.45	5.42E-05	1.07E-05	2.83E-05
EC 10	01 07	0.0405	0.0050	0.045	0.45			0 775 07
<u>rð-13</u>	3.1 - 3.7	0.0195	0.0053	0.015	0.45	0.30E-U/	4.702-08	3.//E-0/
				Average		9.92E-03	4.3/E-05	/.44≿-04

Sample #	Depth	d50	n	n	n	k1	k2	k3
	<u>(m)</u>	<u>(mm)</u>	1	2	3	(cm/s)	(cm/s)	(cm/s)
FS-4	Surface	0.0185	0.4	0.45	0.5	3.38E-07	5.73E-07	9.51E-07
FS-4	6.10 - 6.70	0.044	0.4	0.45	0.5	1.91E-06	3.24E-06	5.38E-06
FS-4	7.6 - 8.2	0.017	0.4	0.45	0.5	2.85E-07	4.84E-07	8.03E-07
FS-5	Surface	0.058	0.4	0.45	0.5	3.32E-06	5.63E-06	9.34E-06
FS-5	3.05 - 3.65	0.071	0.4	0.45	0.5	4.98E-06	8.44E-06	1.40E-05
FS-5	4.57 - 5.18	0.0245	0.4	0.45	0.5	5.93E-07	1.00E-06	1.67E-06
FS-5	7.6 - 8.2	0.011	0.4	0.45	0.5	1.20E-07	2.03E-07	3.36E-07
FS-6	Surface	0.017	0.4	0.45	0.5	2.85E-07	4.84E-07	8.03E-07
FS-6	3.05 - 3.65	0.024	0.4	0.45	0.5	5.69E-07	9.64E-07	1.60E-06
FS-6	4.5 - 5.2	0.0082	0.4	0.45	0.5	6.64E-08	1.13E-07	1.87E-07
FS-6	6.1 - 6.71	0.16	0.4	0.45	0.5	2.53E-05	4.28E-05	7.11E-05
FS-6	7.62 - 8.07	0.21	0.4	0.45	0.5	4.36E-05	7.38E-05	1.23E-04
FS-8	1.5 - 2.1	0.051	0.4	0.45	0.5	2.57 E- 06	4.35E-06	7.23E-06
FS-11	1.5 - 2.1	0.04	0.4	0.45	0.5	1.58E-06	2.68 E- 06	4.44E-06
FS-11	6.1 - 6.71	0.12	0.4	0.45	0.5	1.42E-05	2.41E-05	4.00E-05
FS-11	7.6 - 8.2	0.021	0.4	0.45	0.5	4.36E-07	7.38E-07	1.23E-06
FS-11	9.1 - 9.8	10	0.4	0.45	0.5	9.88E-02	1.67E-01	2.78E-01
FS-11	15.2 - 15.8	6.4	0.4	0.45	0.5	4.05E-02	6.85E-02	1.14E-01
FS-11	16.8 - 17.4	0.91	0.4	0.45	0.5	8.18E-04	1.39E-03	2.30E-03
FS-11	18.3 - 18.9	0.48	0.4	0.45	0.5	2.28E-04	3.86E-04	6.40E-04
FS-11	19.8 - 20.4	0.18	0.4	0.45	0.5	3.20E-05	5.42E-05	9.00E-05
FS-12	16.76 - 17.37	0.12	0.4	0.45	0.5	1.42E-05	2.41E-05	4.00E-05
FS-12	19.81 - 20.42	0.18	0.4	0.45	0.5	3.20E-05	5.42E-05	9.00E-05
FS-13	<u>3.1 - 3.7</u>	0.0195	0.4	0.45	0.5	3.76E-07	6.36E-07	1.06E-06
				Average		5.85E-03	9.92E-03	1.65E-02

Note: Authors Sullivan and Hertel, 1942, prefer an average grain diameter (Bear 1972, p. 167)